Monitoring Report for CSMRI Site First Quarter 2011

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ACRONYMS

bgs	below ground surface
CDPHE	Colorado Department of Public Health and Environment
CERCLA	Comprehensive Environmental Response, Compensation and Liability Act
CFR	Code of Federal Regulations
CSM	Colorado School of Mines
CSMRI	Colorado School of Mines Research Institute
DO	dissolved oxygen
DOC	dissolved organic carbon
EDD	electronic data deliverable
LCS	laboratory control sample
MCL	maximum contaminant level
µg/L	micrograms per liter
μ	micron
mg/L	milligrams per liter
NTU	nephlometer turbidity unit
ORP	oxidation-reduction potential
pCi/L	picoCuries per liter
QC/QA	quality control/quality assurance
TDS	total dissolved solids
USGS	U.S. Geological Survey

1. Introduction

This report presents the first quarter (January, February, March) 2011 results for groundwater and surface water monitoring conducted at the Colorado School of Mines Research Institute (CSMRI) site in Golden, Colorado. The monitoring was conducted by the S.M. Stoller Corporation (Stoller). This is the first quarter sampling event to include new and replacement monitor wells resulting from the flood plain characterization effort.

2. Sampling and Analysis

Stoller obtained quarterly samples of groundwater and surface water on March 1, 2, 3, and 4, 2011, from 14 groundwater monitor wells and three Clear Creek surface water sample locations. Water levels in all monitor wells were obtained on February 28, 2011. Groundwater quality samples were obtained on March 1(CSMRI-4, CSMRI-5, CSMRI-8B, CSMRI-9, CSMRI-10, CSMRI-12, and CSMRI-14); March 2 (CSMRI-1, CSMRI-6C, CSMRI-7C, CSMRI-11B, and CSMRI-13); and March 3 (CSMRI-1B and CSMRI-2). Monitor wells CSMRI-1B, CSMRI-2, and CSMRI-6C required purging on February 28, 2011, and sample collection on subsequent visits over the following days to obtain sufficient sample volume.

Clear Creek surface water samples were collected on March 2, 2011, from sample locations SW-1, SW-2, and SW-3. All aqueous samples were placed on ice in coolers and couriered to ALS Laboratory Group in Fort Collins, Colorado or to TestAmerica, Inc. in Arvada, Colorado for analyses.

Figure 1 presents the monitor well and Clear Creek surface water sample locations at the CSMRI site. The figure also shows the groundwater potentiometric surface elevations posted adjacent to each monitor well location. Potentiometric surface elevations are based on depth to groundwater relative to the surveyed top-of-casing and represent groundwater elevations as measured before purging. Groundwater levels are measured at each well to the nearest 1/100th of a foot (0.01) prior to purging and sample collection. The figure shows a northeasterly component of flow on the bench terrace area and then a northerly component of flow as groundwater flows over the bench terrace of surficial deposits and bedrock down the terrace slope; however, preferential pathways resulting from an uneven bedrock/alluvial interface are thought to exist.

Figure 2 presents hydrographs of groundwater potentiometric elevations for monitor wells CSMRI-1, CSMRI-1B, CSMRI-4, CSMRI-5, CSMRI-6C, CSMRI-8 (abandoned October 2010), CSMRI-9, CSMRI-10, and CSMRI-11B. Monitor wells CSMRI-7C, CSMRI-8B, CSMRI-12, CSMRI-13, and CSMRI-14, which were installed or deepened in January 2011 are also presented in Figure 2. Gaps in the graph denote the intermittent presence of groundwater in the monitor wells because occasionally groundwater is below the bottom depth of a monitor well, even though the bottom of the screened interval is within the underlying bedrock. Monitor wells included on Figure 2 are located within the CSMRI site proper and illustrate historical trends in the water table fluctuations throughout the site.

Figure 3 is a hydrograph of monitor well CSMRI-2. Monitor well CSMRI-2 is located near the southeast corner of the freshman parking lot on West Campus Drive and the former Welch Ditch.

CSMRI-2 is upgradient of the CSMRI site and historically has been used to provide background groundwater quality conditions. Early potentiometric elevation data reflect the use of and leakage from the nearby irrigation ditch. For example, this hydrograph shows a marked seasonal rise in the potentiometric surface during the summer months in 2005 and 2006. In 2007, the Welch Ditch was diverted upstream of CSMRI and piped to Washington Avenue in Golden and then to down-ditch users. From late 2006 through late 2007, the water level remained elevated. From early 2008 through early 2011, water levels reflect a more typical seasonal pattern of fluctuation.

At the request of Colorado Department of Public Health and Environment (CDPHE), Hazardous Materials and Waste Management Division, Radiation Program, two offset monitor wells were installed in the flood plain area in January 2011 to assess groundwater quality and potentiometric elevations in both the alluvial aquifer and the underlying Foxhills Sandstone Formation. Monitor well CSMRI-13 was installed to a depth of 8.25 feet below ground surface (bgs) and is screened through the saturated portion of the alluvial aquifer. Monitor well CSMRI-14 was installed in the Foxhills Sandstone Formation to a depth of 55.8 feet bgs and screened from the interval of 45.3 feet to 55.3 feet bgs. After the monitor wells were installed, the adjacent ground surface and top-of-casing elevations were surveyed to an accuracy of 0.01 feet. The two monitor wells are separated horizontally by 3.88 feet. Both monitor wells were fully developed after installation.

On February 28, 2011, the depth to water in both monitor wells was measured and recorded. The potentiometric surface of monitor well CSMRI-13 (alluvial) is measured at 5,673.06 feet and at monitor well CSMRI-14 (bedrock) at 5,673.83 feet. The 0.77 feet elevation difference between the potentiometric surfaces of the two monitor wells indicates an upwelling of the underlying Foxhills Sandstone Formation into the overlying alluvial formation.

In January 2011, existing monitor wells CSMRI-6C and CSMRI-11B were overdrilled and extended to the top of bedrock. The screened interval now incorporates the full saturated section at each well.

Replacement monitor wells CSMRI-7C and CSMRI-8B were installed at approximately the same location as their predecessors, CSMRI-7B and CSMRI-8, respectively. These two wells were abandoned in early October 2010 before source characterization field activities began. Monitor well CSMRI-7C is located within 1.5 feet of its former location, and CSMRI-8B is within 5.9 feet of its former location.

Monitor well CSMRI-12 is a new flood plain alluvial well and was installed at the request of CDPHE. The well is located approximately midway between wells CSMRI-8B to the west and CSMRI-13/CSMRI-14 to the east. This monitor well was installed to a depth of 8.7 feet and is screened through the saturated portion of the alluvial aquifer (Figure 1).

As-built construction diagrams and lithology logs of replacement monitor wells CSMRI-6C, CSMRI-7B, CSMRI-8B, and CSMRI-11B and new flood plain monitor wellsCSMRI-12, CSMRI-13, and CSMRI-14 are presented in Appendix A.

2.1 Groundwater Sampling

Water quality samples were collected following the procedure outlined in Appendix B, Groundwater Sampling Procedures.

Sample collection forms provide a record of water quality parameters as measured in the field as groundwater was purged from monitor wells. These forms also indicate the volume of water removed from each well. Sample collection forms are provided as Appendix C. After three casing volumes of groundwater were purged, water samples were filtered through a 0.45 micron (μ) filter, collected in laboratory-provided containers, and preserved in the field as appropriate for the analyte and analytical method. Monitor wells CSMRI-1B, CSMRI-2, CSMRI-6C, CSMRI-7C, CSMRI-13, and CSMRI-14 were purged dry and sampled on subsequent days. Monitor wells CSMRI-1B and CSMRI-7C required multiple visits to collect sufficient sample volume because they recharge so slowly.

2.2 Surface Water Sampling

Surface water samples from Clear Creek were collected on March 2, 2011, from three locations: one upstream of the site (SW-1), one downstream of the site (SW-2), and one between the upstream and downstream sample locations and due north of monitor well CSMRI-8B (SW-3) (see Figure 1). All surface water samples were collected following the procedure outlined in Appendix D, Surface Water Sampling Procedures. Surface water samples are filtered through a 0.45μ filter, collected in laboratory-provided containers, and preserved in the field as appropriate for the analyte and analytical method.

Discharge data of stream flow for Clear Creek, as measured by the U.S. Geological Survey (USGS), at Clear Creek gauging station #06719505 (USGS Surface Water Online Database) for the quarter from January 1, 2011 through March 31, 2011 are presented as Figure 4. Tabulated stream flow data for the time period of March 1 through 3, 2011 when the groundwater samples were collected indicate the mean stream flow measurements at the gauging station range from 41 to 44 cubic feet per second.

2.3 Analyses

All samples collected were analyzed using a Comprehensive Environmental Response, Compensation and Liability Act (CERCLA)-certified analytical laboratory. The results received from the laboratory were evaluated based on the following parameters:

- Data completeness
- Holding times and preservation
- Instrument initial calibrations
- Instrument performance checks
- Preparation blanks
- Duplicate sample results
- Laboratory control sample results
- Compound quantization and reporting limits (full validation only)

As a quality control/quality assurance (QA/QC) check, an equipment blank sample was collected in the field by pouring distilled water through a sample bailer. The equipment blank sample was submitted for the identical analytical parameters as the groundwater and surface water samples.

The laboratory control sample (LCS) and LCS duplicate in the radium-228 analyses for monitor well samples CSMRI-1, CSMRI-2, CSMRI-4, CSMRI-6C, CSMRI-7C, CSMRI-8B, and CSMRI-12 had an elevated radiometric recovery and an elevated alpha activity. This can potentially bias the results high; therefore, the analytical laboratory re-analyzed the samples that had activity above the achieved minimum detection concentration and met the control limits criteria for these samples. The laboratory did not re-prepare and reanalyze the remaining eleven samples which were non-detects and are qualified as Estimated (UJ).

Data validation results are presented in Appendix E.

2.3.1 Groundwater Quality Analyses

Summaries of groundwater results for radioisotopes, metals, and inorganic anions and cations are presented in Table 2-1, Table 2-2, and Table 2-3, respectively. Groundwater parameters are reported as picoCuries per liter (pCi/L) for radioisotopes, micrograms per liter (μ g/L) for uranium and iron, and milligrams per liter (mg/L) for all other metals and ions.

ALS Laboratory Group in Fort Collins, Colorado and TestAmerica Laboratories, Inc. in Arvada, Colorado conducted laboratory analyses of the aqueous samples. Analytical samples submitted to ALS were analyzed for radium isotopes (Ra-226 and Ra-228), uranium (U), calcium (Ca), potassium (K), magnesium (Mg), sodium (Na), chloride (Cl), sulfate (SO₄), carbonate as calcium carbonate (CO₃), bicarbonate as calcium carbonate (HCO₃), alkalinity, and dissolved organic carbon (DOC). Pursuant to an agreement with CDPHE, a broader analyte list of metals is sampled only during the second quarter (June) sampling event. Specifically, these metals include silver (Ag), arsenic (As), barium (Ba), cadmium (Cd), chromium (Cr), mercury (Hg), and vanadium (V).

Analytical samples submitted to TestAmerica were tested for the presence of nitrate (NO₃), nitrite (NO₂), ferrous (Fe²) iron, ferric (Fe³) iron, sulfide, and total dissolved solids (TDS). TestAmerica conducts the short holding time analyses because of their close proximity to the CSMRI site.

Groundwater samples were measured onsite for temperature, pH, specific conductance, oxidation-reduction potential (ORP), and turbidity as nephlometer turbidity units (NTU) during the purging and sampling process using a Horiba U-22 multi-probe. Dissolved oxygen (DO) was not measured during sampling for this quarterly event due to an inoperable probe. Monitor well parameter measurements of groundwater and purge volumes are presented on the sample collection forms in Appendix C.

Analytical data from ALS and TestAmerica were transmitted as an electronic data deliverable (EDD) and are included in Appendix F on a compact disk. Data are formatted as a series of Excel spreadsheets. Appendix G presents copies of the chain-of-custody for the CSMRI samples.

2.3.2 Surface Water Analyses

Clear Creek surface water results for radioisotopes, metals, and inorganic anions and cations are presented in Table 2-4, Table 2-5, and Table 2-6, respectively. Surface water parameters are reported as pCi/L for radioisotopes, μ g/L for uranium and iron, and mg/L for all other metals and ions. Surface water samples were measured onsite for temperature, pH, specific conductance, ORP, and NTU as the sampling was conducted. Onsite parameter measurements are presented on the sample collection forms in Appendix B.

2.4 Health and Safety Program

Stoller implements a program to protect the health and safety of field personnel during the environmental monitoring at the CSMRI site. This program has been developed in accordance with requirements of 29 Code of Federal Regulations (CFR) 1910.120.

3. Results

Groundwater analytical results for samples collected from the CSMRI site during the first quarter 2011 for radioisotopes, metals, and anions and cations are summarized on Table 2-1, Table 2-2, and Table 2-3, respectively. Surface water analytical results for samples collected from the CSMRI site during the first quarter 2011 for radioisotopes, metals, and anions and cations are summarized on Table 2-4, Table 2-5, and Table 2-6, respectively. Table 2-7 presents historical data collected by previous consultants for select contaminants of potential concern in groundwater at the site. The historical uranium data presented in Table 2-7 are presented in pCi/L as "activity," more recent (2005 through 2010) analytical data are presented in µg/L as "mass concentration." The December 7, 2000, *Federal Register* discusses the final uranium maximum contaminant level (MCL) and presents a conversion factor of a geometric average mass: activity ratio of 0.9 pCi/g for values near the National Primary Drinking Water Standards MCL, based on data from the National Inorganics and Radionuclides Survey.

Tables H-1 and H-2 in Appendix H present the quarterly historical groundwater radioisotopic and metals sample results, respectively, collected by Stoller since February 2005. Tables H-3 and H-4 in Appendix H present the quarterly historical Clear Creek surface water radioisotopic and metals sample results, respectively, collected by Stoller since February 2005.

3.1 Groundwater Conditions

Groundwater monitor wells are strategically located in areas likely to detect impacts, if any, to groundwater emanating from the site and at locations that represent background water quality. Monitor wells CSMRI-4 and CSMRI-5 are downgradient of the site in the Clear Creek flood plain. Well CSMRI-1 is located along Clear Creek upstream of the site, and well CSMRI-2 is located offsite in the southeast corner of the freshman parking lot on West Campus Drive. Both monitor wells CSMRI-1 and CSMRI-2 are upgradient of the site.

In February 2007, seven new groundwater monitor wells were installed to assess the effectiveness of the source removal excavation that was conducted in 2006. Monitor well CSMRI-8 is located along Clear Creek within the flood plain area; and monitor wells CSMRI-1B, CSMRI-6B, CSMRI-7B, CSMRI-9, CSMRI-10, and CSMRI-11 are located on the bench terrace and encircle the CSMRI site.

In July 2008, two monitor wells (CSMRI-6B and CSMRI-11) were abandoned because of construction activities at the CSMRI site. These two wells were replaced in December 2008 as CSMRI-6C and CSMRI-11B, respectively.

In January 2011, monitor wells CSMRI-6C and CSMRI-11B were overdrilled and lengthened so that the screened interval extended though the full length of the saturated section above bedrock. Monitor wells CSMRI-7C and CSMRI-8B were drilled near their former locations after source characterization field work was completed. Monitor wells CSMRI-12, CSMRI-13, and CSMRI-14 were installed in the flood plain.

3.2 Groundwater Quality

Groundwater samples were collected from 14 monitor wells and tested for the presence of metals and radioisotopes as identified in Section 2.3.1.

Uranium was detected in samples from monitor wells CSMRI-4 (68 μ g/L), CSMRI-8B (310 μ g/L), CSMRI-9 (43 μ g/L), CSMRI-12 (320 μ g/L), and CSMRI-13 (42 μ g/L), all at concentrations exceeding the State of Colorado groundwater standard of 30 μ g/L. Uranium was also detected in samples from the remaining nine groundwater monitor wells but at concentrations below the groundwater standard.

Groundwater from monitor well CSMRI-4 historically has had elevated concentrations of uranium. Values had been declining since 1991 until the last several quarterly sampling events as depicted on Figure 5. Historically, the concentration of uranium in this monitor well spiked once in 1999 and again in 2003. The spike in the uranium concentration in 2003 was attributed to precipitation effects and removal of asphalt and concrete as discussed in Section 4.2.2 of the New Horizons RI/FS (New Horizons 2004). The recent (2009) rise in the uranium concentration in this monitor well appears to be attributed to stormwater discharge from the new Colorado School of Mines (CSM) artificial turf soccer field subdrains. Precipitation collected in the subdrains, discharged near the northern edge of the bench terrace, and flowed down a riprapembedded concrete rundown onto the flood plain. During the process, the discharge water became oxygenated and was introduced into the poorly oxygenized environment of the flood plain area. Uranium present in the saturated sediments of the flood plain was mobilized by the oxygenated discharge pipe was relocated to the east. Recent analytical data suggest the soccer field discharge pipe is no longer affecting this monitor well.

Figure 6 presents the potentiometric surface elevation of groundwater in CSMRI-4 (left Y axis) and the uranium concentration (right Y axis) from 2005 through the first quarter 2011. The figure indicates the uranium concentration in groundwater was fluctuating seasonally from slightly above to slightly below the groundwater standard of $30 \mu g/L$ through seven quarterly sampling events in 2005 and 2006. An ice chest from the fourth quarter 2006 (December) sampling event was lost by the courier service resulting in a gap in the analytical data for CSMRI-4. The concentration of uranium in groundwater at this well has increased since the 2006 surface soil remediation activities and also spiked during the third quarter 2009 sampling event. The uranium concentration in this monitor well has decreased significantly since the third quarter 2009 sampling event and reflects the relocation of the soccer field discharge pipe.

The uranium concentration in groundwater at monitor well CSMRI-9 (43 μ g/L) increased slightly from the previous quarterly sample concentration (37 ug/L). This monitor well is located at the top of the bench terrace that rises above the flood plain and is downgradient of the CSMRI site. Figure 7 presents the historical water table elevations and uranium concentrations since January 2007.

The initial groundwater sample from flood plain alluvial monitor well CSMRI-8B indicates uranium was detected at a concentration of 310 μ g/L. This monitor well is a replacement well for its predecessor, CSMRI-8, and is located within the area where source material was characterized in October and November 2011. The current groundwater concentration for uranium at this time is significantly lower than the predecessor monitor well previous concentration of 520 μ g/L.

Monitor wells CSMRI-12 and CSMRI-13 are new alluvial flood plain monitor wells, and the initial groundwater sample results for groundwater from these wells indicate the presence of uranium at concentrations of 320 μ g/L and 42 μ g/L, respectively.

Monitor well CSMRI-14 was installed in the flood plain area and is screened in the underlying Foxhills Sandstone Formation. The initial groundwater sample from this well indicates the presence of uranium at a concentration of 1.6 μ g/L. This value is below the groundwater quality standard for uranium and is lower than the detected concentration of uranium in Clear Creek surface water samples SW-1, SW-2, and SW-3 at concentrations of 2 μ g/L, 2.1 μ g/L, and 2 μ g/L, respectively.

Quarterly sampling and analytical testing of water quality parameters will continue and trends in the concentration of uranium will continue to be monitored. Future analytical data from the recently retro-fitted wells and newly installed monitor wells in the flood plain area will provide a better overall picture of groundwater conditions across the CSMRI site.

3.2.1 Ionic Balance Evaluation

The ionic testing and balancing is conducted to determine the different groundwater hydrochemical facies within the CSMRI site and to assess the analytical quality control procedures since the sum of the major anions should equal the sum of the major cations when the ionic concentrations are converted to millequivalents per liter.

Groundwater and surface water samples were collected and tested for major anions and cations, DOC, and from the five flood plain monitor wells (CSMRI-4, CSMRI-5, CSMRI-8B, CSMRI-12, and CSMRI-13) ferric/ferrous iron and sulfide. Analytical results for these parameters are presented in Table 2-3 for groundwater and Table 2-6 for surface water.

AqQA[®] geochemical software is used to calculate ionic balances of water samples and to present the graphical representation of anions and cations. Ionic balance calculations for the anions and cations for the water samples generally range from 0.12 to 8.3 percent (monitor well CSMRI-2). Significant inequalities of the ionic balance between the anions and cations suggest internal analytical laboratory quality issues; it could also indicate that an ion is present in the water sample and is not being analyzed. Overall there is good agreement between the anion and cation data sets; indicating that the laboratory procedures are generally performed properly.

Summary sheets from the AqQA® geochemical software for each of the water samples are presented in Appendix I. Dominant water types identified at the CSMRI site include Ca-Cl (CSMRI-1, CSMRI-1B, CSMRI-4, CSMRI-5, CSMRI-6C, CSMRI-8B, CSMRI-9, CSMRI-10, CSMRI-11B, CSMRI-12, and CSMRI-13); Na-HCO₃ (CSMRI-14); Ca-HCO₃ (CSMRI-2 and CSMRI-7C); and Ca-SO₄ (SW-1, SW-2, and SW-3 [all Clear Creek surface water samples]).

A Piper quadrilateral diagram is included in Appendix I and illustrates the overall ionic properties for each water sample. The cation triangle in the lower left of the figure indicates the cation composition of the water samples is generally similar as exhibited by the tight grouping of the plotted results. The anion triangle in the lower right of the figure indicates the anion composition of the water samples is more diverse as exhibited by the dispersed nature of the plotted results.

3.2.2 Comparison of Upgradient and Downgradient Groundwater Quality

Monitor wells CSMRI-4 and CSMRI-5 are downgradient from the upper terrace portion of the site and are located on the Clear Creek flood plain. Monitor well CSMRI-9 is located downgradient of the CSMRI site at the top of the bench terrace above the flood plain. Monitor wells CSMRI-10 and CSMRI-11B are located at the eastern edge of the site, and monitor wells CSMRI-1B and CSMRI-6C are located upgradient of the site.

Uranium was detected in groundwater from downgradient monitor wells CSMRI-4 (68 μ g /L) and CSMRI-9 (43 μ g/L), exceeding the groundwater standard of 30 μ g/L. Uranium was detected in groundwater from monitor well CSMRI-5 (13 μ g/L).

Uranium was detected in upgradient monitor wells CSMRI-6C ($5.5\mu g/L$) and CSMRI-11B ($14\mu g/L$). Both of these wells were overdrilled and lengthened so that the well screen now spans the saturated section of the groundwater column. The current concentration of uranium at CSMRI-6C is lower than previous samples from this well.

3.2.3 Comparison with Previous Groundwater Quality Analyses

Table 2-7 presents historical groundwater analytical results from past sampling events by other consultants dating back to 1991 for radioisotopes of concern. The data indicate fluctuating activities of tested analytes, especially uranium, for monitor well CSMRI-4. The table also presents historical analytical activity results for existing monitor wells CSMRI-1, CSMRI-2, and CSMRI-5 and monitor well CSMRI-3, which has since been closed by others.

As additional data are collected for each sampling quarter and as trends become more defined, graphs of concentration versus time are produced and presented. These analytical data are incorporated to show long-term trends and correlation between the detected concentration of uranium in groundwater, the fluctuating water table, and seasonal variability if present.

3.2.4 Comparison with Colorado Groundwater Standards

As discussed previously, the groundwater standard of 30 µg/L for uranium in groundwater was exceeded in monitor wells CSMRI-4 (68µg/L), CSMRI-8B (310 µg/L), CSMRI-9 (43 µg/L),

CSMRI-12 (320 μ g/L), and CSMRI-13 (42 μ g/L). In January 2008, the CDPHE Water Quality Control Commission adopted the surface water quality standard of 30 μ g/L as the groundwater quality standard in an effort to keep both uranium standards consistent.

The historic spikes in the concentration of uranium in CSMRI-4, as shown on Figure 5, have been attributed to increased precipitation from December 2006 through February 2007 and surface soil remediation activities in the flood plain. Recent analytical results indicate a decreasing concentration trend back to 2007 and 2008 levels prior to impacts from the soccer field storm drain.

Monitor wells CSMRI-8B, CSMRI-12, and CSMRI-13 are new monitor wells and the first quarter 2011 sampling event is the initial sample from each well. Sampling will continue at these wells to assess changes in water quality. Groundwater from monitor well CSMRI-9, as shown on Figure 7, has exceeded the groundwater quality standard for uranium since a large spike was detected in the June 2009 sampling event. The concentration of uranium has since gone down significantly but it still remains above the groundwater standard.

No exceedances of the MCL for Ra-226 and Ra-228 were detected in any of the groundwater samples for this sampling quarter.

3.3 Surface Water Quality

Surface water samples are collected from three locations at the site. Location SW-1 is located over 400 feet upstream from the CSMRI site, SW-2 is downstream from the site, and SW-3 is located adjacent to the Clear Creek bank in the vicinity of monitor well CSMRI-8B.

All surface-water concentrations of tested parameters detected at the CSMRI site from stations SW-1, SW-2, and SW-3 are similar. Water quality results for these locations are presented in Table 2-4 (radioisotopes), Table 2-5 (metals), and Table 2-6 (anions and cations). Established surface water quality standards were not exceeded for any tested analyte at any surface water sample location.

4. Future Activities

Source material characterization and removal activities were conducted in October and November 2010. The stockpiled soil has been characterized and disposal options for the soil are anticipated to be finalized by mid-2011.

Installation of replacement and new groundwater monitor wells was conducted in early January 2011. The lengthened and new monitor wells have been integrated into the quarterly sampling schedule. Sampling of these wells and of the existing wells will continue, and the analytical results will be assessed to determine trends in analyte concentrations in groundwater.

5. References

Colorado Department of Public Health and Environment, Water Quality Control Commission, Regulation No. 41, *The Basic Standards for Ground Water*. Amended: January 14, 2008, Effective: May 31, 2008.

New Horizons Environmental Consultants, Inc. Remedial Investigation/Feasibility Study and Proposed Plan, Colorado School of Mines Research Institute Site, Golden, CO, January 21, 2004.

Stoller 2010. Final Work Plan, Environmental Assessment and Characterization, Colorado School of Mines Research Institute Site, Flood Plain Area, Golden, Colorado, Prepared by The S.M. Stoller Corporation For Colorado School of Mines, August 2010.

USGS Surface Water website: http://nwis.waterdata.usgs.gov

	Summary of Radioisotopes in Groundwater													
			-226 Ci/L)		-228 Ci/L)									
Sample Station	Sample Date	Result	Uncertainty	Result	Uncertainty									
CSMRI-1	3/2/11	0.41	± 0.29	0.91 UJ	± 0.47									
CSMRI-1B	3/3/11	0.33	± 0.29	0.73	± 0.38									
CSMRI-2	3/3/11	1.59	± 0.59	1.45	± 0.55									
CSMRI-4	3/1/11	2.16	± 0.7	1.23	± 0.53									
CSMRI-5	3/1/11	0.65	± 0.4	0.16 UJ	± 0.3									
CSMRI-6C	3/2/11	0.11	± 0.22	1.22	± 0.47									
CSMRI-7C	3/2/11	0.31	± 0.41	0.76	± 0.55									
CSMRI-8B	3/1/11	0.31	± 0.29	0.88	± 0.41									
CSMRI-9	3/1/11	0.28	± 0.28	0.51 UJ	± 0.34									
CSMRI-10	3/1/11	0.22	± 0.32	0.73 UJ	±0.39									
CSMRI-11B	3/2/11	0.14	± 0.14	0.91 UJ	± 0.43									
CSMRI-12	3/1/11	0.69	± 0.41	1.24	± 0.5									
CSNRI-13	3/2/11	0.69	± 0.37	0.97 UJ	± 0.43									
CSMRI-14	3/1/11	0.3	± 0.3	0.67 UJ	± 0.39									
M	CL*		Total R	a = 5										

Table 2-1

*Maximum Contaminant Level – National Primary Drinking Water Regulations pCi/L = picoCuries per Liter UJ = Estimated

Table 2-2 Summary of Metals in Groundwater (Results in mg/L, Uranium results in µg/L)

						(Itesuits III	ing/L, Urai	itum results	· m μ ₅ / L)					
Sample Station	Sample Date	Ag	As	Ва	Са	Cd	Cr	Hg	к	Mg	Na	Pb	U	v
CSMRI-1	3/2/11	NT	NT	NT	53	NT	NT	NT	2.9	16	36	NT	2.6	NT
CSMRI-1B	3/3/11	NT	NT	NT	130	NT	NT	NT	6.2	53	50	NT	9.4	NT
CSMRI-2	3/3/11	NT	NT	NT	78	NT	NT	NT	5.7	35	20	NT	0.83	NT
CSMRI-4	3/1/11	NT	NT	NT	140	NT	NT	NT	12	58	54	NT	68	NT
CSMRI-5	3/1/11	NT	NT	NT	130	NT	NT	NT	4.5	47	55	NT	13	NT
CSMRI-6C	3/2/11	NT	NT	NT	100	NT	NT	NT	7.8	39	66	NT	5.5	NT
CSMRI-7C	3/2/11	NT	NT	NT	100	NT	NT	NT	9.7	46	60	NT	8.3	NT
CSMRI-8B	3/1/11	NT	NT	NT	160	NT	NT	NT	17	58	60	NT	310	NT
CSMRI-9	3/1/11	NT	NT	NT	130	NT	NT	NT	5.7	62	43	NT	43	NT
CSMRI-10	3/1/11	NT	NT	NT	120	NT	NT	NT	6.2	49	39	NT	14	NT
CSMRI-11B	3/2/11	NT	NT	NT	120	NT	NT	NT	5.4	45	57	NT	14	NT
CSMRI-12	3/1/11	NT	NT	NT	140	NT	NT	NT	12	50	52	NT	320	NT
CSMRI-13	3/2/11	NT	NT	NT	120	NT	NT	NT	9	52	56	NT	42	NT
CSMRI-14	3/1/11	NT	NT	NT	26	NT	NT	NT	4.3	13	39	NT	1.6	NT
Detection	i Limits	0.01	0.01	0.1	1	0.005	0.01	0.0002	1	1	1	0.003	0.1	8.1
MCI	*	NE	0.010	2	NE	0.005	0.1	0.002	NE	NE	NE	0.015	30	NE

*Maximum Contaminant Level – National Primary Drinking Water Regulations

NE – Not Established

ND - Tested but not detected above the Detection Limits

NT - Not Tested

J - Estimated value

Sample Station	Sample Date	Bicarbonate as CaCO₃ (mg/L)	Carbonate as CaCO₃ (mg/L)	Total Alkalinity as CaCO₃ (mg/L)	Chloride (mg/L)	Total Dissolved Solids (mg/L)	Dissolved Organic Carbon (mg/L)	Ferric Iron (mg/L)	Ferrous Iron (mg/L)	Total Iron (ug/L)	Nitrate (mg/L)	Nitrite (mg/L)	Sulfate (mg/L)	Sulfide (mg/L)
CSMRI-1	3/2/11	79	ND	79	100	340	ND	NT	NT	NT	1.1	ND	76	NT
CSMRI-1B	3/3/11	230	ND	230	280	710	1.7	NT	NT	NT	2.3	ND	97	NT
CSMRI-2	3/3/11	290	ND	290	23	400	ND	NT	NT	NT	ND	ND	75	NT
CSMRI-4	3/1/11	290	ND	290	220	790	2.1	ND	ND	ND	0.64	ND	150	ND
CSMRI-5	3/1/11	220	ND	220	250	700	1.4	ND	ND	ND	4.0	ND	93	ND
CSMRI-6C	3/2/11	180	ND	180	250	630	1.6	NT	NT	NT	2.5	ND	71	NT
CSMRI-7C	3/2/11	380	ND	380	83	640	1.5	NT	NT	NT	ND	ND	110	NT
CSMRI-8B	3/1/11	340	ND	340	230	880	2.8	ND	ND	ND	1.3	ND	160	ND
CSMRI-9	3/1/11	290	ND	290	240	740	1.6	NT	NT	NT	4.6	ND	96	NT
CSMRI-10	3/1/11	230	ND	230	260	690	1.4	NT	NT	NT	4.1	ND	75	NT
CSMRI-11B	3/2/11	230	ND	230	260	680	1.6	NT	NT	NT	3.2	ND	71	NT
CSMRI-12	3/1/11	320	ND	320	180	770	2.3	0.20	0.83	ND	ND	ND	170	0.13
CSMRI-13	3/2/11	280	ND	280	210	720	2.3	ND	ND	ND	2.2	ND	110	ND
CSMRI-14	3/1/11	180	ND	180	3.6	250	1.8	NT	NT	NT	ND	ND	50	NT
Reporti	ing Limits	5 or 20	5 or 20	5, 10 or 20	1, 2 or 4	10	1	0.20	0.20	100	0.50	0.50	5, 10 or 20	0.050

Table 2-3 Summary of Anions and Cations in Groundwater

mg/L - milligrams per liter

 μ g/L- micrograms per liter ND – Non Detect

NT - Not Tested

Sample Station	Sample Date		226 Si/L)		-228 Si/L)	Total U (μg/L)
Station	Dale	Result	Uncertainty	Result	Uncertainty	Result
SW-1	3/2/11	0.07	± 0.13	0.31 UJ	± 0.3	2
SW-2	3/2/11	0.18	± 0.18	0.38 UJ	± 0.33	2.1
SW-3	3/2/11	0.87	± 0.39	0.56 UJ	± 0.38	2
M	CL*		Total I	Ra = 5		30

Table 2-4 Summary of Radioisotopes in Surface Water

*Maximum Contaminant Level - National Primary Drinking Water Regulations

pCi/L = picoCuries per Liter μ g/L = micrograms per Liter UJ = Estimated value

Table 2-5 Summary of Metals in Surface Water

Sample Station	Sample Date	Ag (mg/L)	As (mg/L)	Ba (mg/L)	Ca (mg/L)	Cd (mg/L)	Cr (mg/L)	Hg (mg/L)	K (mg/L)	Mg (mg/L)	Na (mg/L)	Pb (mg/L)	U (µg/L)	V (mg/L)
SW-1	3/2/11	NT	NT	NT	38	NT	NT	NT	2.7	8.9	17	NT	2	NT
SW-2	3/2/11	NT	NT	NT	40	NT	NT	NT	2.7	9.3	17	NT	2.1	NT
SW-3	3/2/11	NT	NT	NT	40	NT	NT	NT	2.7	9.2	17	NT	2	NT
	on Limits	0.01	0.01	0.1	1	0.005	0.01	0.0002	1	1	1	0.003	0.1	0.01
	CLs*	0.01	0.010	2	NE	0.005	0.1	0.002	NE	NE	NE	0.015	30	NE

*Maximum Contaminant Level - National Primary Drinking Water Regulations

mg/L – milligrams per liter

μg/L- micrograms per liter NE – Not Established

ND = Not Detected at or above the detection limit

B = Reported value was less than the contract lab required detection limit but equal to or greater than the lab MDL.

Table 2-6 Summary of Metals in Surface Water

Sample Station	Sample Date	Bicarbonate as CaCO ₃ (mg/L)	Carbonate as CaCO₃ (mg/L)	Total Alkalinity as CaCO₃ (mg/L)	Chloride (mg/L)	Total Dissolved Solids (mg/L)	Dissolved Organic Carbon (mg/L)	Ferric Iron (mg/L)	Ferrous Iron (mg/L)	Total Iron (µg/L)	Nitrate (mg/L)	Nitrite (mg/L)	Sulfate (mg/L)
SW-1	3/2/11	50	ND	50	31	240	ND	NT	NT	NT	ND	ND	93
SW-2	3/2/11	49	ND	49	32	240	ND	NT	NT	NT	ND	ND	100
SW-3	3/2/11	48	ND	48	31	230	1.1	NT	NT	NT	ND	ND	99
Reporting Lim		5	5	5	0.4	10	1	NT	NT	NT	0.50	0.50	1

mg/L – milligrams per liter

μg/L- micrograms per liter

ND = Not Detected at or above the Reporting Limits

NT = Not Tested

Monitoring Report for CSMRI Site First Quarter 2011

Well ID (d)	Analyte	1/1991 (a)	6/1991 (a)	3/1999 (b)	6/1999 (b)	10/1999 (b)	2/2003 (c)	4/2003 (c)	7/2003 (c)	10/2003 (c)
	Ra-226			0.1	0.3	0.2	<0.55	<0.45	ND (<0.38)	ND (<0.31)
CSMRI-1	U Total			2.09	2.59	1.44	2.4	2.9	0.87	1.4
	Th-230			0.4	0.2	0.2	<0.19	0.21	ND (<0.13)	<0.15
	Ra-226		1.9	1.9	1.4	1.4	1.4	2.8	2.1	1.7
CSMRI-2	U Total	11	5.7	0.55	1.46	0.71	1.5	1.3	1.9	1.3
	Th-230		0	0.1	0.1	0.9	<0.17	0.43	0.20	0.31
	Ra-226		0.6	1.5	1.2	1.6	<0.75	<0.81	ND (<0.49)	<0.98
CSMRI-3	U Total	17	10.4	8.41	12.4	10	12	12	9	10
	Th-230		0	0.3	0.3	1.1	<0.12	ND (<0.15)	ND (<0.17)	ND (<0.14)
	Ra-226		1	<0.4	0.6	0.4	<0.85	<0.42	<0.32	ND (<0.64)
CSMRI-4	U Total	86	57.3	23.4	58.6	33.7	16	34.2	53	19
	Th-230		0	0.7	0.3	0.4	< 0.099	ND (<0.15)	ND (<0.17)	ND (<0.12)
	Ra-226		0.6	2.4	3.3	2.7	ND (<0.49)	1.1	2.6	1.59
CSMRI-5	U Total	14	16.8	3.6	3.6	4	2.8	2.3	2.7	3.3
	Th-230		0	0.2	0.2	1.4	0.062	ND (<0.14)	ND (<0.19)	ND (<0.13)

Table 2-7 **CSMRI** Historical Groundwater Data (Previous Consultants) (All results in picoCuries per liter)

Notes: ND = Not Detected

a - Samples collected by Grant and Associates and analyzed by Barringer Labs

b - Samples collected by URS Greiner Woodward Clyde and analyzed by CORE Labs

c - Samples collected by New Horizons Environmental Consultants and analyzed by Paragon Analytics; Total U activity (pCi/L) calculated from concentration (µg/L) reported by Paragon. d - Well Identification numbers changed from the 1991 data to the 1999 data. Data presented account for this change

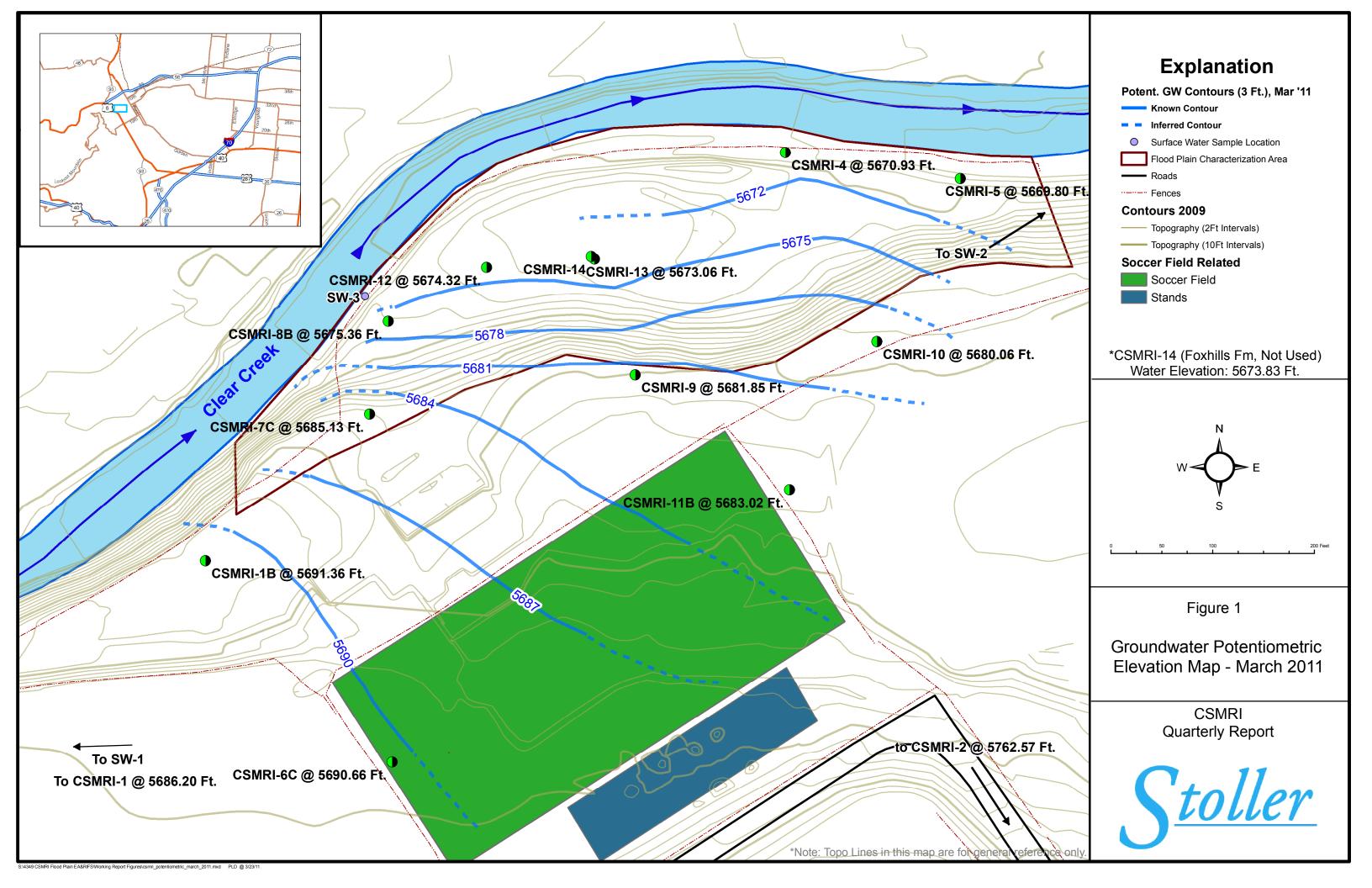
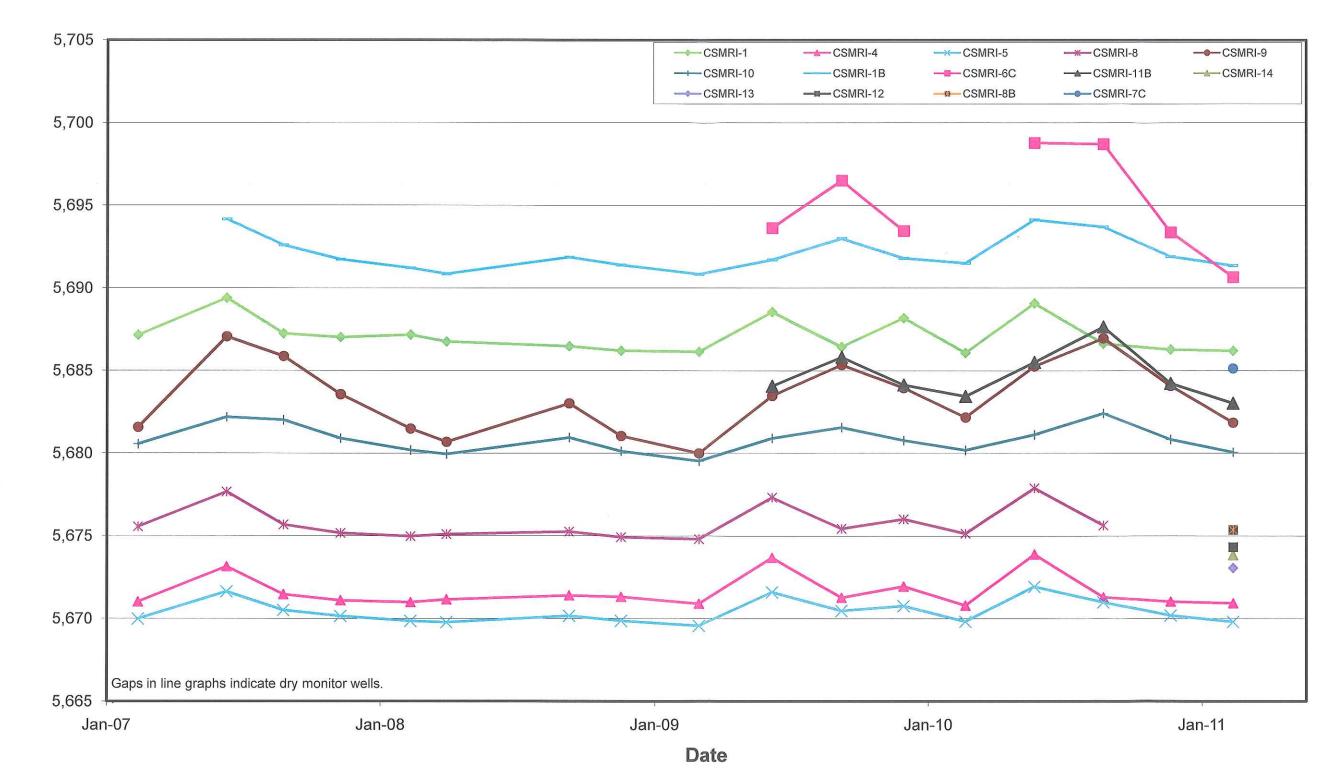


Figure 2 CSMRI All Monitor Wells Hydrograph



Elevation (ft)

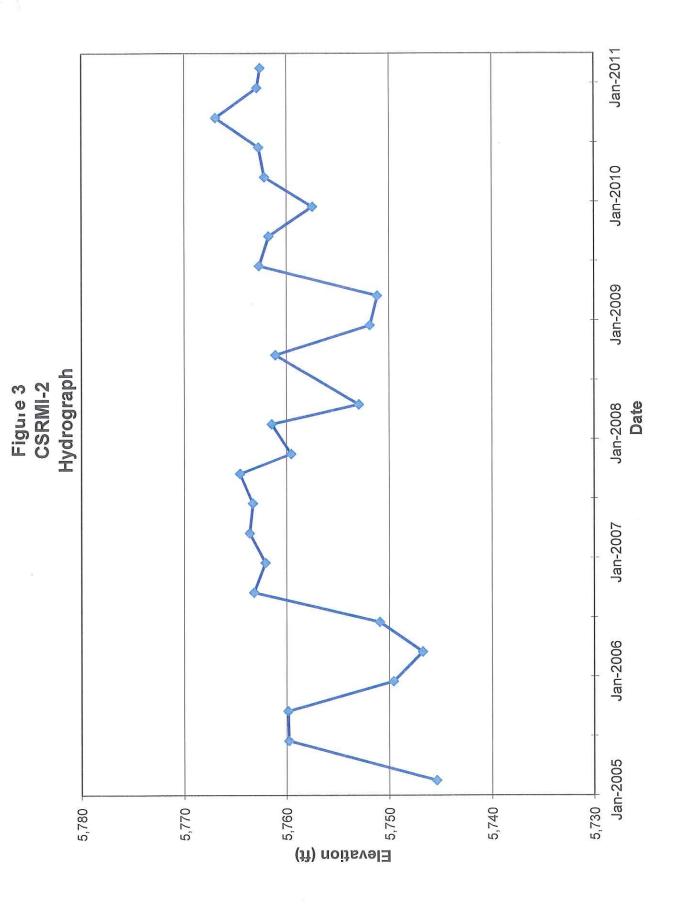
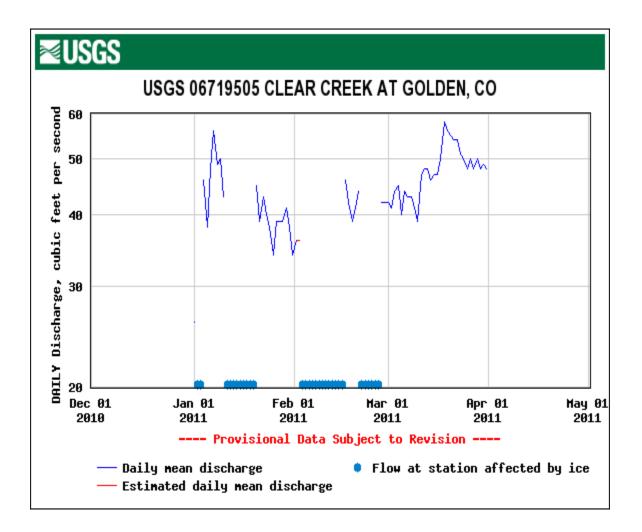
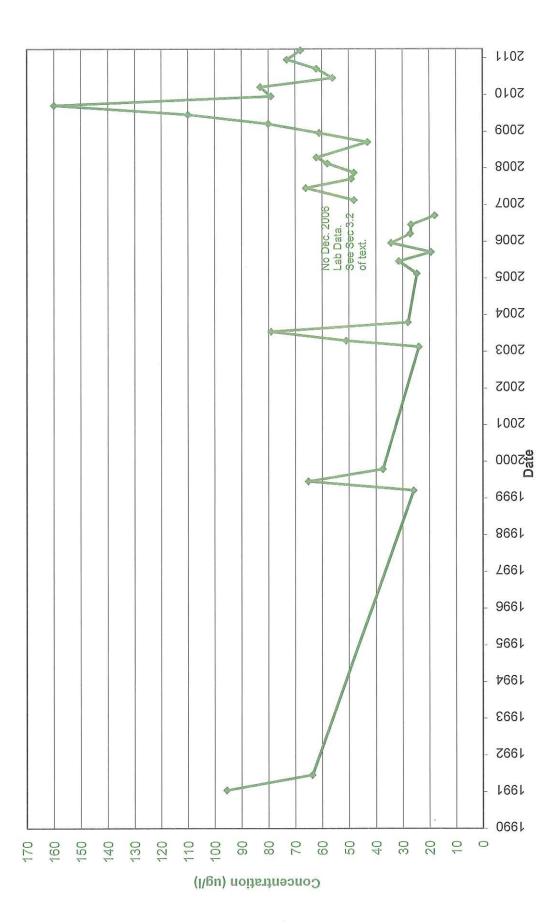
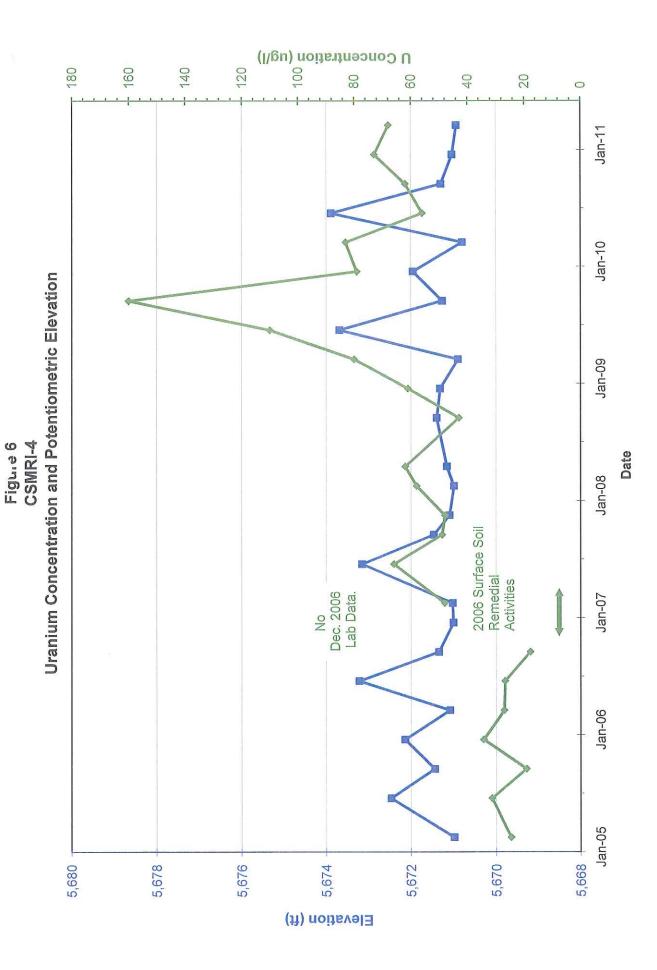


Figure 4 Clear Creek Gauging Graph January - March 2011



Figu، ف 5 CSMRI-4 Historical Total Uranium Concentration (1991 - 2011)





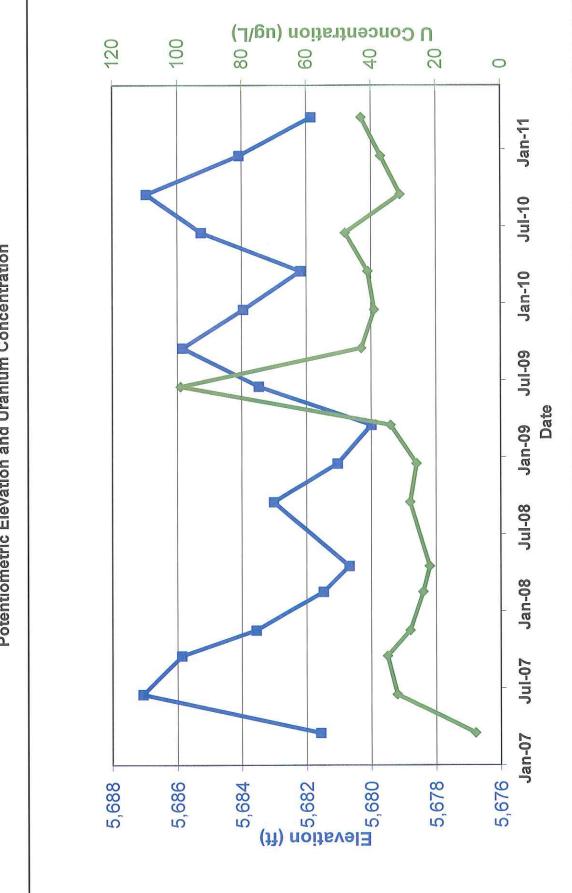
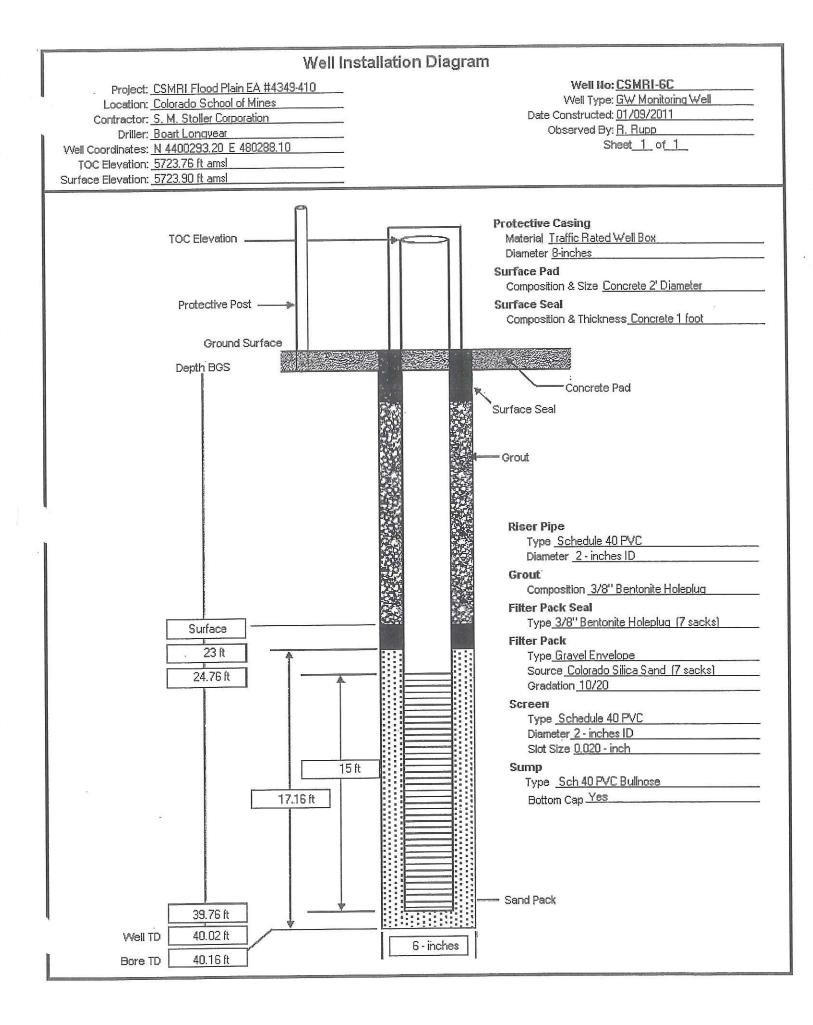


Figure 7 CSMRI-9 Potentiometric Elevation and Uranium Concentration

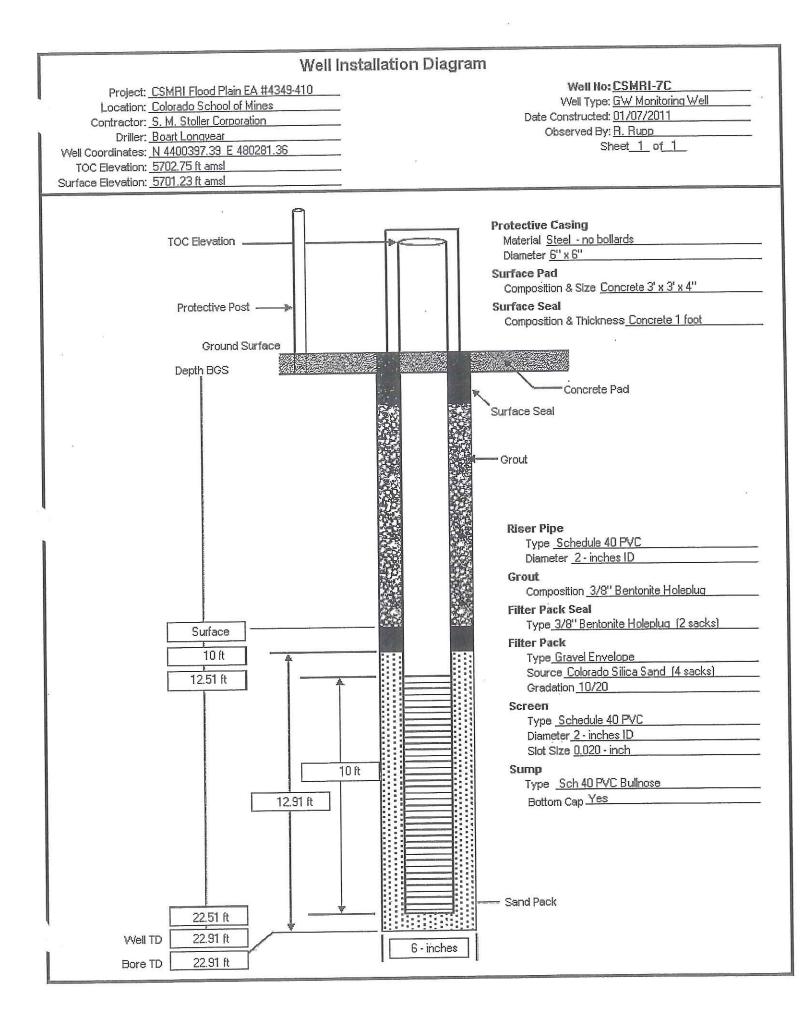
Appendix A Monitor Well Construction As-Builts

and Lithology

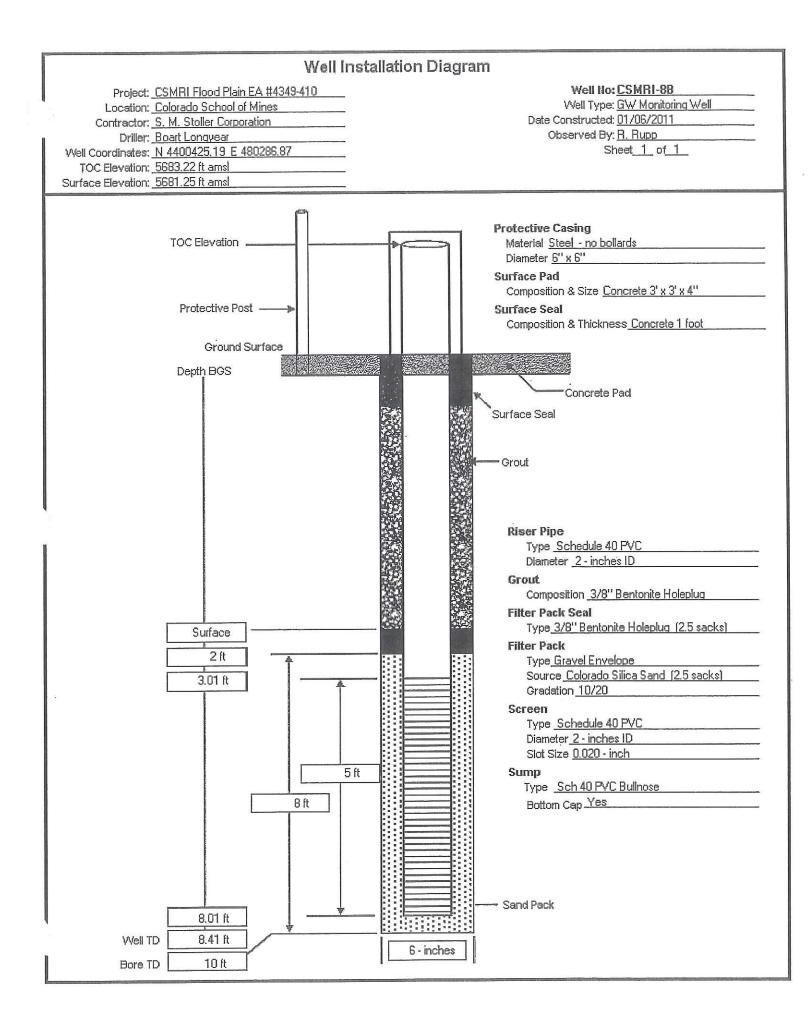
5	Top of Casing Elevation: 5723.76 ft amsl Description ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying	0 GAMMA Screen (uR/hr) Background 14 uR/hr
Drilling Contractor: Boart Longyear Dates Drilled: 01/08/2011 - 01/08/2011 Borehole Type: 6-inches Resonate Son Ground Elevation: 5723.90 ft amsl Completion (1) (1) (1) (1) (1) (1) (1) (1)	Total Depth of Borehole: 40.16 ft bgs Depth to Water: 31.98 ft btoc ic Well Type: 2-inch Sch 40 PVC Top of Casing Elevation: 5723.76 ft amsl Description 9 ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying oist plit-spoon sample blow count: 3,4,10 Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist split-spoon sample blow count: 4,7,11,15 ' Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist ' split-spoon sample blow count: 11,4,3 Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4), 10	GAMMA Screen (uR/hr) Background
Dates Drilled: 01/08/2011 - 01/08/2011 Borehole Type: 6-inches Resonate Son Ground Elevation: 5723.90 ft amsl Completion $\begin{array}{c c c c c c c c c c c c c c c c c c c $	Depth to Water: 31.98 ft btoc ic Well Type: 2-inch Sch 40 PVC Top of Casing Elevation: 5723.76 ft amsl Description Poscription ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying oist plit-spoon sample blow count: 3,4,10 Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist split-spoon sample blow count: 4,7,11,15 ' Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist ' split-spoon sample blow count: 11,4,3 Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4), Image:	Screen (uR/hr) Background
Borehole Type: 6-inches Resonate Son Ground Elevation: 5723.90 ft amsl Completion	ic Well Type: 2-inch Sch 40 PVC Top of Casing Elevation: 5723.76 ft amsl Description ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying oist plit-spoon sample blow count: 3,4,10 Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist split-spoon sample blow count: 4,7,11,15 ' Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist ' split-spoon sample blow count: 11,4,3 Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4),	Screen (uR/hr) Background
Ground Elevation: 5723.90 ft amsl Completion (1) (3) Update (1) (1) Update (1)	Top of Casing Elevation: 5723.76 ft amsl Description Description ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying oist plit-spoon sample blow count: 3,4,10 Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist split-spoon sample blow count: 4,7,11,15 Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist split-spoon sample blow count: 11,4,3 Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4), Elevation:	Screen (uR/hr) Background
Completion $\begin{array}{c c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{$	Description Description ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying oist plit-spoon sample blow count: 3,4,10 Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist split-spoon sample blow count: 4,7,11,15 ' Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist ' split-spoon sample blow count: 11,4,3 Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4),	Screen (uR/hr) Background
5 0 -5' S 5 5 10' R 10 10 10 10 10 10 15 5 15 - 20' poorly 0 -5' S 15 - 20' poorly 15 -5' S 15 - 20' poorly 10 -5' S 19 - 20 20 -5' S 5c (2.5YR 20 -5' S 5c (2.5YR 25 -5' S coarse 25 -5' S coarse 30 -5' S Sw (29 - 30) Refusa 30 -5' S -5' S 35 -6' S -7' dark re and fin 37 - 40 Bed 37 - 40	a a ilty Clay with trace fine sand, dark yellowish brown 4/2), top 1 foot is imported organic topsoil underlaying oist a/2), top 1 foot is imported organic topsoil underlaying oist a plit-spoon sample blow count: 3,4,10 a Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist a split-spoon sample blow count: 4,7,11,15 a ' Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist a ' split-spoon sample blow count: 11,4,3 a Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4), a	Screen (uR/hr) Background
5 CL (10YR sod, mu 4-5' s) 5 5 - 10' poorly 9 - 10' CL 10 - 15 10 10 - 15 medium 15 5 - 20' poorly 20 5 - 20' poorly 15 - 20' sc 15 - 20' 20 5 - 20' 20 - 25 30 5 - 20' sc 21 20 - 25 contain 22 5 - 34 coarse 25 5 - 34 coarse 30 5 - 30' sw 29 - 30 8w 29 - 30 Refusa 35 - 0' 35 6c 37 - 40 8w 37 - 40 37 - 40 woitt	 4/2), top 1 foot is imported organic topsoil underlaying oist pplit-spoon sample blow count: 3,4,10 Silty, Sandy, Clay, dark yellowish brown (10YR 4/2), sorted fine to medium sand, moist split-spoon sample blow count: 4,7,11,15 Sandy Clay, dark reddish brown (2.5YR 3/3), sorted n and fine sand, moist split-spoon sample blow count: 11,4,3 Clayey Sand, 60% sand, 40% clay, dusky red (10R 3/4), 	Background 14 uR/hr
Total I Boreho	 /6) ' split-spoon sample blow count; 2,2,4,4 ' Clayey Sand, 70% sand, 30% clay, light reddish brown (a/4), poorly sorted very coarse to fine sand ' split-spoon sample blow count: 3,7,10,10; sample (ad 20% - 25% clay) ' Sand with gravel, light gray (5YR 7/1), poorly sorted to fine sand with gravel, gravel size ranges from 0.5" to or change at 30' to light brown (7.5YR 6/3) ' split spoon sample blow count: 34,50,7 - Auger I at 30.5' bgs ' Clayey, Sandy, Gravel, 50/50 gravel and sand alluvium, ddish brown (5YR 3/2), poorly sorted coarse, medium, e sand, moist to saturated, some slough .16' Pierre Shale, dark gray (5YR 4/1), weathered and o 37.5'; dry, competent shale from 37.5' to 40' bgs Depth of Borehole 40.16 feet bgs Depth of Well 40.02 feet bgs 	14 uR/hr 14 uR/hr



		CSMRI-7C Colorado School of Mines	Go	lden, Colorado	
CSMRI Flood Plain EA	0		Client: Colorado School of Mines		
Project Number: 4349-41 Geologist: P. Dalen	0				
Drilling Contractor: Boar	t Longvear	Total Depth of Borehole: 22.91 ft bgs			
Dates Drilled: 01/07/201					
Borehole Type: 6-inches		ate Sonic Well Type: 2-inch Sch 40 PVC			
×1.	1.23 ft amsl	Top of Casing Elevation: 5702.75 ft am	sl		
Completion (Ft)	Lithology USCS	Description	Sample	GAMMA Screen (uR/hr)	
	CL	0 - 0.5' Sandy Clay Loam Topsoil, dark reddish brown (5YR 3/4), fine to coarse grain sub-angular sand, some pebbles, poorly sorted, moist		Background 19 uR/hr	
		0.5 - 14' Clayey Sand and Gravel Alluvium, reddish brown (5YR 5/3) to gray (5YR 6/1) layered coloration; fine to coarse grain sub-angular to sub-rounded, poorly sorted sand with abundant pebbles and cobble size material and some boulders. Finer material located mainly in gray colored intervals with 5 - 40% coarse fragments; moist to damp with depth		19 uR/hr	
	GC	,*		19 uR/hr	
				19 uR/hr	
- 15 -	GC	14 - 18' Clayey Sandy Gravelly Alluvium similar to above, with boulder size material causing difficult drilling; bedded light gray (5YR 7/1) and dark reddish brown (5YR 2.5/2); very fine grain sand with silt and clay, gravel and cobbles; iron stain	2	19 uR/hr 19 uR/hr	
	Bed	streaks 18 - 22.91' Pierre Shale, light gray (7.5YR N6), weathered, saturated shale from 18 - 19'; dry, competent shale 19 - 22' bgs	_	19 uR/hr	
	Rx			19 uR/hr	
		Total Depth of Borehole 22.91 feet bgs			
		Total Depth of Well 22.91 feet bgs 2" Sch 40 PVC Blank Casing +1.52' ags to 12.51' bgs 2" Sch 40 0.020" Slot Screen 12.51' to 22.51' bgs 2" Sch 40 PVC Bullnose 22.51' to 22.91' bgs 10/20 Colorado Silica Sand 10' to 22.91' bgs Hydrated 3/8" Bentonite Holeplug surface to 10' bgs Surface Seal poured with 3'x 3'x 4" Concrete Pad 6"x 6" square Steel Protective Casing with Locking Cover			
S.M. STOLLE		0 x 0 square sider risective casing with booking cover		Page 1	

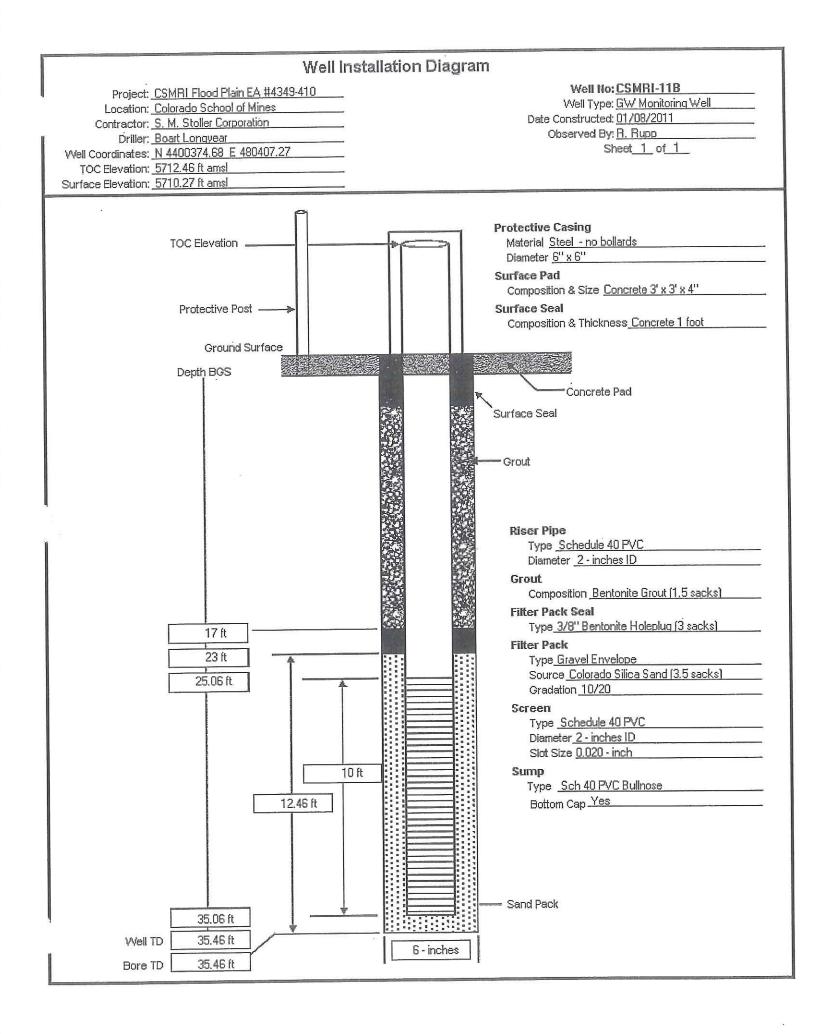


CSMRI Flood Plain EA Colora				Colorado School of	lo School of Mines		Golden, Colorad	
Project Number: 4349-410				Clier	Client: Colorado School of Mines			
Geologist: N. Malczyk				ning: 4400425.19 Easting:	480286.8	5.87		
Drilling Contractor: Boart Longyear			year		Total Depth of Borehole: 10 ft bgs			
Dates Drilled: 01/06/2011 - 01/06/2011			· · · · ·	h to Water: 7.82 ft btoc		4		
Borehole Type: 6 - inches Resonate Sonic					Type: 2-inch Sch 40 PVC			
Ground Elevatio	n: 568	1.25 ft	amsl	Тор	of Casing Elevation: 5683.22 f	t amsl		
Completion	Depth (Ft.)	Lithology	USCS	Descr	iption	Sample	GAMMA Screen (uR/hr)	
		0,00,00	P P	0 - 1' Gravelly, Sandy, Clay; Imp sand, 25% gravel; reddish brown grain sand, low plasticity, dry 1 - 7' Sandy Gravel; minus 6" Im cobbles, 25% sand; reddish brow	(5YR 5/3), fine and very fine ported Fill; 75% gravel and n (5YR 5/3), rounded to		Background 17 uR/hr	
			GW	sub-rounded gravel and cobbles, sub-angular sand; wet 5 - 7' bgs	hne and very fine grain		17 uR/hr	
	- 5		Bed Rx	7 - 10' Pierre Shale, gray (5YR 6 zone, damp at 7' to dry with dept	1		17 uR/hr	
	- 15			Total Depth of Borehole 10 feet Total Depth of Well 8.41 feet bg 2" Sch 40 PVC Blank Casing +1 2" Sch 40 PVC 0.020" Slot Scree 2" Sch 40 PVC Bullnose 8.01' to 10/20 Colorado Silica Sand 2' to Hydrated 3/8" Bentonite Holeplu Surface Seal poured with 3'x 3'x 6"x 6" square Steel Protective Ca	s 97' ags to 3.01' bgs en 3.01' to 8.01' bgs 8.41" bgs 10' bgs g surface to 2' bgs 4" Concrete Pad			

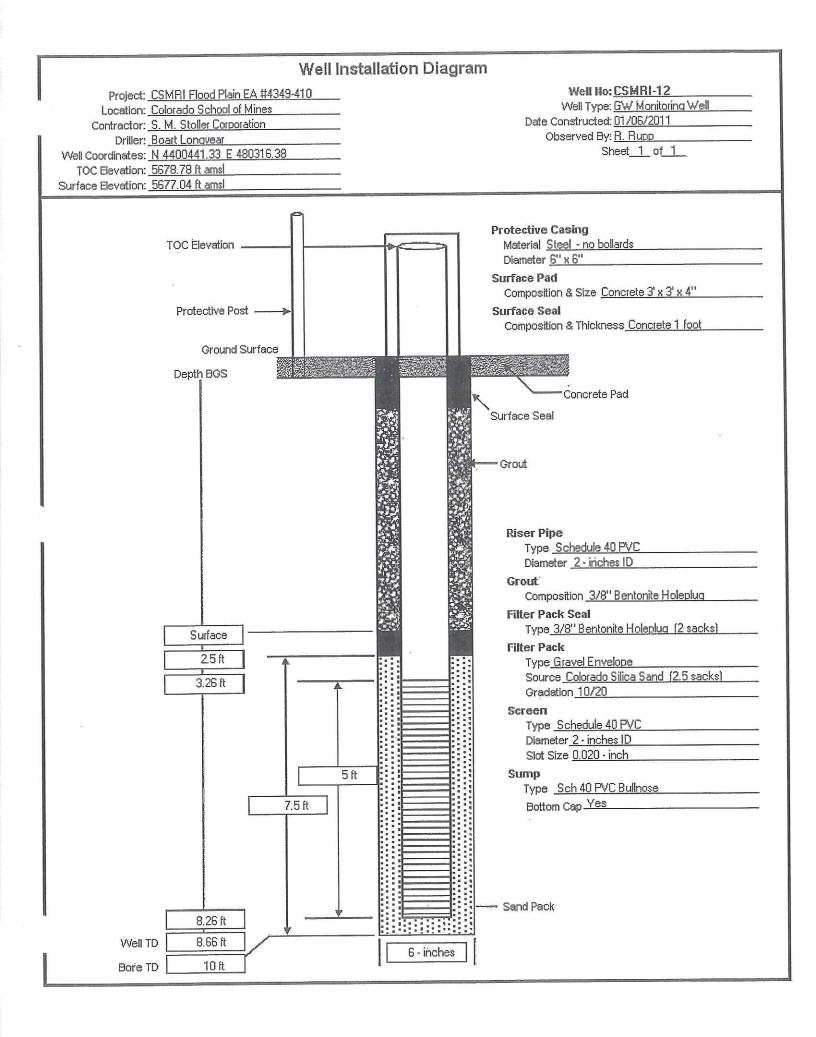


CSMRI Flood Plain EA	Colorado S	chool of Mines		Golden, Colorad	
Project Number: 4349-410		Client: Colorado School of Mines			
Geologist: P. Dalen		Northing: 4400374.68 Easting: 48	30407.2	7	
Drilling Contractor: Boart Long	year	Total Depth of Borehole: 35.46 ft bgs			
Dates Drilled: 01/07/2011 - 01	/07/2011	Depth to Water: 28.87 ft btoc			
Borehole Type: 6-inches I	Resonate Sonic	Well Type: 2-inch Sch 40 PVC			
Ground Elevation: 5710.27 ft	amsl	Top of Casing Elevation: 5712.46 ft a	msl		
Depth (Ft.) Lithology	USCS	Description	Sample	GAMMA Screen (uR/hr)	
	ML very fine and fine sand, t 4 - 5' split-spoon sample 5 - 10' Clayey Silt, olive sand, stiff, dry 9 - 10' split-spoon samp 10 - 15' Silty Sand, dark and fine sand, subrounded 14 - 15' split-spoon samp 15 - 21' Sand, dark brow little silt, trace gravel, in 20 - 21' split-spoon samp broken gravel 21 - 26' Sandy Gravel, I sand, very rough drilling encountered 25 - 26' split-spoon samp 26 - 30' Sandy Gravel (A medium sand to pebbles angular to sub-angular, J 30 - 35.46' Fox Hills Sa moist weathered zone (3 Rx competent rock (32 - 35') sub-rounded sand, trace brownish-red staining in staining in competent ro Total Depth of Borehole Borehole deepened from Total Depth of Well 35.	 blow count: 6, 13, 15, 10 brown (2.5Y 4/3), some very fine le blow count: 7, 12, 13 cyellowish brown (10YR 4/6), very fine ed, some silt ple blow count: 4,3,8 vn (7.5YR 4/4), very fine and fine sand, creasing sand with depth blow count: 50, refusal - contained ight olive brown (2.5Y 5/2), little fine g.2 - 24', 25 - 26', no water ple blow count: 23 - Auger Refusal 26' Alluvium), dark gray (5YR 4/1), and cobbles, with a few boulders, poorly sorted, moist to saturated ndstone, very dark gray (5YR 3/1) in 0 - 32') to medium gray (N5) in dry y. very fine to fine grain, sub-angular to interbedded shale, prominent weathered zone and minor orange ck 35.46 feet bgs 126' to 35.46' bgs with Sonic Rig 		Background 20 uR/hr 20 uR/hr 20 uR/hr 20 uR/hr	
45	2" Sch 40 PVC 0.020" S	asing +2.19' ags to 25.06' bgs Not Screen 25.06' to 35.06' bgs			
	2" Sch 40 PVC Bullnose 10/20 Colorado Silica S				
	Hydrated 3/8" Bentonite Bentonite Grout 0' to 17	Holeplug 17' to 23' bgs			
50	Surface Seal poured with	h 3'x 3'x 4" Concrete Pad ective Casing with Locking Cover			

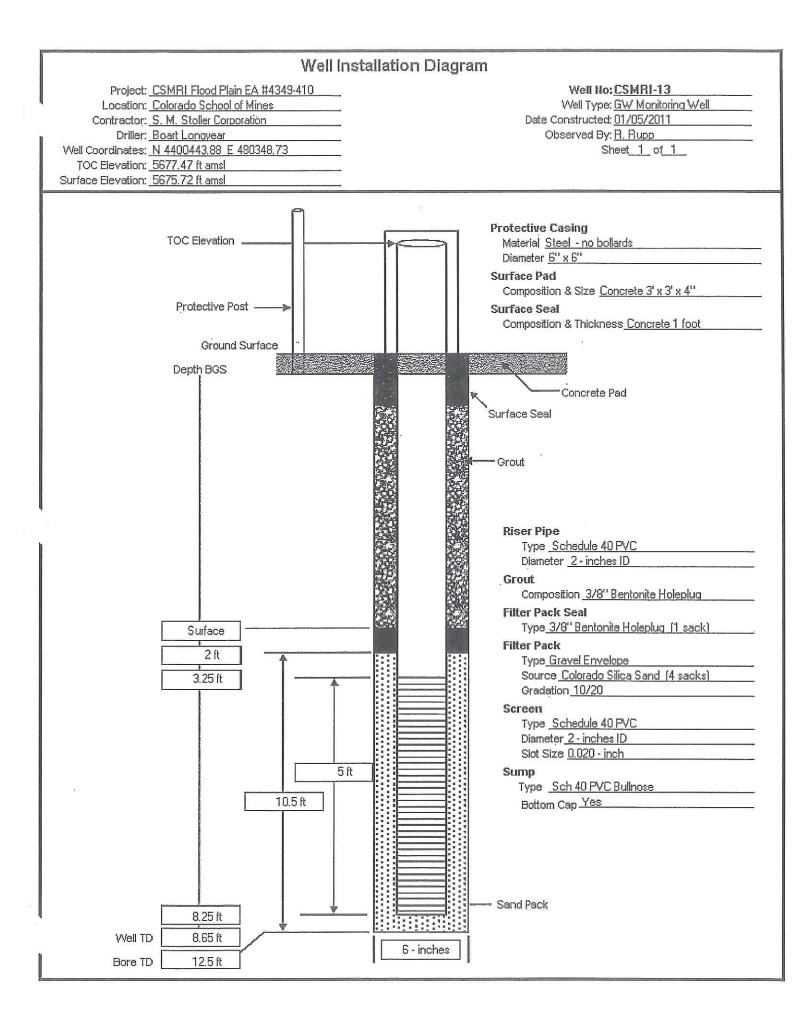
S.M. STOLLER CORPORATION



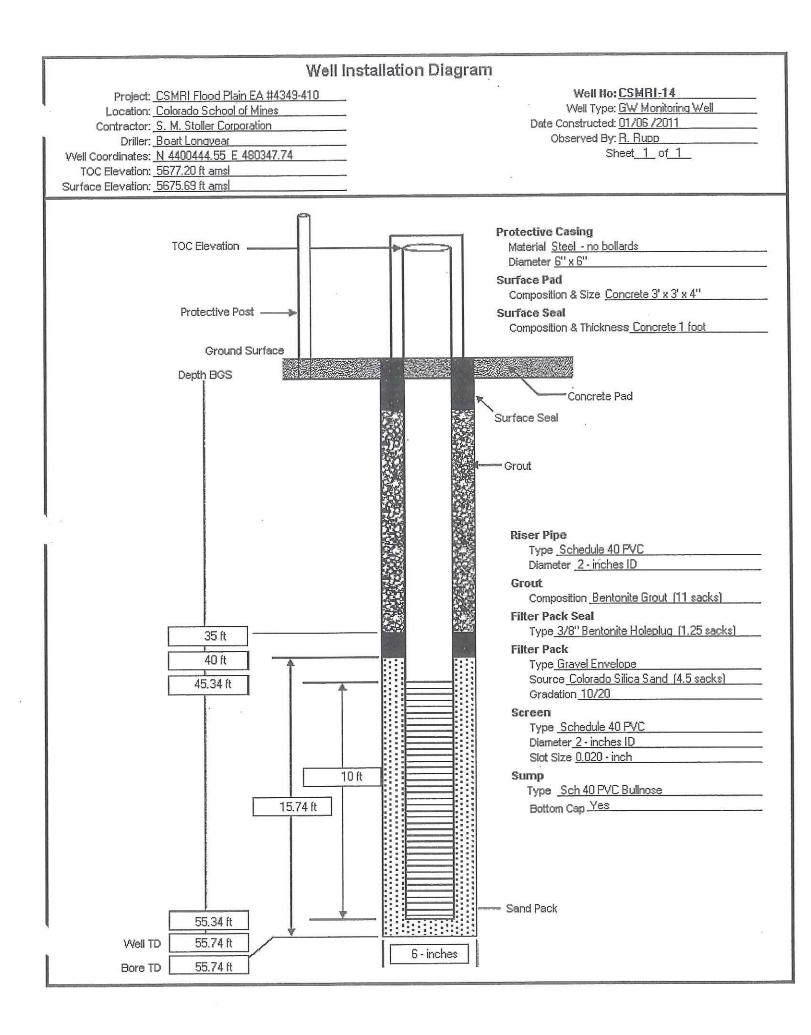
CSMRI Flood Plain EA Col	orado School of Mines	G	Golden, Colorad	
Project Number: 4349-410	Client: Colorado School of Mines	Client: Colorado School of Mines		
Geologist: N. Malczyk		480316.3	8	
Drilling Contractor: Boart Longyear		Total Depth of Borehole: 10 ft bgs		
Dates Drilled: 01/06/2011 - 01/06/2011	Depth to Water: 4.48 ft btoc			
Borehole Type: 6 - inches Resonate Sonic	Well Type: 2-inch Sch 40 PVC			
Ground Elevation: 5677.04 ft amsl	Top of Casing Elevation: 5678.78	ft amsl		
Depth (Ft.) USCS USCS	Description	Sample	GAMMA Screen (uR/hr)	
Image: Constraint of the state of the s	 Bandy Clay, 75% clay, 20% sand, 5% gravel, 5YR 3/1), medium plasticity, medium and fine mean sand, moist, organic Y Creek Alluvium, 70% sand, 30% gravel and brown (5YR 4/4) to reddish gray (5YR 4/2) ly sorted coarse, medium and fine grain 1, moist, becoming saturated below 2.3' bgs Bale, gray (5YR 6/1), moist weathered zone to iable, dry, competent shale 5.5' to 10' bgs Borehole 10 feet bgs Well 8.66 feet bgs Blank Casing +1.74' ags to 3.26' bgs 0.020" Slot Screen 3.26' to 8.26' bgs Bullnose 8.26' to 8.66' bgs Silica Sand 2.5' to 10' bgs Bentonite Holeplug surface to 2.5' bgs ured with 3'x 3'x 4" Concrete Pad teel Protective Casing with Locking Cover 		Background 22 uR/hr 24 uR/hr 23 uR/hr	
S.M. STOLLER CORPORATION				



		CSN	IRI-13		
CSMRI Flood Plain F	EA	Colorado S	School of Mines	Go	lden, Colorado
Project Number: 4349	9-410		Client: Colorado School of Mines		
Geologist: P. Dalen			Northing: 4400443.88 Easting:		3
Drilling Contractor: I			Total Depth of Borehole: 12.5 ft bgs	6	
Dates Drilled: 01/05/	and the second se		Depth to Water: 4.39 ft btoc Well Type: 2-inch Sch 40 PVC		
Borehole Type: 6 -in Ground Elevation: 5		nate Sonic	Top of Casing Elevation: 5677.47 ft	amel	
Completion (1) Debth (1)	USCS USCS		Description	Sample	GAMMA Screen (uR/hr)
	OL OC	sub-rounded, graded san 1 - 5' Post-Piney Creek	dark reddish brown (5YR 3/2), some nd, moist, organic Alluvium, well graded sand, gravel, size distribution increasing with depth,		Background 20 uR/hr
	0000 GW		4), fine to very coarse sub-angular sand;		22 uR/hr
- 5	- 000 SP	fine and fine grain, sub- extensive brownish-ora weathered top of bedroo 6.5 - 12.5' Laramie-Fox	K Hills Sandstone, light gray (7.5YR grain, sub-rounded, well sorted, medium		20 uR/hr
- 10	Bed 		agent kourt trou		21 uR/hr
	_				21 uR/hr
- 15 - 15 		2" Sch 40 PVC 0.020" 2" Sch 40 PVC Bullnos 10/20 Colorado Silica S Hydrated 3/8" Bentonit Surface Seal poured with	55 feet bgs Casing +1.75' ags to 3.25' bgs Slot Screen 3.25' to 8.25' bgs te 8.25' to 8.65' bgs		
S.M. STOLL	ER COF	PORATION			Page 1



CSMRI Flood Plain EA Col	orado School of Mines	G	olden, Colorad	
Project Number: 4349-410	Client: Colorado School of Mine	3		
Geologist: P. Dalen		: 480347.	74	
Drilling Contractor: Boart Longyear	Total Depth of Borehole: 55.74 f	t bgs		
Dates Drilled: 01/05/2011 - 01/05/2011	Depth to Water: 3.72 ft btoc			
Borehole Type: 6 - inches Resonate Sonic	Well Type: 2-inch Sch 40 PVC			
Ground Elevation: 5675.69 ft amsl	Top of Casing Elevation: 5677.2) ft amsl		
Depth (Ft.) uoised	Description	Sample	GAMMA Screen (uR/hr)	
OL 0 - 1' Sandy Silt	Clay, dark reddish brown (5YR 3/2), fine to	7	Background 20 uR/hr	
\times \times \rightarrow \rightarrow \sim \rightarrow \sim \rightarrow \sim \sim \rightarrow \sim	ub-angular, graded sand, moist, organic ey Creek Alluvium, well graded sand, gravel,	_	22 uR/hr	
cobbles, and bour reddish brown (i	Iders, size distribution increasing with depth, YR 4/4), fine to very coarse sub-angular sand,	/ī	21 uR/hr	
sP 5.5 - 6.5' Laram	e-Fox Hills Sandstone, brown (7.5YR 5/2), e grain, sub-rounded, well sorted, medium red-orange iron staining, weathered top of	/	20 uR/hr	
bedrock			20 uR/hr	
	mie-Fox Hills Sandstone, light gray (7.5YR	_	20 uR/hr	
	d fine grain, sub-rounded, well sorted, medium noist, competent bedrock	1	20 uR/hr	
		2	20 uR/hr	
$30 - \frac{8}{R_x}$			20 uR/hr	
			20 uR/hr	
			20 uR/hr	
45			20 uR/hr	
	2		20 uR/hr	
			20 uR/hr	
— — — Total Depth of V — 60 — 2" Sch 40 PVC	orehole 55.74 feet bgs Vell 55.74 feet bgs Blank Casing +1.51' ags to 45.34' bgs			
65 2" Sch 40 PVC 10/20 Colorado	0.020" Slot Screen 45.34' to 55.34' bgs Bullnose 55.34' to 55.74' bgs Silica Sand 40' to 55.74' bgs entonite Holeplug 35' to 40' bgs			
70 Bentonite Grout Surface Seal por	surface to 35' bgs red with 3'x 3'x 4" Concrete Pad cel Protective Casing with Locking Cover			



Appendix B Sample Collection Forms

Groundwater Sampling

1.0 Purpose

This procedure describes actions to be used to sample groundwater from monitoring wells and piezometers. Monitoring wells are generally sampled on a semiannual, quarterly, or monthly basis, or by special request in support for specific projects. All wells are to be sampled using this procedure unless superseded by specific site, facility, or client procedures.

This procedure describes equipment decontamination and transport, site preparation, detection and sampling of immiscible layers, water level measurements, well purging, sample collection, field and analytical parameters, quality assurance/quality control (QA/QC) requirements, and documentation that shall be used for field data collection.

2.0 Scope

This document describes acceptable methods for the sampling of wells and piezometers.

3.0 Responsibilities and Qualifications

Personnel performing groundwater sampling procedures are required to have completed the initial 40-hour OSHA classroom training that meets the Department of Labor requirements at 29 CFR 1910.120(e)(3)(i), and must maintain a current training status by completing the appropriate annual 8-hour OSHA refresher courses. Personnel must also have read the appropriate project, site, or facility Health and Safety Plan(s). Prior to engaging in groundwater sampling activities, personnel must have a complete understanding of the procedures described within this procedure and, if necessary, will be given specific training regarding these procedures by other personnel experienced in the methods described within this procedure.

4.0 Groundwater Sampling Procedures

4.1 Introduction

Many monitoring wells are constructed of either 2-inch stainless steel, or 2- or 4-inch flush threaded PVC casing. Some piezometers are completed as monitoring wells, and they are usually constructed of ¾-inch inside diameter, flush threaded PVC casing. Some wells have been constructed to incorporate a sump below the well screen. Because these vary in length, the well construction diagrams should be consulted to determine the sump lengths for specific wells. Most piezometers are constructed with a flush threaded cap at the bottom of the well screen. However, the well construction diagrams should also be consulted for information about specific piezometers.

Procedures for groundwater sampling are designed to obtain a sample that is representative of the formation water beneath the site in question. Since an analysis of the quality of formation water is desired, standing water within the well must be purged before sampling. Also, a measure of the static water elevations is important to determine the effect of seasonal horizontal and vertical flow gradient changes during site characterization activities.

Groundwater sampling procedures can be initiated after sampling personnel take the required water level measurements and purge the well in accordance with this procedure. Methods for accomplishing each of these activities are included in this procedure in the following sequence:

- Collection of immiscible layers samples, if present
- Well purging
- Groundwater sampling using a bailer

- Groundwater sampling using a peristaltic pump
- Groundwater sampling with a bladder pump

4.2 General Equipment Requirements

Down-hole sampling equipment shall be constructed of inert material such as polytetrafluoroethylene (Teflon[®]) or stainless steel. This equipment shall be assessed on an individual basis prior to use in the field.

The following is a primary list of well sampling and associated equipment:

- Bailers Teflon[®], stainless steel, or other appropriate inert materials
- Teflon[®] coated stainless steel cable with reels
- Peristaltic pumps and tubing
- Water level measuring devices sufficiently accurate to measure water levels to the nearest 0.01 foot
- Graduated purge water containers
- Plastic sheeting
- Distilled or deionized water
- Decontamination equipment and supplies
- Organic vapor detector (OVD)
- Gloves (nitrile)
- Calculator and watch
- Sample containers precleaned to EPA specifications
- pH paper
- Custody tape
- Coolers with sufficient blue ice to cool samples to 4°C
- Preservatives (trace metals grade)
- Disposable in-line 0.45-micron membrane filters
- Logbooks and field forms
- Black waterproof pens
- Portable laboratory equipment for measuring field parameters for pH, temperature, specific conductance, and turbidity
- Total alkalinity reagent
- Beakers and graduated cylinders

Additional equipment may be required to meet project or client health and safety standards, to perform specialized sampling, or to meet personnel and equipment decontamination requirements.

4.3 Equipment Decontamination and Transport

Equipment associated with the tasks involved in groundwater sampling shall be decontaminated upon arrival at the sampling location. All sampling equipment shall be decontaminated between

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sample locations. Decontamination frequency shall be increased appropriately as field conditions dictate.

Transportation of all equipment shall be performed in a manner that eliminates any possibility of cross-contamination. Calibration solutions, fuel, decontamination solutions and wastewater, and all other sources of contamination shall be segregated from sampling equipment during transport. Purge water being transported to holding areas shall be kept in closed containers.

If the decontamination of downhole equipment is not performed at the well, used downhole equipment shall be wrapped in plastic sheeting and/or segregated from clean equipment to eliminate the possibility of cross contamination. The equipment shall then be decontaminated as soon as possible.

4.3.1 Routine Field Decontamination

Decontamination of delicate equipment and the routine decontamination of sampling equipment prior to use at each well shall consist of the following steps:

- Vigorously scrub the equipment with a brush and solution of phosphate-free laboratory grade detergent (e.g., Liquinox) and distilled water.
- Rinse the equipment thoroughly with approved distilled water.
- If the decontaminated equipment is not immediately packaged to eliminate any adhesion of airborne impurities, perform an additional final rinse, or decontamination and rinse, immediately prior to actual sampling operations.

4.3.2 Routine Decontamination of Sampling Pumps

The external surfaces of all non-dedicated pumping equipment shall be decontaminated as described in Subsection 4.3.1. Internal surfaces shall be decontaminated according to the following procedures, except under special situations where the pump(s) must be disassembled and the internal parts cleaned separately (see Subsection 4.3.3). For routine decontamination, the following procedures shall be followed.

- Pump several pump volumes of a solution of a phosphate-free laboratory grade detergent (e.g., Liquinox) and water through the equipment.
- Displace the soap solution immediately by pumping approved distilled water, equivalent to three or more volumes of the pump storage capacity, through the equipment.
- If any detergent solution remains in the pump, continue pumping distilled water through the system until the detergent is no longer visibly present. Sudsing is the common indicator used to determine incomplete rinsing.
- 4.3.3 Unusual Decontamination Requirements

When equipment becomes grossly contaminated, such as from the collection of immiscible layer samples (see Subsection 4.5), routine decontamination of sampling equipment is not considered sufficient and thus is not allowed. This situation and other unusual equipment decontamination problems shall be reported to the field site supervisor. Under certain circumstances, a pump can be disassembled and the parts cleaned separately using approved solvents (i.e., hexane, alcohol, etc.). If specific instructions are required, the field site supervisor shall consult with a management representative for proper decontamination procedures.

4.3.4 Disposition of Decontamination Water

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All water generated during the decontamination of equipment used for the sampling of wells shall be containerized in either a satellite container or in the purge water container in the groundwater sampling vehicle. It will then be disposed of according to the procedure designated in Subsection 4.6.3 of this procedure.

4.4 Site Preparation

Sheet plastic may be used to protect clean equipment from contacting contaminated surfaces. Plastic bags and sheeting, along with the segregation of clean and dirty equipment, can be used to reduce the chances of cross contamination. If a mechanical bailer retrieval system is used, the amount of plastic appropriate for protection of sampling equipment may be lessened. The sampling crew members are responsible for determining the amount of plastic sheeting required.

Disposable nitrile gloves, or gloves made of other approved materials, shall be used at all times when handling sampling equipment. Gloves shall be changed between each site and as often as necessary to ensure the integrity of clean sampling equipment.

4.5 Collection of Immiscible Layer Samples

When specified in the project sampling plan, or when the well to be sampled contains immiscible layers, immiscible phases must be collected before purging activities begin. The method of choice for collecting light non-aqueous phase liquids (LNAPLS) is a bottom valve bailer or peristaltic pump. Dense non-aqueous phase liquids (DNAPL) or "sinkers" shall be collected with a bottom double check valve bailer or peristaltic pump.

In all cases, the bailer shall be carefully lowered into the well so that agitation of the immiscible layer is minimal. Any bailer used to collect immiscible layers shall be dedicated to the well that is sampled. Peristaltic pumps shall be equipped entirely with silicon, or other chemical compatible tubing, when sampling immiscible layers. The project manager shall be responsible for determining the type materials to be used for specific projects. Dedicated equipment used for collecting immiscible layers shall be decontaminated prior to and after use as described in Subsection 4.3 of this procedure, if removed from the well.

Immiscible layer sampling shall be performed as follows.

- Remove dedicated bailers from the well and decontaminate as specified in Subsection 4.3 of this procedure. Decontaminate dedicated pump tubing, if used, prior to use.
- For LNAPLs, carefully lower the bailer intake or sampling port to the midpoint of the immiscible layer and allow it to fill while it is held at this level. The bailer must be lowered into the immiscible layer slowly so that minimal agitation of the immiscible layer occurs. Peristaltic pump intakes must also be lowered to the midpoint of the immiscible layer.
- If a DNAPL layer is being sampled, use either the double check valve bailer or peristaltic pump. Lower the bailer into the well until bottom is encountered. Lower peristaltic pump intakes also to the well bottom. Care must be taken not to immerse the pump intake into accumulated sediments.
- Do not allow the bailer or line to touch the ground at any time or allow the ground to come in contact with other physical objects that might introduce contaminants into the well.
- Decontaminate all equipment immediately after sampling is completed. Suspend dedicated bailers in the well from the well cap above the high water level. Discard silicon tubing used with peristaltic pumps.

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4.6 Well Purging

Purging stagnant water from a well is required so that the collected sample is representative of the formation groundwater. The device used (bailer or pump) depends upon aquifer properties, individual well construction, and data quality objectives. Wells that contain immiscible layers will not be purged unless specified in the site-specific work plan. Any well scheduled for purging and sampling that subsequently is found to contain immiscible layers must be reported to the site supervisor or project manager. The project manager shall be notified immediately prior to continued activities.

Before obtaining water level elevations or initiating purge activities, obtain the following information in reference to the well to be sampled, and enter the applicable information on the sample collection log.

- Location code (well number)
- Previous purge volume (information only)
- Depth to top of screen (bailed wells only)
- Well sample number
- Report Identification Number (RIN)
- Sample event number

Record the location code (well number), date, sampling team members, visitors, well condition, and any other pertinent information on the sample collection log. Enter the well number, time well is opened, and other information regarding the field activities on the Field Activity Daily Log.

The field instruments shall be standardized (to check calibration) and the results recorded on the sample collection form.

Measure the depth to the top of the water column and the total depth of the well in order to determine the height of the water column in the well. Calculate the well casing volume using the well casing inner diameter and the height of the water column in the well. The formula for calculating the volume in gallons of water in the well casing is as follows:

 $(\pi r^2 h)$ 7.481 = gallons; where

 $\pi = 3.142$

r = inside radius of the well pipe in feet

h = linear feet of water in well

7.481 = gallons per cubic foot of water

1 gallon = 3785 ml

Calculations of the volume of water in typical well casings may be done as follows:

a. 2" diameter well:

0.16 gal./ft x (linear ft of water) = gallons of water

b. 4" diameter well:

0.65 gal./ft x (linear ft of water) = gallons of water

c. 3/4" diameter well:

87 ml./ft x (linear ft of water) = milliliters of water

4.6.1 Purging Duration

Purging shall be considered complete if any of the following conditions are met.

- Purging is complete if at least three casing volumes of water are removed from the well, and the last three consecutive pH, specific conductance, and temperature measurements do not deviate by more than the following: 1) pH = ±0.1 pH units; 2) Specific Conductance = ±10% and; 3) temperature ±0.5°C. A turbidity measurement will be taken for every other purge sample for wells that are purged using a bailer. For wells that are equipped with a dedicated bladder pump, the turbidity will be measured each time the parameters are taken. The purge rate should be such that the turbidity is maintained at 5 NTU units or less (if possible). If the readings are not stabilized after three volumes, continue purging until stabilization or until five volumes have been removed. Field parameter measurements shall be collected after every half-casing volume (approximate) is removed from the well. When casing volumes are less than 1-liter, parameter measurements will be collected after each whole casing volume is removed. If readings do not stabilize after five well volumes have been recovered, obtain additional guidance from the project manager concerning the proper course of action.
- 2. A well is considered dewatered when only a few milliliters of water (or none) can be recovered each time the bailer is lowered into the well. When this occurs, a 10-minute recharge rate will be calculated (linearly). If, at the end of the 10-minute period, the well has not recovered sufficiently to continue the purge in thirty minutes, the purge is considered completed. If, at the end of the 10-minute period, there is sufficient water to collect the VOA samples, the samples may be collected at that time. If the well has not recovered sufficient water during the 10 minutes, and depending upon the well history, the samplers may elect to return to the well the same day (preferably within two hours), check the water level, and collect the VOA samples (first), and other samples as feasible. If the sample team cannot return the same day, the well will be checked in 24 hours to determine if sample collection is feasible. If an extended period of time is required to collect samples, the procedures in Subsection 4.8.1 shall be followed. The well will not require an additional purge before sampling.

Wells that dewater (have a slow recharge rate as specified in 2 above) will not be restricted by parameter stabilization requirements. Sampling of these wells will follow the protocol established in Subsection 4.8.

4.6.2 Purging Methods

Wells will be purged by either bailing or pumping. When purging a well, the rate of water withdrawal during purging should not exceed the rate of withdrawal at which the well was developed (if known). All purge times (initiation and completion) and the rate of purging will be recorded on the field log sheets.

4.6.2.1 Bailing

Generalized procedures for purging a well with a bailer are as follows.

• Prepare the sampling site as discussed in Subsection 4.4. Use properly decontaminated equipment to determine the static water level of the well. Measure the total depth of the well. Use this information to determine the volume of water in the well casing.

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- Decontaminate all dedicated bailers prior to initiating purging as described in Subsection 4.3 of this procedure.
- Use a mechanical reel equipped with Teflon[®] coated stainless steel cable attached to a bailer for bailing and sampling operations. Lower the bailer slowly into the well until water is encountered. Minimize agitation of the well water. Avoid lowering the bailer to the bottom of the well so sediments accumulated in the bottom do not become suspended. For wells that dewater, do not allow the bailer to strike the well bottom with force. Raise and lower the bailer carefully to limit surge energy and ensure that cable does not come in contact with any potentially contaminated surfaces. Do not allow the cable to drag along the well casing or against other objects that will cause fraying. Monitor the amount of water purged.

Wells with significant levels of contamination may have dedicated bailers installed. Dedicated bailer systems shall consist of a Teflon[®] bailer with check valve or double check valve for DNAPLS and a 5-foot leader of Teflon[®] coated stainless steel cable. Bailer sampling attachments and the stainless steel reel cable will not be dedicated to individual wells.

Dedicated bailers will be decontaminated at the conclusion of sampling activities and suspended from the well cap above the high water table. If the well interval above the high water table is not adequate to allow for storage in the casing, the dedicated bailers will be stored in labeled and sealed plastic bags at the equipment trailer.

4.6.2.2 Pumping

Pump designs that meet the following criteria are allowed for purging.

- The pump is constructed of a material that does not introduce a source of contamination to the well.
- The pump drive system does not introduce a source of contamination into the well.
- All downhole parts to the pump can be easily decontaminated.
- A return check system that does not allow pumped water to return to the well is integral in the pump design.
- The pump is easily used and does not require excessive amounts of time to install, use, remove, and decontaminate.

The pumps currently in use to purge groundwater include peristaltic pumps and dedicated submersible bladder pumps. A procedure for the use of each style of pump is specific to its applications. User manuals, which accompany each pump, shall be referenced for operating procedures.

Basic operating procedures common to all pumps are as follows.

- Prepare the sampling site as described in Subsection 4.4 regardless of the type of pump being used.
- Use properly decontaminated equipment to determine the static water level and the total depth of the well. This information is utilized to determine the volume of water in the well casing.

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- For wells with dedicated pumps, calculate the minimum purge volume using the pump storage volume and the volume of the discharge tubing. A total depth of a 2-inch well cannot be taken without the removal of the pump.
- Position a dedicated pump near the bottom of the well or according to the information on the well construction form. Monitor the discharge rates and the amount of water purged during purging. The pumping rate for purging can be higher than the pumping rate for sampling, however, the water level in the well should be monitored during purging to avoid excessive water level drawdown.
- Ensure that any tubing that enters the well casing is composed of inert material. Disposable silicon tubing will be used in the drive mechanism of peristaltic pumps and discarded after each well is purged. The air supply for all air-driven pumps (dedicated bladder pumps) will be free of oil (i.e., no hydrocarbon containing substances will be added to the compressor).

4.6.3 Disposition of Purge Water

All water removed from a well during sampling operations shall be collected either in a satellite container or the purge water collection container in the groundwater sampling vehicle. The water from these containers will then be transferred to another approved collection container on the sampling or project site. When the collection container is filled, or is near capacity, it will be transported for disposition or treatment in accordance with approved project plans.

4.7 Measurement of Field Parameters

The following field parameters will be measured during groundwater purging operations unless otherwise specified by the project manager or the approved project work plans.

Parameter	Relative Precision	Minimum Calibration
pH	0.01 pH units	Daily
Conductivity	10 µS/cm	Daily
Temperature	0.1 °C	Weekly
Total Alkalinity (unfiltered)	1 mg/l	None
Turbidity (photometric)	2 FTU (or NTU)	Specified purge samples (bailed wells) Daily (dedicated bladder pump wells)

The measuring equipment shall be stored and handled in a manner that will maintain the integrity of the equipment. Appropriate field manuals will accompany each instrument in the field. Each instrument will also be given an identification number. All logbook and field form references to individual instruments will refer to this number for ease of identification.

Field parameters will be measured at the following intervals.

• Conductivity, pH, temperature, and turbidity shall be measured from the first water removed from the well when initiating well purging procedures. For bailed wells, the initial bail of water will be carefully removed from the well and the water transferred to a sample beaker by decanting the bailer through a bottom control valve. For wells

purged with a peristaltic pump, similarly collect the first water removed in a sample beaker and then measure parameters. For wells with dedicated pumps, measure the parameters of the first recovered water that is collected in the continuous sampler.

- During purging operations, conductivity, pH, and temperature shall be measured for every half-casing volume (one half of the initial casing volume as calculated on the sample collection log form) of water removed from the well (because of the accuracy of the graduated containers for the purge water, the purge volume will be estimated as close as feasible). For wells that have half volumes less than the volume of a sample bailer (approximately 1 liter), only measure parameters after each full casing volume of water is removed from the well. Turbidity will be measured on every other sample recovered for parameters for bailed wells, or wells purged with a peristaltic pump. All parameters, including turbidity, will be measured at predetermined intervals while purging wells with dedicated pumps.
- During purging, if a well is dewatered prior to the measurement of the final required set of parameters, then conductivity, pH, temperature, and turbidity shall be measured immediately before the start of sample collection. These parameters may be delayed until sampling is completed if, at the discretion of the sampling crew, the well recharge has provided insufficient water volume to collect all the samples and also measure parameters. If there is insufficient water for samples and field parameters, the parameters will not be measured.
- Total alkalinity measurements shall be collected only once upon completion of purging. For wells that do not dewater and sample collection proceeds to completion immediately after purging, alkalinity will be measured after the completion of all other final purge field parameters. Wells that dewater and require repeated visits for the collection of samples will have alkalinity measured subsequent to the collection of the sample for inorganic water chemistry. Alkalinity will not be measured if sufficient water is not available.
- For micro purged wells, a purge is considered completed when the parameters have stabilized.
- Whenever a method used to remove well water is changed, a set of field parameters shall be recorded from water removed with the new method.

4.8 Groundwater Sampling

Techniques used to withdraw groundwater samples from a well shall be based on consideration of the parameters of interest. The order of collection, collection techniques, choice of sample containers, preservatives, and equipment are all critical to ensuring that samples are not altered or contaminated. The preferred methods for collection of groundwater samples are either bailing and/or the use of bladder pumps.

Sites shall be prepared prior to sampling as described in Subsection 4.4. All necessary and appropriate information will be recorded on the sample collection log and on the Field Activity Daily Log.

4.8.1 Sample Collection

The following discussion involves collection of groundwater samples using bailers and peristaltic or bladder pumps. Regardless of the collection method, care shall be taken not to alter the chemical nature of the sample during the collection activity by agitating the sample or allowing prolonged contact with the atmosphere. To minimize the potential for

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altering the sample and to maximize the available water, the following sample collection sequence is preferred.

- Radiation Screening
- VOC
- Nitrate/Nitrite, as N
- Dissolved Metals TAL, with Cs, Li, Sr, Sn, Mo, Si
- ^{239/240} Plutonium, ²⁴¹ Americium
- ^{233/234}U, ²³⁵U, ²³⁸U
- Gross alpha and beta
- ^{89/}Strontium
- ¹³⁷Cesium
- ^{226,228}Radium
- Tritium
- Total Metals TAL, with Cs, Li, Sr, Sn, Mo, Si
- TDS, CL, F, SO⁴, CO3, HCO³
- TSS
- BNA
- Pesticides/PCB
- Cyanide
- Orthophosphate

VOC samples shall be collected first and as soon as possible after the well has been purged. If a well is purged using a peristaltic pump, then all other samples shall be collected prior to removing the pump from the well. The VOC sample will then be collected using a bailer.

For wells that dewater, if a sufficient volume of water for VOC sample collection has still not accumulated within 48 hours after the completion of purging, VOCs will not be collected for that well. Other samples may be collected using a maximum of five attempts to recover sufficient sample water for analysis. This procedure is discussed in the following paragraph.

The containers used for sample collection from poor producing wells may differ from those used for high yield wells in some instances due to constraints on obtaining enough sample to fill sample containers. In some instances smaller containers may be utilized, or analyte samples normally collected in separate containers may be combined into a single container. Well histories can be used to identify which wells may require a modified sample suite and an extended sampling period. These wells will initially be sampled for a period of 48 hours after the completion of purging, with the exception of VOC sample collection, which is discussed in the previous paragraphs. The completion of purging will be considered 0 hour. At the end of 48 hours, any partial sample will be measured. The accumulated sample will be compared to the minimum volume requirement identified in Table 1 and the allowed sample holding time. If the minimum volume requirement for the target analyte has not been achieved, then sampling may continue as determined from the well recharge history. All analyte samples that have only minimum sample volumes collected, and all uncollected samples will be documented on the sample collection log.

Table 1

Sample Containers and Preservatives for Groundwater Samples

Parameter	Minimum Container ¹	Preservative	Holding Time
Radiation Screen	120 ml poly	None	NA
VOC - CLP	3 – 40 ml amber glass	Cool to 4° C	4 Days
BNA	1 L amber glass	Cool to 4° C	7 Days
Pesticides/PCB	1 L amber glass	Cool to 4° C	7 Days
TSS	125 ml poly	Cool to 4° C	7 Days
TDS, CI, F, SO4, CO3, HCO3	1 L poly	Cool to 4° C	7 Days
Dissolved Metals - CLP, with Cs, Li, Sr, Sn, Mo, Si	1 L poly	*Filtered, HNO ₃ to pH <2, Cool to 4° C	6 Months
ТОС	125 ml poly	H ₂ SO ₄ < pH2, Cool to 4° C	28 Days
COD	125 ml poly	H ₂ SO ₄ < pH2, Cool to 4° C	28 Days
Total Metals - CLP with Cs, Li, Sr, Sn, Mo, Si	1 L poly	Unfiltered, HNO ₃ to pH <2, Cool to 4° C	6 Months
Orthophosphate	250 ml poly	Filtered, Cool to 4° C	2 Days
Nitrate / Nitrite as N	250 ml poly	H ₂ SO ₄ to pH <2, Cool to 4° C	28 Days
Cyanide	1 L poly	NaOH to pH >12, Cool to 4° C	14 Days
Gross Alpha / Beta	550 ml poly	HNO₃ to pH <2	6 Months
233/2340, 2350, 2380	100 ml poly	Filtered, HNO₃ to pH <2	6 Months
239/240Pu	1 L poly	HNO₃ to pH <2	6 Months
²⁴¹ Am	1 L poly	HNO₃ to pH <2	6 Months
^{89/90} Sr	700 ml poly	Filtered, HNO₃ to pH <2	6 Months
^{226/228} Ra	750 ml poly	Filtered, HNO₃ to pH <2	6 Months
¹³⁷ Cs	2.5 L poly	Filtered, HNO ₃ to pH <2	6 Months

¹ The volume listed is the minimum amount required for analysis. Actual sample volumes may be slightly higher and some parameters may be combined in a single container.

* Some samples may not require filtering if taken from a well with a dedicated pump and turbidity of 5 NTU or less.

The order of sample collection may be changed at the discretion of the sampling team. Changes in the order shall be based on the predicted volume of water that will be recovered and the priority stated in the controlling document. The sampling team shall document their sample selections on the sample collection log.

Sample containers shall be stored away from sunlight and cooled to 4°C prior to filling. Immediately after collection, samples requiring cooling shall be cooled to 4°C. A chilled cooler shall be used as the storage container. Whenever a sample bottle that requires chilling is not being physically handled, it will be placed in the cooler to prevent heating or freezing, exposure to sunlight, and possible breakage.

VOC samples shall be collected using a bailer equipped with a bottom-decanting control valve or directly from the pump discharge line on wells equipped with bladder pumps. The procedures for collecting VOC samples are discussed in Subsections 4.8.1.1 and 4.8.1.2 of this procedure.

VOC vials shall never be filled and stored below capacity because of insufficient quantities of water in the well. Except for the VOC vials, adequate air space should be left in the sample bottles to allow for expansion.

Samples shall be placed in the appropriate containers and packed with ice in coolers as soon as practical. VOC samples will be stored in the cooler in an inverted position immediately after collection. When sampling is complete, the well cap shall be replaced and locked.

Sampling tools, instruments, and equipment shall be protected from sources of contamination before use and decontaminated after use as specified in Subsection 4.3. *Liquids from decontamination operations will be handled in accordance with the procedures in Subsection 4.6.3 of this procedure.* Sample containers shall also be protected from sources of contamination. Sampling personnel shall wear chemical-resistant gloves (e.g., nitrile) when handling samples, and the gloves will be disposed of between well sites.

4.8.1.1 Groundwater Sampling Using a Bailer

This subsection describes the use of a bailer for collecting groundwater samples that may be used to obtain physical, chemical, or radiological data.

A bailer attached to a Teflon[®] coated stainless steel cable is carefully lowered into the well. After filling within the well, the bailer is withdrawn by rewinding the bailer line, and the bailer contents are drained into the appropriate containers. Certain recommendations and/or constraints should be observed when using bailers for sampling groundwater monitoring wells, as follows.

- Use only bottom-filling Teflon[®] bailers or bailers made of other inert materials.
- Ensure that bailers are attached to a Teflon[®] coated stainless steel line that is pre-wound on a reel.
- Do not use bailers constructed with adhesive joints.
- Lower the bailer slowly to the interval from which the sample is to be collected.

VOC samples shall be collected using a bailer equipped with a bottom-decanting control valve. The first water through the valve assembly will be discarded into the purge water container. Vials will be filled by dispensing water through the control valve along the inside edge of the slightly tilted sample vial. Care shall be taken to eliminate aeration of the sample water. The vials will be filled beyond capacity so the resulting meniscus will produce an airtight seal when capped. The capped vial will be checked for trapped air by lightly tapping the vial in an inverted position. If air becomes trapped in the vial, the sample water shall be discarded, and the vial refilled. If two consecutive attempts to fill a VOC vial result in trapped air bubbles, the vial shall be discarded.

The remainder of the sampling water shall be collected in a stainless steel container from which the remaining sample bottles will be filled. Samples requiring filtration shall be filtered and then containerized.

4.8.1.2 Groundwater Sampling Using a Peristaltic Pump

Use of peristaltic pumps shall generally be limited to collecting sample aliquots for radionuclides, metals, and other species that are not subject to volatilization and degassing. Peristaltic pumps shall never be used to collect VOCs or other

volatile species in routine wells, although such samples may be collected for special screening applications. All downhole tubing shall be Teflon[®] except in areas of special concern (e.g., where immiscible layers exist) where special tubing, such as stainless steel or Viton[®], may be required. If so, the project manager will make this determination. Only the portion of tubing that is inserted into the mechanical drive shall be made of silicon. This drive portion of the tubing shall be discarded after each use.

4.8.1.3 Groundwater Sampling Using a Downhole Bladder Pump

Some wells are equipped with dedicated downhole bladder pumps for purging and sampling. These are wells that will normally produce an adequate amount of water during a single visit to complete the required sampling suite. The equipment required to purge and sample a well consists of a pump control unit, a portable air compressor, a continuous sampler for measuring the field parameters, and the necessary sample containers, graduated cylinders, and container(s) to collect the purge and excess water. The following precautions should be observed during the sampling operation.

- Locate the compressor used to power the pump downwind from the well to eliminate the contamination of equipment and samples with exhaust.
- If the flow-through cell will not maintain a full sample chamber (tends to drain back), then clean the check valve on the pump if it is fouled, or replace the pump.
- Calculate the minimum purge volume using the procedure in Section 4.6. Note that a purge is considered completed only when the groundwater parameters have stabilized.
- Upon completion of purging, initiate sampling with the collection of the VOC sample(s). The pump should operate with minimum interruptions while the full sample suite is collected. Allowing the pump to stop for an extended period of time will cause the water trapped in the discharge lines to equilibrate to ambient temperatures, which is not acceptable. During sampling, the pump can be slowed to any rate that allows efficient sampling while also maintaining stable field parameters.
- Measure groundwater parameters periodically during sample collection and record them on the sample collection log to document conditions during sampling.
- Because micropurging is the method used for sampling, adjust the flow rate to limit the drawdown in the well. Also adjust the rate such that the turbidity is below 5 NTU for sampling. If this criterion is met, the samples need not be filtered.
- Operate the pump, pump control unit, and the flow-through cell according to the manufacturer's recommendations.

4.8.1.4 Groundwater Sampling Using a Push Type Sampler

This portion of this procedure describes the use of a Geoprobe[®] Screen Point 15 Groundwater Sampler, or similar type equipment, for collecting groundwater samples at predetermined depths. These samples may be used to obtain physical, chemical, or radiological analyses.

A Geoprobe[®] Screen Point 15 Groundwater Sampler, or equivalent tool, is driven to a predetermined depth by a push type-sampling rig. The Screen Point 15 Groundwater Sampler is equipped with a 41-inch retractable screen and expendable drive point. It can then be partially or fully withdrawn (up to 41 inches) to expose a portion or the entire deployed well screen. After groundwater enters the exposed screen, a sample is collected using either the procedures in Subsection 4.8.1.1, Groundwater Sampling Using a Bailer, or in Section 4.8.1.2, Groundwater Sampling Using a Peristaltic Pump. Note that these samples are collected only for screening purposes because the sampling tool hole has not been completed as a well.

The method for obtaining QC samples using the push type-sampling tool is provided in Subsection 4.8.4.1 for groundwater sampling. Duplicate groundwater samples shall be collected only if there is enough water to collect two full suites of analytes without dewatering the annulus. If insufficient water is available for the collection of a planned QC sample, it shall be explained and documented in the field log book, and the project manager informed. If insufficient water is available for two full suites of analytes, it may be come necessary to prioritize the analyte list. The prioritization sequence should be described in the project-specific work plan.

4.8.2 Sample Filtering and Preservation

Samples for dissolved metals, Gross Alpha/Beta, ^{233/234}Uranium, ²³⁵Uranium, ²³⁸Uranium, ^{89/90}Strontium, ¹³⁷Cesium, ²²⁶Radium, ²²⁸Radium, and orthophosphate shall be filtered in the field at the well location during the sampling event through a disposable 0.45-micrometer membrane filter. If a peristaltic or bladder pump is used, a disposable filter may be attached directly to the sample delivery line so that the sample is filtered directly into the sample container as it exits the delivery line. Discharge pressure shall be gauged so it does not exceed 50 psi. Alternatively, sample water may be collected in a stainless steel container and filtered with a peristaltic pump. Before sample collection, 100 to 200 milliliters of sample water shall be passed through the filter in order to rinse the filter and filtration apparatus of possible contaminating substances.

Preservatives shall be added to the sample bottles prior to the introduction of the filtered sample water. The preservative shall be added in aliquots appropriate to the size of the bottle.

After sample collection has been completed, the pH of preserved samples shall be checked as follows.

- Pour a small amount of sample from the sample bottle directly onto approved pH paper. Use care so that the threaded neck of the bottle does not contact the pH paper. Do not, under any circumstances, insert the pH paper into the sample bottle.
- Check the pH paper against the supplied color chart. If the appropriate pH has not been achieved, add additional preservative to the sample in 5 ml aliquots and repeat the pH test after each addition.

4.8.3 QA/QC Samples

The frequency and types of field QA/QC samples collected during groundwater sampling are described in project-specific work plans or quality assurance plan documents. These documents detail the applicable criteria for collecting QA/QC samples.

4.8.3.1 Duplicates

Duplicate samples shall be collected only from wells that produce enough water to collect two full suites of analytes without dewatering. Wells that produce sufficient water shall be incorporated into the sampling program such that the required duplicate frequency can be maintained.

Wells scheduled for duplicate sample collection shall be sampled as described in Subsection 4.8 of this procedure, and in relevant sections of project-specific work plans and/or quality assurance documents. Field duplicates are collected following the same sampling procedures used to obtain the real samples. With the exception of VOCs, the typical procedure for a location is to collect the real and duplicate of each sample at the same time, in two equal portions, with each portion going to the laboratory in separate containers. This is accomplished by alternately filling two sample bottles one half at a time to minimize heterogeneity. Note that real and duplicate VOC samples shall be collected independently to reduce the possibility of volatilization of the sample.

When a well with a dedicated pump is being used for sample collection, all samples shall be collected in the normal order, with duplicate VOC samples being collected first. The remaining samples will be sampled as described above.

If a well is being used for matrix spike (MS) and matrix spike duplicate (MSD) samples, the duplicate shall be collected after collection of the MS and MSD.

All duplicate samples shall be given a sample number different from the original sample and the information recorded on the sample collection log and/or the field QC sample collection log.

4.8.3.2 Matrix Spike and Matrix Spike Duplicate

MS and MSD samples shall be collected only from wells that produce enough water to collect the required suites of analytes without dewatering. MS and MSD samples are not collected on a routine basis, but will be collected if so designated in a site-specific sampling plans, or if requested by the project manager.

MS and MSD samples shall be collected as follows.

- Purge the well as described in Subsection 4.6 of this procedure..
- After completion of purging, collect VOC samples. Collect the real sample followed by the MS and MSD. Collect these samples in immediate succession.
- Collect the remaining samples not requiring filtering. For each sample parameter, collect the original sample, MS, and MSD concurrently. Fill the original sample bottle one-third full followed by the MS and MSD sample bottles, which are also filled one-third full. Rotate each bottle in the sequence, filling in one-third full until all three bottles are full. For analytes not requiring an MSD, collect only the original sample and the MS.
- After the real sample, MS, and MSD (where appropriate) are collected for one parameter, repeat the process for the next parameter.
- Similarly, collect samples requiring filtering. When a bailer is used, fill a stainless steel bucket with sample water. As samples are collected and the reservoir of water in the bucket is depleted, add more water with discretion. When a pump is used, attach the filter directly to the discharge line. Fill

sample bottles as described above, partially filling the original sample, MS, and MSD in rotating sequence until each parameter bottle is full.

- Radiochemistry samples may have more than one bottle for each parameter group. In this case, include all required bottles in the rotating sequence.
- Field parameter measurements are not be required for MS and MSD samples.
- Retain the original sample number for MS and MSD samples. However, add a suffix of MS or MSD to the sample number to correspond with each QA/QC sample. Record all information on the field QC groundwater sample collection log.

4.8.3.3 Replicates and Splits

Replicate and split samples shall be collected in the same manner as described for the MS and MSD. Seek instruction from the project manager for replicates and splits exceeding three samples. Record all information will be recorded on the groundwater sample collection logs.

4.8.3.4 Field Equipment Rinses

Wells scheduled for equipment rinsate samples shall be sampled as described in Subsection 4.8 of this procedure, and field equipment rinses shall be collected as described in this Subsection and in relevant portions of project-specific QC documents and work plans. Field equipment rinses shall be collected in a manner designed to reflect sampling techniques. All equipment used during sampling will be fully decontaminated as described in Subsection 4.3, then rinsed with distilled or deionized water. The rinse water will then be collected in bottles identical to those used for the original sample, and assigned a separate sample number. Analytes requiring filtration will be filtered using a new filter and tubing as required for the real sample. All information will be recorded on groundwater sample collection logs.

4.8.3.4.1 Bailed Wells

After completion of sampling, all equipment shall be decontaminated. Prior to leaving the well location, the equipment rinse will then be collected as follows.

- Fill the bailer with distilled or deionized water by pouring the water into the top opening.
- Decant the rinse water to the VOC vials through the bottom valve just as was done during sample collection.
- For the remaining unfiltered samples, fill the bailer with distilled or deionized water each time additional rinsate is needed. Transfer the rinsate to sample bottles or to a stainless steel bucket and then to sample containers in the same manner used during collection.
- Collect filtered samples in an identical manner as the real samples. Fill the bailer with distilled or deionized water. Then transfer the rinse water to a stainless steel bucket. Filter the rinse water in the bucket through a new disposable filter.

- Preserve rinse samples in the same manner as the real samples.
- 4.8.3.4.2 Pumped Wells

Rinsate samples are not routinely collected from wells that are equipped with dedicated bladder pumps because the samples from these wells are collected directly from the pump discharge line. However, wells sampled using peristaltic pumps for sampling may be selected for rinsate sampling, with equipment used in sample collection (down hole tubing, filter tubing and the stainless steel bucket used for sample water collection, etc.) being decontaminated prior to rinsate sampling. The tubing at the pump head will be replaced, and a new filter used for filtered analytes. To collect the samples, distilled or deionized water will be poured into the decontaminated stainless steel bucket and pumped, using the decontaminated tubing, into the sample containers. The equipment used to collect the real VOC samples will also be decontaminated, rinsed, and used to collect the VOC rinse samples. All samples will be preserved at the same pH levels as the real samples.

4.8.3.5 Distilled Water Blanks

Distilled water sample blanks are not submitted on a routine basis, but will be made up if so designated in a site-specific sampling plan. Samples of the distilled or deionized water used for the final decontamination of equipment will be transferred directly to sample bottles to determine any baseline contamination the water may have introduced into the samples. Five-gallon bottles of the distilled or deionized water will be opened in a controlled area, such as the bottle storage room, and then poured directly into the appropriate sample bottle. A Teflon[®], glass, or stainless steel funnel may be used to help control flows into small mouth bottles. Blank samples will be preserved to the appropriate pH required for each analyte. All information will be recorded on groundwater sample collection logs.

4.9 Sample Handling and Control

Pre-cleaned sample containers will be obtained from a contract analytical sample container source. Preserving solution will be added to the bottles by a laboratory, the sample manager or qualified sampling personnel. The bottles will be labeled to indicate the preservative added.

The sampling containers, preservation requirements, and holding times for the various types of analyses are shown in Table 1. Groundwater samples will be properly labeled so that they can be easily identified. The sample numbering system will be assigned by project-specific sampling plan documents. A sample identification (ID) number will be assigned to each sample suite. The sample ID number will contain the following information as part of a nine to twelve character, alpha-numeric code:

Character(s)	Description	Code
1 and 2	Project ID	GW
3 through 7	Sample Number	00001 to 99999
8 and 9	Subcontractor ID	Alpha (e.g. TE = Tierra Environmental Consultants)
10, 11, and 12	QA/QC	MS for matrix spike, MSD for matrix spike duplicate

In addition to a sample number, each well sampled will be assigned a current Record Identification Number (RIN), an event number (specific to the RIN), and bottle numbers that are specific to the RIN and event number.

5.0 Records

All field activities shall be recorded on a Field Activity Daily Log or Groundwater Sample Collection Log. Additional logs may be required to record QC samples and for recording well status. Refer to specific project, site, or facility work plans for further information. Summary information of the day's activities or other pertinent information should always be recorded on the field forms. Under some circumstances, the project manager may assign a bound field logbook to the field personnel that will remain in their custody during all sampling activities. The cover of each logbook shall contain the following information at a minimum:

- Name of the organization to which the book is assigned
- Book number
- Project name
- Start and end dates

Logbook pages shall be sequentially numbered and marked with the book number before any data are recorded. All data and information pertinent to field sampling shall be recorded in the logbook or on the field forms that identify all required data entries. Enough detail must be included in the documentation to reconstruct the sampling event. Field form entries shall include the following minimum information:

- Date and time
- Names of field personnel
- Names of all visitors
- Location of field activities
- Description of sampling sites including weather conditions
- All field observations and comments
- Field parameters
- Sample identification information
- References to all prepared field activity forms and chain-of-custody records

Field logbooks, when required on specific projects, shall normally be kept only by the field sampling team leaders and the site supervisor and shall typically be used only to summarize field activities and to document project information not required by the procedure field forms.

Permanent ink shall be used for all entries in the logbooks and on the field forms. Mistakes shall be crossed out with a single line, initialed, and dated. Unused pages or partial pages shall be voided by drawing a line through the blank sections and initialing and dating the mark. Any deviation from this procedure shall require documentation in the site supervisor's logbook.

The field activity daily log narrative should create a chronological record of the sampling team's activities, including the time and location of each activity. Descriptions of problems encountered, personnel contacted, deviations from the procedure, and visitors on site shall also be included. The weather conditions, date, signature of the person responsible for entries, and the number of field activity daily log sheets used to record media team activities for a given day shall also be included.

The Groundwater Levels Measurement/Calculations Form and the Chain of Custody Record (see *Containing, Preserving, Handling, and Shipping Soil and Water Samples*) shall also be completed for each site. All blank fields on the forms must be completed or voided.

6.0 References

- Environmental Protection Agency, 1982, Test Methods for Evaluating Solid Waste, SW-846, Volume II. Field Methods, 2nd edition.
- Environmental Protection Agency, 1986a, Engineering Support Branch Standard Operating Procedures and Quality Assurance Manual, EPA Region IV Environmental Service Division.
- Environmental Protection Agency, September 1986b, RCRA Ground Water Monitoring Technical Enforcement Guidance Document, OSWER-9950.1.
- Environmental Protection Agency, 1987a, A Compendium of Superfund Field Operations Methods, EPA/540/P-87/001. 1987.
- Environmental Protection Agency, 1987b, Data Quality Objectives for Remedial Activities, Development Process, EPA/540/G-87/003.
- Environmental Protection Agency, December 1988, User's Guide to the Contract Laboratory Program.

APPENDIX A

STANDARD GROUNDWATER FORMS

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Appendix C Surface Water Sampling Procedures

	Sample Location: CSMRI-1
& tallor	Date: 3/2/11
<u> + U + V + V / / / / / / / / / / / / / / / /</u>	Project Name: Colorado School of Mines Project Number: 4349-430
	Sample Type: GW SW EB Dup
and the second	Sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations				
Measured TD = 25.02 (ft)	(+.28)	TD =	25.40	(ft)
Initial Water Volume = 2.74	(gal)	Depth to Water =	8.22	(ft)
3X Water Volume = 8,22	(gal)	Water Column =	17.18	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°C,)°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
1315	1.37	8.42	6.67	789	0.00	146	169.0	bron
1318	2.74	8.71	6.89	734	0.00	146	243.0	
1321	4.11	7.99	6.94	785	0.00	145	264.0	
1324	5-48	7.68	6.97	743	0.00	146	321.0	
1327	6.85	7.38	6.98	754	0.00	147	303.0	
1330	8,22	7.30	6.98	752	6.00	148	349.0	X

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Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	1335	ALS
Cations	500 mL Plastic	HNO ₃		1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄	-		ALS
NO ₂ , NO ₃	1 L Plastic	none		t.	TA
Total Dissolved Solids	1 L Plastic	none	+		TA

Comments: DO probe was broken

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Stoller	Sample Location: CSMRT-1B Date: O2-28-2011 3/2/11 3/2/11, 3/3/11 Project Name: Colorado School of Mines Project Number: 4349-430 Sample Type: GW SW EB Dup Sampler: Nick Malczyk, Pete Dalen
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Purge Volume Calculations			
Measured TD = .73, 4/1 (ft)	(+.28)	TD = 23.69	(ft)
Initial Water Volume = 0,47	(gal)	Depth to Water = 20.78	(ft)
3X Water Volume = 1.41	(gal)	Water Column = 2,9/	(ft)

Time	Volume	Temp	pH	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	° () ° F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
040	0.47	10.99	6.75	1380	0.00	110-65-68	laat	Brown
~	0.94							
	1.91							
							-	
								+JP

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/3/11	1000	ALS
Cations	500 mL Plastic	HNO3	. 1	1	ALS
Anions	500 mL Plastic	none	·		ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none		6	TA
Total Dissolved Solids	1 L Plastic	none	*		TA

Comments: DO probe not working

	Sample Location: CSM R.J-2
Stoller	Date: 2/28/11, 3/3/11 Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: (SW) SW EB Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations	1.20			
Measured TD = 95.12 (ft)	(+.28)	TD =	95.40	(ft)
Initial Water Volume = 6.14	(gal)	Depth to Water =	56.98	(ft)
3X Water Volume = 18, 44	(gal)	Water Column =	38.42	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(CO°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
345	3.07	13.59	7.32	568	0.00	57	10007	brown
1359	6.14	12.59	7.23	571	0.00	97	1000+	
1406	9.21	1253	7.18	616	0.00	111	10001	
1415	12.28	11.59	7.40	643	0.00	122	lovot	4
-								h

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/3/11	0930	ALS
Cations	500 mL Plastic	HNO ₃	. 1		ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none		*	TA
Total Dissolved Solids	1 L Plastic	none	*	+	TA
					_
					m

Comments: DO probe was not wanting

(+ illor	Sample Location: CSMRJ-4 Date: 3/1/11 Project Name: Colorado School of Mines
\$ <u>+UUUU</u>	Project Number: 4349-430
	Sample Type: (W) SW EB Dup
	Sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations				
Measured TD = 17.33 (ft)	(+.28)	TD =	17.61	(ft)
Initial Water Volume = 1.64	(gal)	Depth to Water =	7.32	(ft)
3X Water Volume = 4.38	(gal)	Water Column =	10.29	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°G)°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
248	0,92	8.00	7.05	1840	0.00	127	44.8	born
250	1.64	6.89	7.04	1670	0.00	124	16.9	
1252	2.46	6.66	6.98	1640	0.00	130	5-0	
1254	3.28	6.64	6.94	1630	0.00	134	19.9	
1256	3.56	6.46	6.93	1620	0.00	136	5.2	
1258	4.38	6.36	6.96	1600	0.00	137	17.1	t

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/1/11	1300	ALS
Cations	500 mL Plastic	HNO3			ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none	e		TA
Total Dissolved Solids	1 L Plastic	none			TA
Ferrous Iron	1 L Plastic	none			TA
Ferric Iron	500 mL Plastic	HNO ₃			TA
Sulfide	250 mL Plastic	ZnAc	¥_	V	TA
					- n

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Comments: DO probe was not warking

	Sample Location: CSMRJ-5
& tollow	Date: 3/1/11
$\otimes UUUUUU$	Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
2012-11-201-	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations				
Measured TD = 10,99 (ft)	(+.28)	TD =	11.27	(ft)
Initial Water Volume = 0,71	(gal)	Depth to Water =	6.81	(ft)
3X Water Volume = 2.13	(gal)	Water Column =	4.46	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°(C,)F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
1305	0.71	8.71	7.03	1560	0,00	129	78.8	brown
1308	1.42	7.73	7.01	1530	0.00	1.30	477.0	
1311	2.13	7.57	6.99	1510	6.00	131	442.0	*
							State and state and state and state	Aren

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/1/11	1315	ALS
Cations	500 mL Plastic	HNO ₃	1	1	ALS
Anions	500 mL Plastic	none			ALŚ
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none			TA
Total Dissolved Solids	1 L Plastic	none			TA
Ferrous Iron	1 L Plastic	none			TA
Ferric Iron	500 mL Plastic	HNO ₃			ТА
Sulfide	250 mL Plastic	ZnAc	¥	\forall	TA
					form

4

Comments: DO pruse was not warting

1 in

· ·	Sample Location: CSMRI-6C
& taller	Date: $0,2-2,8-201,3/2/11$ Project Name: Colorado School of Mines
<u> + U + U + U + U + U + U + U + U + U + </u>	Project Number: 4349-430
	Sample Type: GW SW EB Dup
19 19 19 19 19 19 19 19 19 19 19 19 19 1	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations			
Measured TD = 39.25 (ft)	(+.28)	TD = 60.03	(ft)
Initial Water Volume = 1.1088	(gal)	Depth to Water = $33./0$	(ft)
3X Water Volume = 3.33	(gal)	Water Column = $6,95$	(ft)

Time	Volume	Temp	рН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(C °F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
1015	1.11	12.06	6.99	1330	O.OD	59	1000 +	Brown
021	7.77	11.60	7.No	1180	0.05	50	10254	Brich
O2ie	3.33	11.83	7.18	1160	0.67	69	1000+	Brown
·				.,				
							+	- PÚ

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	1045	ALS
Cations	500 mL Plastic	HNO ₃	_,]	1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none	ŀ		TA
Total Dissolved Solids	1 L Plastic	none	+	+	TA
					-

Comments: NO probe not working

	Sample Location: CSMRT - 7C
	Date: 02+28=2011, 3/1/11, 3/2/11
N 6066€7	Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	Sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations			
Measured TD = 34.15 (ft)	(+.28)	TD = 24,43	(ft)
Initial Water Volume = 1,09	(gal)	Depth to Water = 17.62	(ft)
3X Water Volume = 3,27	(gal)	Water Column = 6,8/	(ft)

Time	Volume	Temp	рН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°C, >F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
052	1.09	11.06	7.23	1110	0.00	87	1600+	Bran
1057	2,18	11.25	7.23	1070	0.00	64	1000+	Brain
101	3.27	11.21	7.32	1500	0.00	16	1250'	Brown
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						<u> </u>		
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Analysis	Container	Preservative	Date	Time	Lab	
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	0900	ALS	
Cations	500 mL Plastic	HNO ₃	1	1	ALS	
Anions	500 mL Plastic	none			ALS	
DOC	125 mL Amber	H ₂ SO ₄			ALS	
NO ₂ , NO ₃	1 L Plastic	none		t	TA	
Total Dissolved Solids	1 L Plastic	none	\downarrow	4	TA	
~						

Comments: DC probe not working

Ctoller	Sample Location: <u>CSMRT - 813</u> Date: 02-28-2011, 3//// Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	Sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculation	IS				
Measured TD = /0//	(ft)	(+.28)	TD =	10.39	(ft)
Initial Water Volume = (), 4(2	(gal)	Depth to Water =	7.86	(ft)
3X Water Volume = 1,20)	(gal)	Water Column =	2.53	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(C) °F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
1:14	0.40	6.90	7.19	1470	000	PE 1	1000 t	Brown
1:17	0.80	5.99	7.10	1420	0.00	-5	1000+	Brown
1120	1=20	6.06	7,08	1970	0.00	-2	1000t	Brown
5-			V					
					<u> </u>			6.1

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/1/11	1405	ALS
Cations	500 mL Plastic	HNO ₃		1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none	e		TA
Total Dissolved Solids	1 L Plastic	none			TA
Ferrous Iron	1 L Plastic	none			TA
Ferric Iron	500 mL.Plastic	HNO ₃			TA
Sulfide	250 mL Plastic	ZnAc	4	*	TA

Do probe not working

Mi - II - a	Sample Location: CCMRI-9
Stoller	Project Name: Colorado School of Mines Project Number: 4349-430
	Sample Type: (GW) SW EB Dup Sampler: Nick Malczyk, Pete Dalen

Purge Volume	Calculations					
Measured TD =	33,10	(ft)	(+.28)	TD =	33,38	(ft)
Initial Water Volum	e= 1,20		(gal)	Depth to Water =	25.86	(ft)
3X Water Volume =	2.60		(gal)	Water Column =	7.52	(ft)

Time	Volume	Temp	рН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(*C,)F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
0932	1,20	12.72	6.87	1650	0.00	153	1000+	bon
0935	2.40	12.49	6.95	1620	0.00	152	1000t	1
0938	3.60	12.61	6.96	1620	0.00	152	10001	4
								MM

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/1/n	0945	ALS
Cations	500 mL Plastic	HNO ₃	, ,	1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄		1	ALS
NO ₂ , NO ₃	1 L Plastic	none		6	TA
Total Dissolved Solids	1 L Plastic	none	+	+ +	TA

comments: DO proLe HZS not working

	Sample Location: CEMRIT-10
C+nllnm	Date: 3/1/11
	Project Name: Colorado School of Mines
<u></u>	Project Number: 4349-430
	Sample Type: W SW EB Dup
and the second se	Sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations			_	
Measured TD = 27.87 (ft)	(+.28)	TD =	28,15	(ft)
Initial Water Volume = 0,63	(gal)	Depth to Water =	24.24	(ft)
3X Water Volume = 1.89	(gal)	Water Column =	3.91	(ft)

Time	Volume	Temp	рH	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(D°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	- 1
0913	0.63	11.43	6.34	1520	0.00	175	1000+	brean
0915	1.26	11.41	6.64	1500	0.00	170	766	1
0917	1.89	11.39	6.84	1520	0.00	164	495	1
								m

Container	Preservative	Date	Time	. Lab
1 gallon Cube	HNO ₃	3/1/11	0920	ALS
500 mL Plastic	HNO3	1	1	ALS
500 mL Plastic	none			ALS
125 mL Amber	H ₂ SO ₄			ALS
1 L Plastic	none			TA
1 L Plastic	none	V	+	TA
			+	
	1 gallon Cube 500 mL Plastic 500 mL Plastic 125 mL Amber 1 L Plastic	1 gallon Cube HNO_3 500 mL Plastic HNO_3 500 mL Plasticnone125 mL Amber H_2SO_4 1 L Plasticnone	1 gallon Cube HNO3 3/1/11 500 mL Plastic HNO3 1 500 mL Plastic none 1 125 mL Amber H2SO4 1 1 L Plastic none 1	1 gallon Cube HNO3 3/1/11 092-0 500 mL Plastic HNO3 1 500 mL Plastic none 1 125 mL Amber H2SO4 1 1 L Plastic none 4

comments: DO probe was not werking

	Sample Location: (SMRI-11 B
1 + n / nm	Date: 02-28-2011, 3/2/11
	Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations			
Measured TD = 37.37 (ft)	(+.28)	TD = 37.65	(ft)
Initial Water Volume = 1,30	(gal)	Depth to Water = 29.444	(ft)
3X Water Volume = 3,90	(gal)	Water Column = S. 21	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(℃)°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
303	1.30	13.24	7.05	1250	0.00	22	1000 *	Brown
307	2.60	1271	7.00	1230	0.00	22	10as+	Brown
30	3.90	12.51	6.97	1230	0.00	16	1000+	Brown
5								
								$\vdash - ll$

Container	Preservative	Date	Time	Lab
1 gallon Cube	HNO ₃	3/2/11	1015	ALS
500 mL Plastic	HNO ₃			ALS
500 mL Plastic	none			ALS
125 mL Amber	H ₂ SO ₄			ALS
1 L Plastic	none		6.	TA
1 L Plastic	none	4	4	TA
	_			
	1 gallon Cube 500 mL Plastic 500 mL Plastic 125 mL Amber 1 L Plastíc	1 gallon CubeHNO3500 mL PlasticHNO3500 mL Plasticnone125 mL AmberH2SO41 L Plasticnone	1 gallon CubeHNO33/2/11500 mL PlasticHNO31500 mL Plasticnone1125 mL AmberH2SO411 L Plasticnone1	1 gallon Cube HNO3 3/2/11 1015 500 mL Plastic HNO3 1015 500 mL Plastic none 125 mL Amber H2SO4 1 L Plastic none 4

Comments: DO probe inof working

Stoller	Sample Location: <u>CSMRJ-12</u> Date: <u>3/,////</u> Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: 🐨 SW EB Dup
	Sampler: Nick Malczyk, Pete Dalen

Purge Volume (Calculations					
Measured TD =	10.11	(ft)	(+.28)	TD =	10.39	(ft)
Initial Water Volume	= 0.95	.	(gal)	Depth to Water =	4.46	(ft)
3X Water Volume =	2.85	-	(gal)	Water Column =	5.93	(ft)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(C, F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
1105	0.95	6.14	7.13	1580	0.00	-29	1000+	bran
107	1.90	5.56	7.12	1560	0.00	-27	lovot	
189	2.85	5.81	7.08	1530	0.00	-3z	1000+	+

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO₃	3/1/11	1430	ALS
Cations	500 mL Plastic	HNO ₃	, , ,	1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none	e		TA
Total Dissolved Solids	1 L Plastic	none			TA
Ferrous Iron	1 L Plastic	none			TA
Ferric Iron	500 mL.Plastic	HNO ₃			TA
Sulfide	250 mL Plastic	ZnAc	¥	4	TA
					non

comments: 120 probe was not working

Ctoller	Sample Location: (SMAT ~ 1'3 Date: 52-28-2011, <u>z/2/11</u> Project Name: Colorado School of Mines
	Project Number: 4349-430 Sample Type: W SW EB Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations				
Measured TD = 10.10	(ft) (+.28)	TD =	10.38	(ft)
Initial Water Volume = 0,95	(gal)	Depth to Water =	4.41	(ft)
3X Water Volume = 2, 8,5	(gal)	Water Column ≈	5,97	(ft)

Time	Volume	Temp	pH	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°C,°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
224	0.95	6.81	7.17	1300	B-GG V	5	1000 +	Brown
227	1.90	PU 55.96	7.18	1/60	0.00	-27	10000	Brown
1237	2.83	6.51	7.14	1220	1.00	-23	1000 *	Brown
								L PD

Cube Plastic Plastic Amber ic	HNO ₃ HNO ₃ none H ₂ SO ₄	3/2/11	093	ALS ALS ALS ALS ALS ALS
Plastic Amber	none H ₂ SO ₄			ALS ALS
Amber	H ₂ SO ₄			
				ALS
ic .				
	none	E		TA
ic	none			TA
tic	none			TA
Plastic	HNO3			TA
Plastic	ZnAc	+	-	TA
F	ic Plastic	ic none Plastic HNO ₃	ic none None None None None None None None	ic none None Nastic HNO ₃

comments: DO prese not working

Ctoller	Sample Location: <u>CSMRJ~14</u> Date: <u>D2/28-2011</u> <u>3/1/11</u> Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculations			
Measured TD = 5695 (ft)	(+.28)	TD= 5223	(ft)
Initial Water Volume = 8,62	(gal)	Depth to Water = 3,37	(ft)
3X Water Volume = 25.56	(gal)	Water Column = 55.86	(ft)

Time	Volume	Temp	рН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	€C)°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
11:46	431	10.56	8.16	4300	0,00	-39	1000 t	Greyipba
2:05	8.62	14.58	8.18	4080	000	-28	1000 "	Gizy
2:14	12.93	1.83	8,12	4//0	0-00	-7	iDa0+	Grey
~~~~	17.24			•				/
	21.55							
	25.56							L PD

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/1/11	1355	ALS
Cations	500 mL Plastic	HNO ₃	1	1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none	Y	Ł	TA
Total Dissolved Solids	1 L Plastic	none	+	T	TA
<u> </u>					

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Comments: DO probe net monthly

F F	Sample Location: SLJ-/
Stoller	Date: 3/2/11 Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculat	IUIIS			
Measured TD =	(ft)	(+.28)	TDe	(ft)
Initial Water Volume =	NEM	(gal)	Depth to Water =	Nom (ft)
3X Water Volume =		(sal)	Water Column =	鱼

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°C,)F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
348	NA	2.13	7.34	465	0.00	120	21.4	clear
-							0.000	
				Contraction of the state of the				
							· ·	ma

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	1350	ALS
Cations	500 mL Plastic	HNO ₃	1		ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄	1		ALS
NO ₂ , NO ₃	1 L Plastic	none			TA
Total Dissolved Solids	1 L Plastic	none	The second secon	Ŧ	TA
					10

Comments: DO probe was broken

	Sample Location: SW-2
& + allam	Date: 03-02-2011
	Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculatio	ons ·			
Measured TD =	(ft)	(+.28)	TD.=	(ft)
Initial Water Volume =		PD (gal)	Depth to Water =	PD (ft)
3X Water Volume =		(gal)	Water Column =	tet

Time	Volume	Temp	pH	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°C)°F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
117	NA	1.01	6.27	483	0.00	145	19.5	Clear
~								
					1			
								PD

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	1117	ALS
Cations	500 mL Plastic	HNO ₃		1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none			ТА
Total Dissolved Solids	1 L Plastic	none	+	-	TA
					- 07

Comments: DS prob not working

	Sample Location: SU-3
& +n/low	Date: 3/2/11
	Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW EB Dup
	Sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculati	ions			
Measured TD	(ft)	(+.28)	TD =	(ft)
Initial Water Volume =	N	en (gal)	Depth to Water =	Nr.M (ft)
3X Water Volume =		(gal)	Water Column =	Tit

Time	Volume	Temp	рН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(FC,)F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
140	NA	0.83	6.61	471	0.00	129	19.3	clees
~								-
								Arn.

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	1145	ALS
Cations	500 mL Plastic	HNO3	1	1	ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none		6.	TA
Total Dissolved Solids	1 L Plastic	none	-	+	TA
1					_
a c					

comments: DO probe was broken

	Sample Location: Earlyment Blank
	Date: 03 - 2-2011
$\sim$ TOLLEY	Project Name: Colorado School of Mines
	Project Number: 4349-430
	Sample Type: GW SW (EB) Dup
	sampler: Nick Malczyk, Pete Dalen

Purge Volume Calculation	5			
Measured TD =	(ft)	(+.28)	TD =	(ft)
Initial Water Volume =		(gal)	Depth to Water =	P) (ft)
3X Water Volume =		(gal)	Water Column =	(#)

Time	Volume	Temp	pН	Conductivity	DO	ORP	Turbidity	Appearance
	(gal)	(°C, °F)	(SU)	(uS/cm)	(mg/L)	(mV)	(NTU)	
-								
			_					
			•					
								PP

Analysis	Container	Preservative	Date	Time	Lab
Ra-226, -228, Diss. Uranium	1 gallon Cube	HNO ₃	3/2/11	0955	ALS
Cations	500 mL Plastic	HNO ₃			ALS
Anions	500 mL Plastic	none			ALS
DOC	125 mL Amber	H ₂ SO ₄			ALS
NO ₂ , NO ₃	1 L Plastic	none		ŧ	TA
Total Dissolved Solids	1 L Plastic	none	$\downarrow$	4	TA
~~~~					

Comments:

APPA BD

Appendix D Data Validation Reports

Surface Water Sampling

1.0 Scope and Objective

1.1 Scope

This procedure provides instructions and establishes requirements for the collection and documentation of surface water samples by Stoller personnel. This procedure applies to the collection of surface water samples from streams, rivers, ponds, lakes, seeps, impoundments, and other surface sources.

1.2 Objective

The objective of this procedure is to establish a uniform method for the collection of surface water samples that provides representative samples in a safe and responsible manner.

2.0 Definitions

Composite Sample – A sample that is comprised of roughly equal amounts of water collected from a set of sample locations known as a sample group.

Grab Sample – A single sample collected at one sampling point over a short period of time. Grab sample results are representative of the sample location at the time of sample collection. Also called a catch sample.

Peristaltic Pump – A self-priming, low volume pump consisting of a rotor and ball bearing rollers. Tubing placed around the rotors is squeezed by the rotors as they revolve. The squeezing produces a wavelike contractual movement which causes water to be drawn through the tubing. The peristaltic pump is limited to sampling at depths of less than 25 feet.

3.0 Responsibilities and Qualifications

3.1 Project Manager

The Project Manager is responsible for ensuring that surface water samples are properly and safely collected. This will be accomplished through staff training and by maintaining quality control (QC). At a minimum, project management shall:

- 3.1.1 Verify that personnel have reviewed, and are familiar with, site-specific work plans which address surface water sampling, this procedure, and any associated procedures.
- 3.1.2 Ensure that hazards are identified and analyzed with respect to collecting surface water samples, and develop and implement controls to minimize hazards.
- 3.1.3 Provide personnel with training in the operation of surface water sampling equipment and the requirements of this procedure.
- 3.1.4 Periodically review field generated documentation associated with surface water sampling to ensure compliance with project requirements and implement corrective action if necessary.
- 3.1.5 Receive feedback from field sampling personnel in order to continually improve surface water sampling process.

3.2 Site Supervisor

The Site Supervisor is responsible for directing and overseeing all field activities, including sampling, to ensure that site-specific plan requirements are met in a safe and efficient manner within the established safety envelope.

Stoller

3.3 Field Sampling Personnel

Field sampling personnel are responsible for the proper sample collection and documentation of the sampling event in accordance with this procedure. At a minimum, field sampling personnel have the responsibility to:

- 3.3.1 Familiarize themselves with site-specific work plans, surface water sampling procedures, potential hazards, and health and safety plan.
- 3.3.2 Implement the controls to minimize hazards.
- 3.3.3 Be familiar with sampling equipment and its proper use.
- 3.3.4 Properly complete field documentation.
- 3.3.5 Provide feedback to project manager in order to improve sampling process.

4.0 Equipment/Materials and Calibration

4.1 Equipment/Materials

A number of devices are available for the collection of surface water samples. These devices are constructed of a number of materials including, but not limited to: stainless steel, glass, Teflon[®], Tygon[®]. The sampling and analytical requirements, as well as site characteristics, must be taken into account when determining the proper surface water sampling equipment to use. The site-specific work plans should identify the specific equipment to be used, and methods for safely using equipment.

4.2 Calibration

Equipment shall be calibrated in accordance with manufacturer's recommendations and calibration documentation shall be maintained in project files.

5.0 Method

5.1 Field Preparation

Field preparation requires the organization of sample containers, sample labels, and documentation in an orderly, systematic manner to promote consistency and traceability of all data.

- 5.1.1 General sampling areas will be predetermined to ensure coverage of the various impact scenarios and should be described in project-specific work plans. The location of each sampling point shall be surveyed or mapped and staked as described in Section 5.1.6 prior to sampling.
- 5.1.2 In flowing water, surface water sampling shall be conducted from downstream locations first, then proceed to upstream locations to avoid potential cross contamination from disturbing the substrate.
- 5.1.3 Prior to sampling and between sampling locations, sampling equipment shall be decontaminated.
- 5.1.4 Appropriate personal protective equipment shall be used, as specified in the project-specific health and safety plan.

5.1.5 All pertinent information (date, site name, identification number, and location) shall be recorded on a Field Activity Daily Log (FADL) and a Sample Collection Log, as appropriate. Field conditions, unusual circumstances, and weather conditions shall be noted.

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- 5.1.6 Due to the nature of sampling an aqueous environment, additional steps are required to verify and mark sample locations. Depending on the project needs, it may be useful to use a Global Positioning System (GPS) to verify and mark the sample locations. Refer to *Field Mapping with a Global Positioning System* for details. The following steps shall be followed by the sampler in addition to the field preparation requirements described in Section 5.1.1.
 - 5.1.6.1 Place a marker (stake) on the shore approximately perpendicular to the sampling location and mark the sample number on the stake.
 - 5.1.6.2 If the sample location is accessible by foot, use a measuring tape to measure the distance between the marked point and the sample location station. Record the compass bearing from the sample location to the shore marker.
 - 5.1.6.3 If the sample location is accessible only by boat, use a rangefinder to estimate the distance to the shore marker to obtain the most accurate measurement. Record the compass bearing from the sample location to the shore marker. It is recommended that the boat's position on the water be stabilized to prevent drifting.
 - 5.1.6.4 Determine and record the distance and direction of each shore marker from a reference point shown on the topographic map and mark all points on a map or use a GPS, if available.
- 5.1.7 Quality Control samples, including field and source blanks, shall be collected in accordance with the project-specific work plan.

5.2 Surface Water Sample Collection Using a Transfer Container

The device most commonly used to collect grab surface water samples is a transfer container (beaker, flask, etc.) made of inert material such as glass, stainless steel or Teflon[®]. When sampling with a transfer container, the procedure is as follows:

- 5.2.1 Survey and clearly map sampling points as described in Section 5.1.6 prior to sampling. The sample should be collected as close to the mapped location as possible. If the collection point must be moved, the new location must be approved and documented.
- 5.2.2 Dip the transfer container into the surface water. Always use a clean, properly decontaminated transfer container at each sample location.
- 5.2.3 Filter the sample if required.
- 5.2.4 Fill the sample bottle, allowing the sample stream to flow gently down the inside of the bottle with minimal turbulence.
- 5.2.5 Cap the bottle and handle the sample according to the procedures outlined in Project *Sample Shipping*.
- 5.2.6 Label the sample and document the sampling event.

5.3 Surface Water Sample Collection Using a Peristaltic Pump

A device used to collect composite surface water samples is a peristaltic pump. Samples to be analyzed for volatile organic analysis cannot be composited. When sampling with a peristaltic pump, the procedure is as follows:

5.3.1 Survey and clearly map sampling points as described in Section 5.1.6 prior to sampling. The sample should be collected as close to the mapped location as possible. If a collection point must be moved, the new location must be approved and documented.

- 5.3.2 Attach the appropriate tubing to the peristaltic pump. Always use new tubing at each sample location. Do not try to decontaminate and reuse tubing.
- 5.3.3 If filtering is required, attach the filtering device to the discharge end of the tubing.
- 5.3.4 Lower the intake end of the tubing into the water and begin pumping. If the pump is computerized, program the pump to collect the sample at the desired intervals and flow rate. If the pump is not programmable, record the discharge rate (compute discharge rate by dividing an amount of water collected by the time it took to collect it). Collect the sample at the desired interval.
- 5.3.5 Fill the sample bottle, allowing the sample stream to flow gently down the inside of the bottle with minimal turbulence. The programmable pump will perform this automatically.
- 5.3.6 Cap the bottle and handle the sample according to the procedures outlined in Project *Sample and Shipping*.
- 5.3.7 Label the sample and document the sampling event.

6.0 Required Inspection/Acceptance Criteria

None.

7.0 Records

The following records generated as a result of implementation of this procedure shall be maintained in a safe manner and submitted to project central files for storage and disposition.

Field Activity Daily Log

Sample Collection Log

Chain of Custody

8.0 References

8.1 Others

U.S. Environmental Protection Agency. 1987. EPA Compendium of Superfund Field Operations Methods, EPA 540/P-87/001a, OSWER 9355.0-14. Washington, DC.

U.S. Environmental Protection Agency. 1988. EPA Guidance for Conducting Remedial Investigation and Feasibility Studies under CERCLA, Interim Final OSWER Directive 9355.3-01. Washington, DC.

American Public Health Association, American Water Works Association, Water Pollution Control Federation. 1985. *Standard Methods for the Examination of Water and Wastewater*, 16th Edition, American Public Health Association, Washington, DC.

DATA VALIDATION REPORT

To:Robert HillFrom:John GarrettDate:March 13, 2011Project/Site:Colorado School of MinesProject No.:4349-430SDG No.:1103074 Radium-226

This report presents the radiological data validation for the data obtained during the field activities for the above referenced work assignment. The purpose of this review is to provide a technical evaluation of the radiological results that were obtained by ALS Laboratory Group Procedure SOP 783R9 for Radium-226 by Radon Emanation Counting for SDG 1103074 from ALS Laboratory Group (Fort Collins, CO). This report consists of eighteen water samples for the Colorado School of Mines/4349-430 project collected on March 1, 2011, March 2, 2011, and March 3, 2011 and submitted to ALS Laboratory Group on March 4, 2011. The samples were analyzed for Radium-226 by Radon Emanation Counting Method 903.1 on March 31, 2011 and April 1, 2011. The analysis was conducted by ALS Laboratory Group. The field sample numbers and corresponding laboratory numbers are presented below:

Client Sample Number	Laboratory Sample Number	Matrix	Collection Date
CSMRI-10	1103074-1	Water	March 1, 2011
CSMRI-9	1103074-2	Water	March 1, 2011
CSMRI-4	1103074-3	Water	March 1, 2011
CSMRI-5	1103074-4	Water	March 1, 2011
CSMRI-14	1103074-5	Water	March 1, 2011
CSMRI-8B	1103074-6	Water	March 1, 2011
CSMRI-12	1103074-7	Water	March 1, 2011
CSMRI-7C	1103074-8	Water	March 1, 2011
CSMRI-13	1103074-9	Water	March 1, 2011
Equipment Blank	1103074-10	Water	March 1, 2011
CSMRI-11B	1103074-11	Water	March 1, 2011
CSMRI-6C	1103074-12	Water	March 1, 2011
SW-2	1103074-13	Water	March 1, 2011
SW-3	1103074-14	Water	March 1, 2011
CSMRI-1	1103074-15	Water	March 1, 2011
SW-1	1103074-16	Water	March 1, 2011
CSMRI-2	1103074-17	Water	March 1, 2011
CSMRI-1B	1103074-18	Water	March 1, 2011

Data validation was conducted in accordance with the Analytical Services Statement of Work for the following modules: Gas Proportional Counting Module RC04-v2, October 1, 2002 and U.S. DOE Quality Systems for Analytical Services Revision 2.6 (QSAS).

The radiological data were evaluated based on the following parameters:

- * Data Completeness
- * Holding Times and Preservation
- * Instrument Initial Calibrations
- * Instrument Performance Checks
- * Preparation Blanks
- * Duplicate Sample Results
- * Laboratory Control Samples (LCS) Results
- * Laboratory Control Samples Duplicate (LCSD) Results
- * Compound Quantitation and Reporting Limits (full validation only)

Data Completeness

The data package was complete as per ALS Laboratory Group Procedure SOP 783R9 for Radium-226 by Radon Emanation Counting Method 903.1.

Holding Times and Preservation

Analytical holding times were evaluated and all criteria were met. However, holding time requirements are not applicable to radiochemistry analyses unless the isotopes of interest have short half-lives. The holding times for Radium-226 were met. No action was necessary.

Calibrations

The instruments were calibrated at the required frequency.

Initial Calibration

All instruments were calibrated properly using NIST traceable SRM.

Instrument Performance Checks

All isotopes were within criteria.

Preparation Blanks

Preparation/Method Blanks were performed at the required frequency. Radium-226 was not detected in the Method Blank above the MDC or the RDL.

Duplicate Sample Analysis

Due to volume constraints the laboratory prepared a Laboratory Control Sample Duplicate (LCSD in lieu of a client sample duplicate. Duplicate analyses were performed at the required frequency. All isotopic activities for Radium-226 original LCS and duplicate LCSD analysis were within the limits of the statistical test for equivalency.

Matrix Spike/Matrix Spike Duplicates

Matrix spike/matrix spike duplicates were performed for the samples in this SDG and reporting criteria were met.

Laboratory Control Samples

LCS analyses were performed at the required frequency. All recoveries were within 75-125% limits. No calculation errors or transcription errors were found.

Analyte Quantitation and Reporting Limits

Analyte quantitation was evaluated for all samples. No calculation or transcription errors were found. The results and reporting limits were correctly reported.

Overall Comments

Overall, the data are of good quality and are usable as reported by the laboratory without qualification.

The laboratory reported that the ICP-AES measurement of the added barium carrier prior to chemical separation. Several samples showed barium concentrations less than zero. The laboratory manually adjusted the values to 0.0 in order to avoid a low bias. All QC criteria were within control limits and no action was necessary. The data are not affected.

DATA QUALIFIER DEFINITIONS

For the purpose of Data Validation, the following code letters and associated definitions are provided for use by the data validator to summarize the data quality.

- R Reported value is "rejected." Resampling or reanalysis may be necessary to verify the presence or absence of the compound.
- J The associated numerical value is an estimated quantity because the Quality Control criteria were not met.
- U J The reported quantitation limit is estimated because Quality Control criteria were not met. Element or compound was not detected.
- U The material was analyzed for, but was not detected above the level of the associated value. The associated value is either the sample quantitation limit or the sample detection limit.
- NR Result was not used from a particular sample analysis. This typically occurs when more than one result for an element is reported due to dilutions and reanalysis.

DATA VALIDATION REPORT

To:	Robert Hill
From:	John Garrett
Date:	April 22, 2011
Project/Site:	Colorado School of Mines
Project No.:	4349-430
SDG No.:	1103074 Radium-228

This report presents the radiological data validation for the data obtained during the field activities for the above referenced work assignment. The purpose of this review is to provide a technical evaluation of the radiological results that were obtained by ALS Laboratory Group PA SOP 714R12 for Radium-228 by gas proportional counting for SDG 1103074 from ALS Laboratory Group (Fort Collins, CO). This report consists of eighteen water samples for the Colorado School of Mines/4349-430 project collected on March 1, 2011, March 2, 2011, and March 3, 2011 and submitted to ALS Laboratory Group on March 4, 2011. The samples were analyzed for Radium-228 by Radon Gas Proportional Counting on March 26, 2011, with reanalysis performed on April 6, 2011 (six samples) and April 12, 2011 (one sample). All analyses were conducted by ALS Laboratory Group. The field sample numbers and corresponding laboratory numbers are presented below:

Client Sample Number	Laboratory Sample Number	Matrix	Collection Date
CSMRI-10	1103074-1	Water	March 1, 2011
CSMRI-9	1103074-2	Water	March 1, 2011
CSMRI-4	1103074-3	Water	March 1, 2011
CSMRI-5	1103074-4	Water	March 1, 2011
CSMRI-14	1103074-5	Water	March 1, 2011
CSMRI-8B	1103074-6	Water	March 1, 2011
CSMRI-12	1103074-7	Water	March 1, 2011
CSMRI-7C	1103074-8	Water	March 2, 2011
CSMRI-13	1103074-9	Water	March 2, 2011
Equipment Blank	1103074-10	Water	March 2, 2011
CSMRI-11B	1103074-11	Water	March 2, 2011
CSMRI-6C	1103074-12	Water	March 2, 2011
SW-2	1103074-13	Water	March 2, 2011
SW-3	1103074-14	Water	March 2, 2011
CSMRI-1	1103074-15	Water	March 2, 2011
SW-1	1103074-16	Water	March 2, 2011
CSMRI-2	1103074-17	Water	March 3, 2011
CSMRI-1B	1103074-18	Water	March 3, 2011

Data validation was conducted in accordance with the Analytical Services Statement of Work for the following modules: Gas Proportional Counting Module RC04-v2, October 1, 2002 and U.S. DOE Quality Systems for Analytical Services Revision 2.6 (QSAS).

The radiological data were evaluated based on the following parameters:

- * Data Completeness
- * Holding Times and Preservation
- * Instrument Initial Calibrations
- * Instrument Performance Checks
- * Preparation Blanks
- * Duplicate Sample Results
- * Laboratory Control Samples (LCS) Results
- * Laboratory Control Samples Duplicate (LCSD) Results
- * Compound Quantitation and Reporting Limits (full validation only)

Data Completeness

The data package was complete as per ALS Laboratory Group Procedure SOP 724R11 for Radium-228 by Gas Flow Proportional Counting for SDG 1103074.

Holding Times and Preservation

Analytical holding times were evaluated and all criteria were met. However, holding time requirements are not applicable to radiochemistry analyses unless the isotopes of interest have short half-lives. The holding times for Radium-228 were met. No action was necessary.

Calibrations

The instruments were calibrated at the required frequency.

Initial Calibration

All instruments were calibrated properly using NIST traceable SRM.

Instrument Performance Checks

All isotopes were within criteria.

Preparation Blanks

Preparation/Method Blanks were performed at the required frequency. All isotopes that were analyzed had activities that were below their respective MDC's in their QC batch preparation blanks.

Duplicate Sample Analysis

Due to insufficient sample volume the laboratory prepared and analyzed a Laboratory Control Sample Duplicate (LCSD) in lieu of a client sample duplicate.

The laboratory re-prepared and reanalyzed samples CSMRI-4, CSMRI-8B, CSMRI-12, CSMRI-7C, CSMRI-6C, CSMRI-1, and CSMRI-2 due to failed LSC and LSCD. However, the laboratory duplicates associated with the samples were from a different client and were provided by the laboratory upon request.

All isotopic activities for Radium-228 LCS and LCSD analyses were within the limits of the statistical test for equivalency. No action was required.

Matrix Spike/Matrix Spike Duplicates

Matrix spike/matrix spike duplicates were not performed for the samples in this SDG, nor were any required.

Laboratory Control Samples

The LCS and LCSD radiometric recoveries for samples CSMRI-4, CSMRI-8B, CSMRI-12, CSMRI-7C, CSMRI-6C, CSMRI-1, and CSMRI-2 failed outside criteria high, the alpha counts also exceeded the crosstalk contamination threshold which will result in high bias. The laboratory re-prepared and reanalyzed the samples and met the control limits criteria for these samples. The laboratory did not re-prepare and reanalyze the remaining eleven samples which were non-detects CSMRI-10, CSMRI-9, CSMRI-5, CSMRI-14, CSMRI-13, Equipment Blank, CSMRI-11B, SW-2, SW-3, SW-1, and CSMRI-1B and these samples will be qualified as Estimated (UJ).

LCS analyses were performed at the required frequency. All recoveries for the reanalyzed samples were within 75-125% limits. No calculation errors or transcription errors were found.

Analyte Quantitation and Reporting Limits

Analyte quantitation was evaluated for all samples. No calculation or transcription errors were found. The results and reporting limits were correctly reported.

Overall Comments

The laboratory did not re-prepare and reanalyze samples CSMRI-10, CSMRI-9, CSMRI-5, CSMRI-14, CSMRI-13, Equipment Blank, CSMRI-11B, SW-2, SW-3, SW-1, and CSMRI-1B associated with the failed LCS and LCSD which were non-detects. The samples are qualified as Estimated (UJ).

The laboratory reported that the ICP-AES measurement of the added barium carrier prior to chemical separation had a concentration of less than the concentration added. The laboratory manually adjusted the values to the known concentration to calculate the chemical yield in order to avoid a low bias in all samples including the QC. All samples reported barium concentrations less than that known to be added. The results as reported are accepted without qualification.

DATA QUALIFIER DEFINITIONS

For the purpose of Data Validation, the following code letters and associated definitions are provided for use by the data validator to summarize the data quality.

- R Reported value is "rejected." Resampling or reanalysis may be necessary to verify the presence or absence of the compound.
- J The associated numerical value is an estimated quantity because the Quality Control criteria were not met.
- U J The reported quantitation limit is estimated because Quality Control criteria were not met. Element or compound was not detected.
- U The material was analyzed for, but was not detected above the level of the associated value. The associated value is either the sample quantitation limit or the sample detection limit.
- NR Result was not used from a particular sample analysis. This typically occurs when more than one result for an element is reported due to dilutions and reanalyses.



DATA VALIDATION REPORT

To:Robert HillFrom:John GarrettDate:April 17, 2011Project/Site:Colorado School of MinesProject No.:4349-430SDG No.:1103074 Inorganics Wet Chemistry

This report presents the inorganic anions data validation for the data obtained for eighteen CSMRI water samples collected on March 1, 2011, March 2, 2011, and March 3, 2011 and submitted to ALS Laboratory Group on March 4, 2011 for the above referenced work assignment. The purpose of this review is to provide a technical evaluation of the inorganic anions results that were obtained by preparation method MCAWW, May 1994, and EMSL Rev 2.1 Alkalinity, Bicarbonate, and Carbonate by Method 310.1 ALS SOP 1106R8, Sulfate, and Chloride by Method 300.0 Rev 2.1 ALS SOP 1113R11 from ALS Laboratory Group. (Fort Collins, CO). The water samples were analyzed for Bicarbonate, Carbonate, and Total Alkalinity, on March 14, 2011, Sulfate and Chloride on March8, 2011 and March 9, 2011. All analyses were conducted by ALS Laboratory Group. The field sample numbers and corresponding laboratory numbers are presented below:

Client Sample Number	Laboratory Sample Number	Matrix	Collection Date
CSMRI-10	1103074-1	Water	March 1, 2011
CSMRI-9	1103074-2	Water	March 1, 2011
CSMRI-4	1103074-3	Water	March 1, 2011
CSMRI-5	1103074-4	Water	March 1, 2011
CSMRI-14	1103074-5	Water	March 1, 2011
CSMRI-8B	1103074-6	Water	March 1, 2011
CSMRI-12	1103074-7	Water	March 1, 2011
CSMRI-7C	1103074-8	Water	March 1, 2011
CSMRI-13	1103074-9	Water	March 1, 2011
Equipment Blank	1103074-10	Water	March 1, 2011
CSMRI-11B	1103074-11	Water	March 1, 2011
CSMRI-6C	1103074-12	Water	March 1, 2011
SW-2	1103074-13	Water	March 1, 2011
SW-3	1103074-14	Water	March 1, 2011
CSMRI-1	1103074-15	Water	March 1, 2011
SW-1	1103074-16	Water	March 1, 2011
CSMRI-2	1103074-17	Water	March 1, 2011
CSMRI-1B	1103074-18	Water	March 1, 2011

Data validation was conducted in accordance with the USEPA Contract Laboratory Program National Functional Guidelines for Inorganic Data Review.

The Inorganic data were evaluated based on the following parameters:

- * Data Completeness
- * Holding Times and Preservation
- * Initial and Continuing Calibration Verification
- * Contract Required Detection Limit (CRDL)
- * Preparation/ Initial (ICB)/ and Continuing (CCB) Calibration Blanks
- * Interference Check Sample (ICSA) Results
- * Matrix Spike Results
- * Duplicate Sample Results
- * Laboratory Control Samples (LCS) Results
- * Serial Dilution Sample Results
- * Compound Quantitation and Reporting Limits (full validation only)
- * All criteria were met for this parameter

Data Completeness

The data package was complete. No results were qualified as a result of the missing data.

Holding Times and Preservation

The water samples were received intact at cooler temperatures 2.6 °C, 1.6°C, and 3.6°C.

Initial and Continuing Calibration Verification

Initial and Continuing Calibration Verification standards were analyzed at the required frequency and all were within the required 90-110%. No action was necessary.

Contract Required Detection Limit (CRDL)

All CRDL %Rs CRI %Rs were within 80-120% limits. No action was necessary.

Preparation and Initial/ Continuing Calibration Blanks

Preparation and Initial/Continuing Calibration Blank analyses were performed at the required frequency. Preparation and Initial/ Continuing Calibration Blanks are evaluated to assess the level of contamination in the preparation and analytical processes.

Preparation and Initial/ Continuing Calibration Blanks were prepared and analyzed at the required frequencies.

All of the blanks that were analyzed had concentrations that were below their respective Reporting Limits (RLs).

However, if blank results were above the Instrument Detection Limits (IDLs) and below the RLs, it caused the associated sample results to be qualified for contamination as estimated and non-detected [UJ 107]. If blank results were below the negate IDL and above the negate RL, it caused the associated sample results to be qualified for negative contamination as estimated [J 107]. No sample results were qualified due to blank contamination.

Matrix Spike/Matrix Spike Duplicate Results

MS/MSD analyses were performed at the required frequency. All MS/MSD percent recoveries were within 75-125% limits with the following exception:

The chloride concentration was above the analytical range in the native sample CSMRI-1B and MS/MSD recoveries could not be evaluated. The associated LCS, ICV, and CCV results were within control limits and no action was necessary.

Duplicate Sample Analysis

Duplicate analyses were performed at the required frequency. All original sample/duplicate sample and MS/MSD differences were less than 20% RPD or less than the RDL for results less than (5)(RDL). No actions were necessary.

Laboratory Control Samples

LCS analyses were performed at the required frequency. The laboratory analyzed laboratory control samples for all analytes. All recoveries were within 80-120% limits. No action was necessary.

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Serial Dilution Results

All %Ds were less than 10% for all analytes.

Analyte Quantitation and Reporting Limits

Analyte quantitation was evaluated for all samples. No calculation or transcription errors were found. The results and reporting limits were correctly reported.

Overall Comments

The chloride Matrix Spike and Matrix Spike Duplicate recoveries were outside criteria low in sample CSMRI-10 and the Matrix spike recovery low in CSMRI-1B but greater than 75%. The sample results were greater than 4X the Spike level and the associated LCS, ICV, and CCV results were within control limits and no action was necessary.

Samples CSMRI-10, CSMRI-9, CSMRI-4, CSMRI-5, CSMRI-13, CSMRI-11B, CSMRI-6C, SW-2, SW-3, CSMRI-1, CSMRI-2, SW-1, and CSMRI-1B were analyzed at a dilution in order to bring chloride concentrations into analytical range of the IC. Sample CSMRI-14 was analyzed at a dilution in order to bring chloride and sulfate concentrations into analytical range of the IC

Reduced aliquots were analyzed for samples CSMRI-10, CSMRI-9, CSMRI-4, CSMRI-5, CSMRI-10, CSMRI-8B, CSMRI-12, CSMRI-7C, CSMRI-13, CSMRI-11B, CSMRI-6C, CSMRI-2, and CSMRI-1B for alkalinity, bicarbonate, and carbonate. The laboratory elevated the reporting limits accordingly.

The results as reported are accepted without qualification.

DATA QUALIFIER DEFINITIONS

For the purpose of Data Validation, the following code letters and associated definitions are provided for use by the data validator to summarize the data quality.

- R Reported value is "rejected." Resampling or reanalysis may be necessary to verify the presence or absence of the compound.
- J The associated numerical value is an estimated quantity because the Quality Control criteria were not met.
- U J The reported quantitation limit is estimated because Quality Control criteria were not met. Element or compound was not detected.
- U The material was analyzed for, but was not detected above the level of the associated value. The associated value is either the sample quantitation limit or the sample detection limit.
- NR Result was not used from a particular sample analysis. This typically occurs when more than one result for an element is reported due to dilutions and reanalyses.



DATA VALIDATION REPORT

To:Robert HillFrom:John GarrettDate:April 17, 2011Project/Site:Colorado School of MinesProject No.:4349-430SDG No.:1103074

This report presents the inorganic metals data validation for the data obtained for eighteen water samples collected on March 1, 2011, March 2, 2011, and March 3, 2011 and submitted to ALS Laboratory Group and submitted to ALS Laboratory Group on March 4, 2011 for the above referenced work assignment. The purpose of this review is to provide a technical evaluation of the inorganic metals results that were obtained by SW-846, 3rd edition, Method 6010B and ALS Laboratory Group SOP 834R8 for trace metals by Inductively Coupled Plasma (ICP) atomic emission spectrometry analysis, Total Uranium by Method 6020A ALS Laboratory Group Procedure SOP 827R7 by Inductively Coupled Plasma mass spectrometry (ICP-MS) analysis for SDG 1103074 by ALS Laboratory Group (Fort Collins, CO). The Total ICP metals and Uranium by ICP-MS were extracted on March 9, 2011 and analyzed on March 10, 2011. All analyses were conducted by ALS Laboratory Group. The field sample numbers and corresponding laboratory numbers are presented below:

Client Sample Number	Laboratory Sample Number	Matrix	Collection Date
CSMRI-10	1103074-1	Water	March 1, 2011
CSMRI-9	1103074-2	Water	March 1, 2011
CSMRI-4	1103074-3	Water	March 1, 2011
CSMRI-5	1103074-4	Water	March 1, 2011
CSMRI-14	1103074-5	Water	March 1, 2011
CSMRI-8B	1103074-6	Water	March 1, 2011
CSMRI-12	1103074-7	Water	March 1, 2011
CSMRI-7C	1103074-8	Water	March 1, 2011
CSMRI-13	1103074-9	Water	March 1, 2011
Equipment Blank	1103074-10	Water	March 1, 2011
CSMRI-11B	1103074-11	Water	March 1, 2011
CSMRI-6C	1103074-12	Water	March 1, 2011
SW-2	1103074-13	Water	March 1, 2011
SW-3	1103074-14	Water	March 1, 2011
CSMRI-1	1103074-15	Water	March 1, 2011
SW-1	1103074-16	Water	March 1, 2011
CSMRI-2	1103074-17	Water	March 1, 2011
CSMRI-1B	1103074-18	Water	March 1, 2011

Data validation was conducted in accordance with the USEPA Contract Laboratory Program National Functional Guidelines for Inorganic Data Review.

The metals data were evaluated based on the following parameters:

- * Data Completeness
- * Holding Times and Preservation
- * Initial and Continuing Calibration Verification
- Contract Required Detection Limit (CRDL)
 Preparation/ Initial (ICB)/ and Continuing (CCB) Calibration Blanks
- * Interference Check Sample (ICSA) Results
- * Matrix Spike Results
- * Duplicate Sample Results
- * Laboratory Control Samples (LCS) Results Serial Dilution Sample Results
- * Compound Quantitation and Reporting Limits (full validation only)
- * All criteria were met for this parameter

Data Completeness

The data package was complete except for the missing CRDL (2B) and IDL (10) QC Summary Forms. No results were qualified as a result of the missing data.

Holding Times and Preservation

Analytical holding times were evaluated and all criteria were met.

The water samples were all field filtered and had a pH less than 2. Samples were received at cooler temperatures 2.6 °C, 1.6°C, and 3.6°C.

Initial and Continuing Calibration Verification

Initial and Continuing Calibration Verification standards were analyzed at the required frequency and all were within the required 90-110% limits for ICP. No action was necessary.

Contract Required Detection Limit (CRDL)

No CRDL or CRI standard recovery summary forms (EPA Form 2b) were included in the data package. The reviewer obtained the %Rs from the instrument raw data. All CRDL %Rs for ICP were within 80-120% limits. No action was necessary.

Preparation and Initial/ Continuing Calibration Blanks

Preparation and Initial/Continuing Calibration Blank analyses were performed at the required frequency. Preparation and Initial/ Continuing Calibration Blanks are evaluated to assess the level of contamination in the preparation and analytical processes.

Preparation and Initial/ Continuing Calibration Blanks were prepared and analyzed at the required frequencies.

All of the blanks that were analyzed had concentrations that were below their respective Reporting Limits (RLs).

However, if blank results were above the Instrument Detection Limits (IDLs) and below the RLs, it caused the associated sample results to be qualified for contamination as estimated and non-detected [UJ 107]. If blank results were below the negative IDL and above the negative RL, it caused the associated sample results to be qualified for negative contamination as estimated [J 107]. No sample results were qualified due to blank contamination.

Interference Check Sample (ICSA) Results

Interference Check Samples were prepared and analyzed at the required frequencies. No action was necessary.

Matrix Spike/Matrix Spike Duplicate Results

MS/MSD analyses were performed at the required frequency. All ICP and ICP-MS percent recoveries were within 75-125% limits with the following exception:

Duplicate Sample Analysis

Duplicate analyses were performed at the required frequency. All ICP, and ICP-MS original sample/duplicate sample and MS/MSD differences were less than 20% RPD or less than the RDL for results less than (5)(RDL).

Laboratory Control Samples

LCS analyses were performed at the required frequency. The laboratory analyzed laboratory control samples for all metals. All recoveries were within 80-120% limits. No action was necessary.

Serial Dilution Results

No Serial Dilutions were performed.

Analyte Quantitation and Reporting Limits

Analyte quantitation was evaluated for all samples. No calculation or transcription errors were found. The results and reporting limits were correctly reported.

Overall Comments

No CRDL or CRI standard recovery summary forms (EPA Form 2b) were included in the data package. The reviewer obtained the results from the raw data. No action was necessary.

DATA QUALIFIER DEFINITIONS

For the purpose of Data Validation, the following code letters and associated definitions are provided for use by the data validator to summarize the data quality.

- R Reported value is "rejected." Resampling or reanalysis may be necessary to verify the presence or absence of the compound.
- J The associated numerical value is an estimated quantity because the Quality Control criteria were not met.
- U J The reported quantitation limit is estimated because Quality Control criteria were not met. Element or compound was not detected.
- U The material was analyzed for, but was not detected above the level of the associated value. The associated value is either the sample quantitation limit or the sample detection limit.
- NR Result was not used from a particular sample analysis. This typically occurs when more than one result for an element is reported due to dilutions and reanalyses.

DATA VALIDATION REPORT

To:	Robert Hill	
From:	John Garrett	
Date:	April 16, 2011	
Project/Site:	Colorado School of Mines	
Project No.:	4349-430	
SDG No.:	1103074 DOC	

This report presents the Dissolved Organic Carbon data validation for the data obtained for eighteen CSMRI water samples collected on March 1, 2011, March 2, 2011, and March 3, 2011 and submitted to ALS Laboratory Group on March 4, 2011 for the above referenced work assignment. The purpose of this review is to provide a technical evaluation of eleven Dissolved Organic Carbon results that were obtained by MCAWW, May 1994, Method 415.1, SOP 670R14 Dissolved Organic Carbon by Method 415.1 from ALS Laboratory Group (Fort Collins, CO). The water samples were analyzed March 9, 2011. All analyses were conducted by ALS Laboratory Group. The field sample numbers and corresponding laboratory numbers are presented below:

Client Sample Number	Laboratory Sample Number	Matrix	Collection Date
CSMRI-10	1103074-1	Water	March 1, 2011
CSMRI-9	1103074-2	Water	March 1, 2011
CSMRI-4	1103074-3	Water	March 1, 2011
CSMRI-5	1103074-4	Water	March 1, 2011
CSMRI-14	1103074-5	Water	March 1, 2011
CSMRI-8B	1103074-6	Water	March 1, 2011
CSMRI-12	1103074-7	Water	March 1, 2011
CSMRI-7C	1103074-8	Water	March 1, 2011
CSMRI-13	1103074-9	Water	March 1, 2011
Equipment Blank	1103074-10	Water	March 1, 2011
CSMRI-11B	1103074-11	Water	March 1, 2011
CSMRI-6C	1103074-12	Water	March 1, 2011
SW-2	1103074-13	Water	March 1, 2011
SW-3	1103074-14	Water	March 1, 2011
CSMRI-1	1103074-15	Water	March 1, 2011
SW-1	1103074-16	Water	March 1, 2011
CSMRI-2	1103074-17	Water	March 1, 2011
CSMRI-1B	1103074-18	Water	March 1, 2011

Data validation was conducted in accordance with the USEPA Contract Laboratory Program National Functional Guidelines for Organic Data Review (CLP).

The Dissolved Organic Carbon data were evaluated based on the following parameters:

- * Data Completeness
- * Holding Times and Preservation
- * Initial and Continuing Calibration Verification
- * Contract Required Detection Limit (CRDL)
- * Preparation/ Initial (ICB)/ and Continuing (CCB) Calibration Blanks
- * Interference Check Sample (ICSA) Results
- * Matrix Spike Results
- * Duplicate Sample Results
- * Laboratory Control Samples (LCS) Results
- * Serial Dilution Sample Results
- * Compound Quantitation and Reporting Limits (full validation only)
- * All criteria were met for this parameter

Data Completeness

The data package was complete. No results were qualified as a result of the missing data.

Holding Times and Preservation

Analytical holding times were evaluated and all criteria were met.

The water samples were all field filtered and had a pH less than 2. Samples were received at cooler temperatures 2.6 °C, 1.6°C, and 3.6°C.

Initial and Continuing Calibration Verification

Initial and Continuing Calibration Verification standards were analyzed at the required frequency and all were within the required 90-110%. No action was necessary.

Contract Required Detection Limit (CRDL)

All CRDL %Rs CRI %Rs were within 80-120% limits. No action was necessary.

Preparation and Initial/ Continuing Calibration Blanks

Preparation and Initial/Continuing Calibration Blank analyses were performed at the required frequency. Preparation and Initial/ Continuing Calibration Blanks are evaluated to assess the level of contamination in the preparation and analytical processes.

Preparation and Initial/ Continuing Calibration Blanks were prepared and analyzed at the required frequencies.

All of the blanks that were analyzed had concentrations that were below their respective Reporting Limits (RLs).

However, if blank results were above the Instrument Detection Limits (IDLs) and below the RLs, it caused the associated sample results to be qualified for contamination as estimated and non-detected [UJ 107]. If blank results were below the negate IDL and above the negate RL, it caused the associated sample results to be qualified for negative contamination as estimated [J 107]. No sample results were qualified due to blank contamination.

Matrix Spike/Matrix Spike Duplicate Results

MS/MSD analyses were performed at the required frequency. All MS/MSD percent recoveries were within 75-125% limits. No action was necessary.

Duplicate Sample Analysis

The laboratory analyzed a LSC and LCSD in lieu of client sample duplicate. The duplicate analyses were performed at the required frequency. All original sample/duplicate sample and MS/MSD differences were less than 20% RPD or less than the RDL for results less than (5)(RDL). No actions were necessary.

Laboratory Control Samples

LCS analyses were performed at the required frequency. The laboratory analyzed laboratory control samples for all analytes. All recoveries were within 80-120% limits. No action was necessary.

Serial Dilution Results

No dilutions were required.

Analyte Quantitation and Reporting Limits

Analyte quantitation was evaluated for all samples. No calculation or transcription errors were found. The results and reporting limits were correctly reported.

Overall Comments

The overall data are acceptable as reported.

DATA QUALIFIER DEFINITIONS

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- NR Result was not used from a particular sample analysis. This typically occurs when more than one result for an element is reported due to dilutions and reanalyses.

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Sample Station	CSMRI-1B	CSMRI-2	CSMRI-4	CSMRI-5

Table H-1 Historical Summary of Radioisotopes in Groundwater (Stoller)

Table H-1

Page 1

U-235	(pCi/l) NT	NT N		+	+				_	-		-		-	NT NT	TN TN TN	NT NT NT	NT NT	NTN	NT NT	NT NT NT	NT NT	NT NT NT NT	NT NT	NT NT	NT NT NT	NT NT	TN TN TN	NT NT	NT NT	NT NT	TN TN TN	NT NT NT	NT	NT NT NT	NT NT	TN TN TN TN	NT NT	NT NT NT	NT	NT NT NT NT	NT NT	NT NT NT	NT NT	NT NT NT	TN	NT NT	NT NT NT NT	TN TN	NT NT	NT NT NT NT	NT NT	NT NT NT NT	NT NT		NT NT	NT NT	1	NT NT			_
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H1. Sample Samole Date	Station 2/27/2007	CSMRI-6B 9/10/2007 11/27/2007	2/28/2008	7/11/08 (DRY)	3/16/09 (DRY)	8	CSMRI-6C 7/2/18/2009	3/8/10 (DRY)	9/8/2010	12/8/2010	2/27/2007 6/26/2007	9/10/2007	11/26/2007	2/20/2000 4/15/08 (DRY)	9/24/08 (DRY)	CSMRI-7B 12/3/08 (DRY)	6/24/09 (DRY)	9/25/09 (DRY)	3/8/10 (DRY)	6/10/2010	12/8/2010	3/8/2007	6/27/2007	11/27/2007	2/27/2008	4/17/2008 9/25/2008	CSMRI-8 12/5/2008	3/18/2009	9/24/2009	12/16/2009	6/8/2010	9/8/2010	2/27/2007	6/26/2007	9/10/2007	2/27/2008	4/15/2008 9/24/2008	CSMRI-9 12/5/2008	3/16/2009 6/22/2009	9/24/2009	3/11/2010	6/9/2010	9/8/2010 12/7/2010	3/1/2007	9/10/2007	11/26/2007	4/15/2008		3/16/2009	9/25/2009	12/16/2009 3/11/2010	6/9/2010	9/8/2010 12/7/2010	3/1/2007	6/26/2007 9/10/2007		4/15/2008	12/3/08 (DRY) 3/16/09 (DRY)	6/24/2009	9/25/2009 CSMRL11B 12/18/2009		6/8/2010

Table H-1 Historical Summary of Radioisotopes in Groundwater (Stoller)

-waxmum Contaminant Level - Naional Prinnary Drinking Wate **5 CCR 1002-41 Reg 41 - Colorado Groundwater Standards pCtH - picoenties per liter J - Estimated NE - Ne festablished NT - not tested MT - micrograms per liter

Table H-1

Zn 2n 0.032 0.032 0.034 0.015 (B) 0.049 0.048 0.048 0.048 0.048 0.038 0.048 0.048 0.038 0.048 0.048 0.048 0.048 0.048 0.038 0.048 0.038 0.048 0.038 0.017 (B) 0.037 0.038 0.015 (I) 0.037 0.038 0.015 (I) 0.037 0.007 0.0370000000000	ND ND 0.0012 (E) 0.0033 (E) 0.0033 (E) 0.0033 (E) NT NT NT NT NT NT NT NT NT NT NT NT NT	NT NT NT NT NT NT NT NT NT NT NT NT NT N	0.067 0.047 0.047 0.047 0.047 0.047 0.047 0.047 0.047 0.044 0.11 0.054 0.112 0.0034 0.125 0.0034 0.122 0.0034 0.112 0.0034 0.112 0.0034 0.117 0.0033 0.117 0.117 0.117 0.0034 0.117 0.0034 0.117 0.0034 0.117 0.0034 0.117 0.0034 0.117 0.0033 0.117 0.0034 0.117 0.0033 0.117 0.0033 0.117 0.0033 0.117 0.0033 0.117 0.0033 0.117 0.0033 0.117 0.0033 0.0172 0.0033 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.0030 0.00000000
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Statition C C S MR R - -	CSMRI-18 CSMRI-18 CSMRI-2	CCSSWR CCSSWR	C S MR

Table H-2Historical Summary of Metals in Groundwater (Stoller)(Results in milligrams per liter - U in micrograms per liter)

Table H-2

Page 1

Table H-2 Historical Summary of Metals in Groundwater (Stoller) (Results in milligrams per liter - U in micrograms per liter)

NT NU ND ND 0.0051 (B) ND 0.0048 NT NT NT NT NT NT NT NT NT NT NT	NT 0.0041 (B) NT NT NT NT NT NT NT NT NT NT NT	0.0024 (B) 0.0059 0.056 0.011 (B) 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.032 0.011 (B) 0.011 (B) 0.012 (B) 0.011 (B) 0.011 (B) 0.012 (B) 0.011 (B) 0.011 (B) 0.011 (B) 0.011 (B) 0.011 (B) 0.011 (B) 0.012 (B) 0.011 (B) 0.011 (B) 0.012 (B) 0.011 (B) 0.012 (B) 0.011 (B) 0.01	ND 0.0095 (B) 0.0097 (B) 0.015 (B) 0.011 0.015 (B) 0.011 0.011 0.011 NT NT NT NT NT NT NT NT NT NT NT NT NT		0.0023 (B) ND 0.0033 (B) ND ND NT NT NT NT NT NT NT NT NT NT NT NT NT
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Station CSMRI-6B CSMRI-6B	CSMRI-7B	CSMRI-8	CSMRI-9	CSMR1-10	CSMRI-11 CSMRI-11 CSMR1-11B CSMR1-11B Detection Limits Maximum Cont

*Maximum Contaminan ND - non detect NE - not established NT - not tested (B) - Detected above In

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Table H-2

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Sample	Sample	Ka-220	Ra-228	10-228	h-230	Th-232	U-234	1-235	1.738
Station		(pCI/I)	(pCi/l)	(pCi/l)	(pCi/l)	(pCi/l)	(pCi/l)	(pCi/l)	(IDCI/I)
	2/25/2005	0	0.58	0.018	-0.026	-0.001	0.89	0.083	0.65
	6/14/2005	0.14	0.05	0.05	-0.025	0.016	0.246	0.021	0.251
	9/7/2005	0.18	0.42	0.041	0.25	0.102	0.35	0.031	0.35
	12/20/2005	-0.31	0.47	0.028	0.197	-0.005	0.64	0.041	0.7
	3/15/2006	-0.16	0.35	0.059	0.125	0.005	0.6	0.029	0.53
	6/14/2006	0.13	0.45	0.16	0.53	0.062	0.11	0.08	0.19
	9/13/2006	-0.03	0.25	-0.019	-0.035	0.01	0.37	-0.005	0.34
	3/1/2007	-0.1	0.25	-0.038	0.15	0.026	NT	NT	NT
	6/27/2007	0.13	0.77	0.006	0.016	0.014	NT	NT	μ
	9/11/2007	0.15	0.74	0.063	0.088	0.012	NT	NT	ΓL
	11/27/2007	0.2	0.24	0.026	0.049	0.025	ΝŢ	NT	ΓN
SW-1	2/27/2008	0.1	0.48	0.014	0.002	0.024	NT	NT	NT
	4/18/2008	0.06	-0.07	-0.023	-0.026	0.012	NT	NT	NT
	9/25/2008	0.18	-0.01	NT	NT	NT	NT	NT	NT
	12/3/2008	-0.06	0.34	NT	NT	NT	NT	NT	NT
	3/16/2009	0.14	0.73	NT	NT	NT	NT	NT	NT
	6/24/2009	0.33	1.228 J	NT	NT	NT	NT	NT	NT
	9/24/2009	-0.08	0.37	NT	NT	NT	NT	NT	NT
	12/17/2009	0.1	0.42	NT	NT	NT	NT	NT	ΝT
	3/9/2010	-0.04	0.2	NT	NT	NT	NT	NT	NT
	6/9/2010	0.07	0.44 (J)	NT	NT	NT	NT	NT	ΤN
	9/9/2010	0.04	0.21	NT	NT	NT	NT	NT	NT
	12/8/2010	0.03	0.58	NT	NT	NT	NT	NT	NT
	2/25/2005	0.45	0.06	0.011	-0.016	0.033	0.8	0.066	0.42
	6/14/2005	0.04	0.29	0.071	-0.028	0.007	0.259	0.032	0.23
	9/7/2005	-0.08	0.24	-0.013	0.107	0.051	0.54	0.014	0.54
	12/20/2005	0.09	0.07	-0.003	0.126	0	0.71	0.067	0.49
	3/15/2006	-0.04	-0.15	0.009	0.184	0.01	0.79	0.004	0.51
	6/14/2006	0.03	0.04	0.172	0.24	0.1	0.39	0	0.48
	9/13/2006	0.11	0.35	0.009	-0.03	0.01	0.43	-0.006	0.3
	3/8/2007	0.12	0.73	0.047	-0.055	0	τN	NT	ΝΤ
	6/28/2007	0.02	0.78	0.028	0.014	0	NT	NT	NT
	9/11/2007	0.1	0.27	0.066	0.068	0.002	NT	NT	ħ
0.000	11/26/2007	0.11	0.36	0.007	0	0.012	ΔL	NT	μ
SW-2	2/26/2008	0.1	0	-0.01	0.113	0.011	ĻΝ	NT	NT
	4/18/2008	0.13	0.58	0.015	0.24	0.024	NT	NT	NT
	9/24/2008	-0.16	-0.02	NT	NT	NT	τz	NT	τN
	12/3/2008	0.1	0.46	IN	IN	IN	IN I	IN	Į.
	3/ 10/2009	0.20	67.0	-N	Z	Z		N	z
	0/24/2009	0.03	0.47 0	NI	N	Z	Z	Z	z
	12/17/2009	0.03	0.44	NT	NT	L L	TN	TN	Z L
	3/9/2010	-0.03	0.97	NT	TN TN	TN	IN	TIN	
	6/9/2010	0.07	-0.06	NT	NT	LN LN	TN TN	IN	
	9/9/2010	0.2	0.16	NT	NT	NT	TN	NTN	NT
	12/8/2010	0.02	0.24	NT	NT	TN	NT	NT	IZ
	6/10/2010	0.39	0.01	NT	NT	NT	NT	NT	IN
SW-3	9/9/2010	0.13	0.21	NT	NT	NT	NT	NT	NT
	12/8/2010	0.06	0.26	NT	NT	NT	NT	NT	NT
	* 101 4			1	F COOC	1100 000 H 000 H		1	

**5 CCR 1002-31 Reg 31 – Colorado Surface Water Standards pCi/l - picoCuries per liter µg/l – micrograms per liter

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Table H-3

-						(Results in		sults in milligrams per liter - U in micrograms per liter)	U in micre	ograms p	er liter)						
Station	Sample Date	Ag	As	Ba	g	8	5	å	¥	Mg	Ŷ	R	å	Še	Э	^	5
	2/25/2005	Q	QN	QN	QN	DN	DN	DN	QN	ND	Q	QN	QN	QN	1.97	ND	0.2
	6/14/2005	Q	QN	QN	#	ND	QN	DN	1.1	2.8	QN	5.2	QN	QN	0.75	QN	0.09
	9/7/2005	9	0.0037 (B)	0.029 (B)	20	QN		QN	2.2	4.4	0.0044 (B)	8.5	ND	0.0045 (B)	1.04	QN	0.063
	12/20/2005	2	9	0.042 (B)	35	0.00057 (B)	UN	0.000034 (B)	3.7	7.6	0.004 (B)	19	ND	QN	2.11	QN	0.22
	0007/01/2			0.04 (B)	3/	U.UUU84 (B)	0.0	0.000024 (B)	3.7	8.5	0.0048 (B)	23	QN	Q	1.59	0.00067 (B)	0.19
	9/13/2006				2.0					9.1	0.0042 (B)	3.1	ON C	QN	0.61	Q	0.029
	3/1/2007		GN	0.040 (B)	17	0.0011 (B)	UN 00000 / UN		1.2	4.4	0.0045 (B)	Q'D	N		- 1	ON C	0.053
	6/27/2007	2	Q	0.018 (B)	10	UN UN	UN ND	0 0000068 (B)	0 93 (B)	25	0.0040 (B)	3.7	UN UN		1.1		0.067
	9/11/2007	QN	QN	0.032 (B)	21	QN	QN	0.000019	1.7	5.2	0.0029 (B)	7.4	GN	C C C	76.0		0.078
	11/27/2007	QN	QN	0.042 (B)	33	0.00076 (B)	QN	0.00027 (B)	2.8	8.2	0.0032 (B)	15	QN	Q	1.8	QN	0.18
SW-1	2/27/2008	Q	Q	0.042 (B)	36	ND	ND	ND	3.3	9.6	0.0022 (B)	19	QN	QN	2	QN	0.15
	4/18/2008	QN	QN	0.044 (B)	35	0.00044 (B)	ND	ND	3.4	6	0.0034 (B)	23	QN	QN	1.9	QN	0.13
	9/25/2008	NT	ΔŢ	NT	23	NT	NT	NT	1.9	5.1	NT	0	NT	NT	1.1	NT	NT
	12/3/2008	IN	NT	NT	32	μ	NT	NŢ	ю	7.1	NT	15	NT	NT	1.6	NT	NT
	3/16/2009	IN	μŢ	NT	35	μT	NT	ΝŢ	3.1	8.9	NT	17	NT	NT	1.9	NT	NT
	6/24/2009	0.00078	0.0032	0.017	8.7	0.00016	0.00041	0.000024	0.92	2.1	3.3	0.0019	0.00035	NT	0.55	NT	NT
	9/24/2009	z	IN	IN	25 (J)	ΔL	NT	NT	1.4	5.5 (J)	NT	9.7 (J)	NT	NT	1.1	NT	NT
	12/17/2009	IZ !	LΝ	NT	39	NT	NT	NT	2.8	8.5	NT	18	NT	NT	1.7	NT	NT
	3/9/2010		IN	TN NT	40	IN	NT	NT	2.8	11	ħ	21	NT	NT	2	NT	NT
	0102/8/0	NN.	NN.	0.012 (B)	8.4	ON .	0.001 (B)	0.000027 (B)	0.47 (B)	1.9	μŢ	2.8	ND	NT	0.46	QN	NT
	9/9/2010		IN!	IN	23	NT	NT	NT	1.7	5.1	μŢ	თ	NT	NT	~	NT	NT
	12/8/2010	z	Z	IN	38	NT	NT	NT	2.5	8.3	NT	14	NT	NT	1.6	NT	NT
	2/25/2005	QN	QN	Q	QN	QN	QN	DN	QN	ND	QN	Q	ND	Q	1.29	DN	0.17
	6/14/2005	Q	QN	Q	ŧ	QN,	QN	QN	1.1	2.8	QN	4.8	QN	QN	0.69	ND	0.085
	9/7/2005	2	2	0.028 (B)		Q	QN	QN	2.1	4.4	0.0037 (B)	8.7	QN	0.0037 (B)	1.62	QN	0.051
	CUU2/U2/21	NN.	AN .	0.042 (B)		0.00043 (B)	QN	0.000034 (B)	3.8	8	0.0038 (B)	19	QN	QN	1.5	QN	0.21
	3/15/2006	UN UN	UN ND ND	0.042 (B)	66	0.00053 (B)	0.00055 (B)	0.000022 (B)	3.8	8.9	0.0046 (B)	25	QN	QN	1.52	0.00053 (B)	0.2
	0/14/2000		U.UUZZ (B)	(E) LUTI (E)	8.4	NN.	ON .	ON .	-	1.9	0.0045 (B)	m	QN	DN	1.44	QN	0.031
	9/13/2006		O DOED ID	0.03 (B)	21	ON UN		QN	2.1	4.4	0.0048 (B)	8.5	QN	Q	0.89	Q	0.04
	2002/00/2	NID		0.040 (b)	20	U.UUU04 (B)		DND CONCEPT	4.2	8.8	0.0014 (B)	27	Q .	QN	1.7	Q	0.17
	9/11/2007	Q	Q	0.033 (B)	21	ON ON	GN	0 00001	(a) cs:n	5.1	0.0035 /BV	3.3			10.0		6/0.0
	11/26/2007	DN	QN	0.044 (B)	35	0.0005 (B)		0.00027 (B)	2.9	8.6	0.0027 (B)	15	QN	QN	1.7	Q	0.19
SW-2	2/26/2008	QN	QN	0.051	35	0.0005 (B)	ND	ND	3.1	9.2	0.0023 (B)	21	DN	QN	2	QN	0.15
	4/18/2008	QN	QN	0.045 (B)	35	0.0005 (B)	QN	DN	3.4	9.1	0.0031 (B)	23	ND	QN	1.8	QN	0.14
	9/24/2008	IN	NT	NT	23	τN	NT	NT	1.9	5.1	NT	6	NT	NT	0.99	NT	NT
	12/3/2008	IN	NT	N	31	NT	NT	NT	e	7.5	NT	15	NT	NT	1.5	NT	NT
	3/16/2009	IN	NT	IN	37	NT	NT	μŢ	3.5	9.7	μ	19	NT	NT	1.9	NT	NT
	6/24/2009	0.00078	0.0032	0.016	8.7	0.00016	0.00041	0.000027	0.9	2.2	3.3	0.0019	0.00035	NT	0.059	NT	NT
	9/24/2009	M	NT	NT	25 (J)	NT	NT	NT	1.4	5.5 (J)	ΝŢ	9.4 (J)	NT	NT	1.1	NT	NT
	12/17/2009	NT	NT	NT	42	NT	NT	μŢ	в	9.8	ΤN	19	NT	NT	1.9	NT	NT
	3/9/2010	0.01	0.01	0.1	-	0.005	0.01	0.0002	1	-	NT	-	0.003	NT	2	0.01	NT
	6/9/2010	2	Q !	0.012 (B)	8	QN	Q	0.000024 (B)	0.49 (B)	1.9	NT	2.7	QN	NT	0.52	QN	NT
	9/9/2010	Z	IN !	Į.	23	TN -	LN :	NT	1.7	5.3	NT	9.2	NT	NT	-	NT	NT
	12/8/2010	Z	z	N N N	40	IN IN	Z	IN TN	2.5	8.8	Łz	14	NT	NT	1.7	NT	IN
CIM.2	0/10/2010	UN FN	NU	U.UIZ (B)	0.4 00	NU	ND 1	0.000024 (B)	0.5 (B)	1.9	LZ	2.7	Q !	IN	0.49	Q	NT
5	10101010	LIV	- LI		62	IN	N	Z	1.1	2.2	z	с. ²	N	Z	0.98	Z	LZ

Tabie H-4 Historical Summary of Metals in Surface Water (Stoller) Results in millerams ner liter - 17 in microorams ner liter

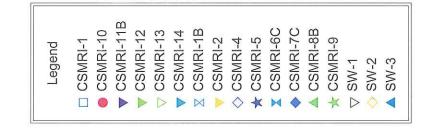
MCL* 0.01 0.01

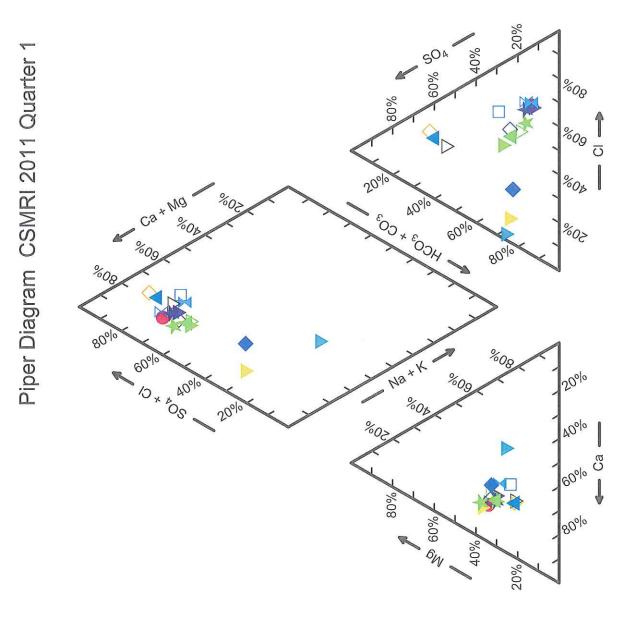
ND - Non Detect

(B) – Detected above Instrument Detection Level but helow Re

Table H-4

Page





			CSM	IRI	-1
NN/- 4 70	0.01				
Water Type Dissolved Solids	Ca-Cl	2	10 mg/I		Measured
	340.92 mg/kg 0.99729 g/cm ²		40 mg/L		Calculated
Density	0.99729 g/cm 752 μmho/cm				Measured
Conductivity					Measured
Hardness (as CaCO		14	00.00 /I		
Total	198.77 mg/kg		98.23 mg/L		Calculated
Carbonate	129.94		29.59		
Non-Carbonate	68.828	68	8.642		
Primary Tests					
Anion-Cation Bala	100				
Anions		5.47			
Cations		5.6			
% Difference		1.156			OK
Measured $TDS = C$	alculated TDS				
Measured		340.925			
Calculated		364.990			
Ratio		0.934			Not within range 1.0 to 1.2
Measured EC = Ca	lculated EC				an na hair an suite an suite an suite an suite an suite an the suite and the suite and the suite and the suite
Measured		752.000			
Calculated		579.768			
Ratio		1.297			Not within range 0.9 to 1.1
Secondary Tests					
Measured EC and	Ion Sums:				
Anions		0.727876			Not within preferred range
(0.9-1.1)					* 47653787 82 85
Cations		0.744899	N.		Not within preferred range
(0.9-1.1)		121 122 2			
Calculated TDS to	EC ratio	0.485			Not within preferred range
(0.55-0.7)		12 19 22			
Measured TDS to F	EC ratio	0.453			Not within preferred range
(0.55-0.7)					
Organic Mass Bala					
$DOC \ge Sum of Org$					
DOC unavailable					

			CSMF	RI-10
	0.01			
Water Type	Ca-Cl		(00 /I	1
Dissolved Solids	691.69 mg/kg		690 mg/L	Measured
Density	0.99755 g/cm			Calculated
Conductivity	1520 µmho/ci	m		Measured
Hardness (as CaCO	5.			
Total	502.65 mg/kg		501.42 mg/L	Calculated
Carbonate	378.21		377.28	
Non-Carbonate	124.45		124.14	
Primary Tests				
Anion-Cation Bala	nce			
Anions		11.8		
Cations		11.9		
% Difference		0.120		OK
Measured TDS = C	Calculated TDS			
Measured		691.69	93	
Calculated		785.22	21	
Ratio		0.881		Not within range 1.0 to 1.2
Measured EC = Ca	lculated EC			(.)
Measured		1520.0	000	
Calculated		1139.9	961	
Ratio		1.333		Not within range 0.9 to 1.1
Secondary Tests				
Measured EC and	Ion Sums:			
Anions		0.7794	408	Not within preferred range
(0.9-1.1)				
Cations		0.7812	287	Not within preferred range
(0.9-1.1)				control of Burgerstrom March Art
Calculated TDS to (0.55-0.7)	EC ratio	0.517		Not within preferred range
Measured TDS to I	EC ratio	0.455		Not within preferred range
(0.55-0.7)				
Organic Mass Bala				
$DOC \ge Sum of Org$		1 100	17	
Dissolved Organi	ic Carbon	1.400		01/
Sum of Organics		0.000	mg/L	OK

		CSM	RI-11B
Water Type	Ca-Cl		
Dissolved Solids	681.67 mg/kg	680 mg/L	Measured
Density	0.99755 g/cm		Calculated
Conductivity	1230 µmho/cr		Measured
Hardness (as CaCC			
Total	486.14 mg/kg	484.95 mg/L	Calculated
Carbonate	378.21	377.28	Curtosiniou
Non-Carbonate	107.94	107.67	
Primary Tests			
Anion-Cation Bala	nce		
Anions		11.9	
Cations		12.3	
% Difference		1.536	OK
Measured TDS = C	Calculated TDS		
Measured		681.673	
Calculated		793.548	
Ratio		0.859	Not within range 1.0 to 1.2
Measured $EC = Ca$	Iculated EC		
Measured		1230.000	
Calculated		1155.589	
Ratio		1.064	OK
Secondary Tests			
Measured EC and	Ion Sums:		
Anions		0.970448	Within preferred range (0.9-
1.1)		a 1212 0122012	
Cations		1.000720	Within preferred range (0.9-
1.1)		2 22 2	
Calculated TDS to		0.645	OK
Measured TDS to I		0.554	OK
Organic Mass Bala			
$DOC \ge Sum of Org$		1 (00 /7	
Dissolved Organi	ic Carbon	1.600 mg/L	01/
Sum of Organics		0.000 mg/L	OK

COMDI 11D

1

22

			CSMR	RI-12
Water Type	Ca-Cl			
Dissolved Solids	771.84 mg/kg		770 mg/L	Measured
	0.99761 g/cm		770 mg/L	Calculated
Density Conductivity	1530 μmho/ci			Measured
Hardness (as CaCO		un		Wiedsuled
	57		555 40 mm /I	Colorited 1
Total Carbonate	556.81 mg/kg 526.17	5	555.48 mg/L	Calculated
Non-Carbonate	30.642		524.91 30.569	
Non-Cardonate	30.042		30.309	
Primary Tests				
Anion-Cation Bala	nce			
Anions		13.1		
Cations		13.7		28
% Difference		2.190		OK
Measured $TDS = C$	alculated TDS			
Measured		771.84		
Calculated		926.2	10	
Ratio		0.833		Not within range 1.0 to 1.2
Measured $EC = Ca$	Iculated EC	1620 (200	
Measured		1530.0		
Calculated Ratio		1235.5	5/1	Not within you go 0.0 to 1.1
Secondary Tests		1.238		Not within range 0.9 to 1.1
Measured EC and 1	Ion Sume			
Anions	ion Sums.	0.855	1/13	Not within preferred range
(0.9-1.1)		0.055	145	Not within preferred range
Cations		0.8934	141	Not within preferred range
(0.9-1.1)		0.075-	111	Not within preferred range
Calculated TDS to	EC ratio	0.605		OK
Measured TDS to E		0.504		Not within preferred range
(0.55-0.7)		010 0 1		rice within preferred range
Organic Mass Bala	nce			
$\overline{DOC} \ge Sum of Org$				
Dissolved Organi		2.300	mg/L	
Sum of Organics		0.000		OK

			CSMF	I-13	
Water Type	Ca-Cl				
Dissolved Solids	721.75 mg/kg		720 mg/L	Measu	
Density	0.99758 g/cm ³			Calcu	
Conductivity	1220 µmho/cn	n		Measu	ired
Hardness (as CaCO	3)				
Total	515.02 mg/kg		513.78 mg/L	Calcu	lated
Carbonate	460.41		459.3		
Non-Carbonate	54.611		54.479		
Primary Tests					
Anion-Cation Balan	nce				
Anions		12.2			
Cations		12.9			
% Difference		2.787		OK	
Measured $TDS = C$	alculated TDS				
Measured		721.75			
Calculated		841.23	9		
Ratio		0.858		Not w	ithin range 1.0 to 1.2
Measured EC = Ca	lculated EC				
Measured		1220.0			
Calculated		1171.5	16		
Ratio		1.041		OK	
Secondary Tests					
Measured EC and]	lon Sums:				
Anions		1.0026	32	Withi	n preferred range (0.9-
1.1)					
Cations		1.0601	16	Withi	n preferred range (0.9-
1.1)					
Calculated TDS to	EC ratio	0.690		OK	
Measured TDS to F	EC ratio	0.592		OK	
Organic Mass Bala	nce				
$DOC \ge Sum of Org$	anics				
Dissolved Organi	c Carbon	2.300 1	mg/L		
Sum of Organics		0.000 1	mg/L	OK	
204756242			1411-1111		

		CSM	RI-14
Water Type	Na-HCO ₃		
Dissolved Solids	250.7 mg/kg	250 mg/L	Measured
Density	0.99722 g/cm ²		Calculated
Conductivity	4110 µmho/cr		Measured
Hardness (as CaCO		ĉ.	
Total	118.79 mg/kg	118.46 mg/L	Calculated
Carbonate	118.79	118.46	
Non-Carbonate	0.0	0.0	
Primary Tests			
Anion-Cation Bala	nce		
Anions		4.07	
Cations		4.17	
% Difference		1.303	OK
Measured TDS = C	alculated TDS		
Measured		250.697	
Calculated		316.781	
Ratio		0.791	Not within range 1.0 to
Measured EC = Ca	Iculated EC		
Measured		4110.000	
Calculated		382.118	
Ratio		10.756	Not within range 0.9 to
Secondary Tests			
Measured EC and	Ion Sums:		
Anions		0.098935	Not within preferred rat
(0.9-1.1)			
Cations		0.101547	Not within preferred ra
(0.9-1.1)		(147) - on fain in	
Calculated TDS to	EC ratio	0.077	Not within preferred rat
(0.55-0.7)		0.0.01	
Measured TDS to I	C ratio	0.061	Not within preferred ra
(0.55-0.7)			
Organic Mass Bala			
$DOC \ge Sum of Org$	anics	1 000 //	
	c (arbon	1.800 mg/L	
Dissolved Organics	le curbon	0.000 mg/L	ОК

			CSMR	AI-1B
	0.01			
Water Type	Ca-Cl		710 //	Measured
Dissolved Solids	711.73 mg/kg		710 mg/L	
Density	0.99757 g/cm			Calculated
Conductivity	1380 µmho/cr	n		Measured
Hardness (as CaCO	March and an			
Total	544.19 mg/kg		542.86 mg/L	Calculated
Carbonate	378.2		377.28	
Non-Carbonate	165.99		165.58	
Primary Tests				
Anion-Cation Bala	nce			
Anions		12.7		
Cations		13.2		
% Difference		1.880		OK
Measured $TDS = C$	Calculated TDS		5 10	
Measured		711.73		
Calculated		850.56	59	
Ratio		0.837		Not within range 1.0 to 1.2
Measured $EC = Ca$	lculated EC	1200	200	
Measured		1380.0		
Calculated		1235.9	<i>¥</i> /9	
Ratio		1.117		Not within range 0.9 to 1.1
Secondary Tests	I C			
Measured EC and	ion Sums:	0.919	276	Within matamad range (0.0
Anions		0.9195	7/0	Within preferred range (0.9-
1.1) Cations		0.9552	225	Within preferred range (0.9-
1.1)		0.9332	233	within preferred range (0.9-
Calculated TDS to	FC ratio	0.616		OK
Measured TDS to I		0.516		Not within preferred range
(0.55-0.7)		0.510		Not within preferred funge
Organic Mass Bala	nce			
$DOC \ge Sum of Org$				
Dissolved Organi		1.700	mg/L	
Sum of Organics		0.000		ОК
Sam er SrBannes			0	pressent)

			CSN	MRI	-2
Water Type	Ca-HCO ₃				
Dissolved Solids	401.07 mg/kg		400 mg/L		Measured
Density	0.99733 g/cm				Calculated
Conductivity	643 µmho/cm	l.			Measured
Hardness (as CaCO	3)				
Total	339.8 mg/kg		338.9 mg/L		Calculated
Carbonate	339.8		338.9		
Non-Carbonate	0.0		0.0		
Primary Tests					
Anion-Cation Balar	100				
Anions		6.59			
Cations		7.79			
% Difference		8.317			Not within $\pm 2\%$
Measured TDS = C	alculated TDS	0.0 1 1			
Measured		401.06	59		
Calculated		528.10)8		
Ratio		0.759			Not within range 1.0 to 1.2
Measured EC = Ca	lculated EC				ç
Measured		643.00	00		
Calculated		649.75	56		
Ratio		0.990			OK
Secondary Tests					
Measured EC and I	on Sums:				
Anions		1.0252	225		Within preferred range (0.9-
1.1)					
Cations		1.2112	237		Not within preferred range
(0.9-1.1)					
Calculated TDS to]	EC ratio	0.821			Not within preferred range
(0.55-0.7)					
Measured TDS to E		0.624			OK
Organic Mass Bala					
$DOC \ge Sum of Org$	anics				
DOC unavailable					

		CSM	RI-4
Water Type	Ca-Cl		
Dissolved Solids	791.88 mg/kg	790 mg/L	Measured
Density	0.99763 g/cm		Calculated
Conductivity	1600 μmho/cr		Measured
Hardness (as CaCC			Wiedstriet
Total	589.82 mg/kg	588.42 mg/L	Calculated
Carbonate	476.83	475.7	
Non-Carbonate	112.99	112.72	
Primary Tests			
Anion-Cation Bala	nce		
Anions		13.2	
Cations		14.4	
% Difference		4.384	OK
Measured TDS = C	Calculated TDS		
Measured		791.878	
Calculated		926.838	
Ratio		0.854	Not within range 1.0 to 1.2
Measured EC = Ca	Iculated EC		
Measured		1600.000	
Calculated		1281.321	
Ratio		1.249	Not within range 0.9 to 1.1
Secondary Tests			
Measured EC and	Ion Sums:		
Anions		0.825258	Not within preferred range
(0.9-1.1)		0.00005	
Cations		0.900935	Within preferred range (0.9-
1.1)	FC /	0.670	01/
Calculated TDS to		0.579	OK
Measured TDS to I	EC ratio	0.495	Not within preferred range
(0.55-0.7)			
Organic Mass Bala			
$DOC \ge Sum of Org$ Dissolved Organi		2.100 mg/I	
Sum of Organics	ic Carbon	2.100 mg/L 0.000 mg/L	ОК
Sum of Organics		0.000 mg/L	UK

			CSM	RI-5
	G G1			
Water Type	Ca-Cl			1-01-0-
Dissolved Solids	701.71 mg/kg		700 mg/L	Measured
Density	0.99756 g/cm			Calculated
Conductivity	1510 µmho/cr	m		Measured
Hardness (as CaCO	3)			
Total	519.42 mg/kg	5	518.16 mg/L	Calculated
Carbonate	361.76		360.88	
Non-Carbonate	157.66		157.28	
Primary Tests				
Anion-Cation Bala	nce			
Anions		12		
Cations		12.9		
% Difference		3.394		OK
Measured TDS = C				
Measured		701.7	12	
Calculated		805.465		
Ratio		0.871		Not within range 1.0 to 1.2
Measured EC = Calculated EC				5
Measured		1510.0	000	
Calculated		1182.516		
Ratio		1.277		Not within range 0.9 to 1.1
Secondary Tests				
Measured EC and	lon Sums:			
Anions		0.795	895	Not within preferred range
(0.9-1.1)				
Cations		0.851817		Not within preferred range
(0.9-1.1)				
Calculated TDS to (0.55-0.7)	EC ratio	0.533		Not within preferred range
Measured TDS to H	EC ratio	0.465		Not within preferred range
(0.55-0.7)				
Organic Mass Bala				
$DOC \ge Sum of Org$		1 100	1	
Dissolved Organi	c Carbon	1.400		ov
Sum of Organics		0.000	mg/L	OK

	CSMRI-6C				
Water Trees	Ca-Cl				
Water Type Dissolved Solids	631.57 mg/kg		630 mg/L		Measured
Dissolved Solids	0.99751 g/cm		030 mg/L		Calculated
Conductivity	1160 µmho/ci				Measured
Hardness (as CaCO		п			Ivieasured
Total	Sector and the sector of the s		410.2 mm/I		Calculated
Carbonate	411.33 mg/kg 296		410.3 mg/L 295.26		Calculated
Non-Carbonate	115.33		115.04		
-					
Primary Tests					
Anion-Cation Bala	nce	11.0			
Anions		11.2 11.3			
Cations % Difference		0.497			ОК
Measured TDS = C	0.497			UK	
Measured	alculated 1D5	631.5	74		
Calculated			718.090		
Ratio			0.880		Not within range 1.0 to 1.2
Measured $EC = Ca$	Iculated EC	0.000			ittor within range 1.0 to 1.2
Measured		1160.0	000		
Calculated		1086.			
Ratio			1.068		OK
Secondary Tests					
Measured EC and	lon Sums:				
Anions		0.9619	936		Within preferred range (0.9-
1.1)					
Cations		0.971:	543		Within preferred range (0.9-
1.1)					
Calculated TDS to EC ratio		0.619			OK
Measured TDS to F	EC ratio 0.544			Not within preferred range	
(0.55-0.7)					
Organic Mass Bala					
$DOC \ge Sum of Org$		1 (00	1		
Dissolved Organi	c Carbon	1.600			OV
Sum of Organics		0.000	mg/L		OK

		CS	MRI-'	7C
Water Type	Ca-HCO ₃			
Dissolved Solids	641.59 mg/kg	640 mg/L		Measured
Density	0.99752 g/cm			Calculated
Conductivity	1000 µmho/cr			Measured
Hardness (as CaCO				
Total	440.22 mg/kg	439.13 mg/I	Ĺ	Calculated
Carbonate	440.22	439.13		
Non-Carbonate	0.0	0.0		
<u>Primary Tests</u> Anion-Cation Bala	200			
Anions	lice	10.3		
Cations		11.6		
% Difference		6.124		Not within \pm 5%
Measured TDS = C	alculated TDS	0.124		Not within ± 570
Measured	alculated 100	641.594		
Calculated		790.665		
Ratio		0.811		Not within range 1.0 to 1.2
Measured EC = Ca	lculated EC			
Measured		1000.000		
Calculated		975.182		
Ratio		1.025		OK
Secondary Tests				
Measured EC and I	lon Sums:			
Anions		1.029092		Within preferred range (0.9-
1.1)				
Cations		1.163350		Not within preferred range
(0.9-1.1)				
Calculated TDS to	EC ratio	0.791		Not within preferred range
(0.55-0.7)	and the	The Manager		
Measured TDS to F		0.642		OK
Organic Mass Bala				
$DOC \ge Sum of Org$		1 500 7		
Dissolved Organi	c Carbon	1.500 mg/L		01/
Sum of Organics		0.000 mg/L		OK

			CSMR	[-8B		
Water Type	Ca-Cl					
Dissolved Solids	882.03 mg/kg		880 mg/L	Measured		
Density	0.9977 g/cm ³			Calculated		
Conductivity	1470 µmho/cr	n		Measured		
Hardness (as CaCO	·3)					
Total	639.84 mg/kg		638.36 mg/L	Calculated		
Carbonate	559.01		557.72			
Non-Carbonate	80.833		80.646			
<u>Primary Tests</u> Anion-Cation Bala	0.60					
Anions	lice	14.6				
Cations		15.8				
% Difference		3.985		ОК		
Measured TDS = Calculated TDS						
Measured		882.0	32			
Calculated		1028.	670			
Ratio	0.857			Not within range 1.0 to 1.2		
Measured EC = Ca	lculated EC					
Measured		1470.	000			
Calculated		1386.4	481			
Ratio		1.060		OK		
Secondary Tests						
Measured EC and	Ion Sums:					
Anions		0.992	574	Within preferred range (0.9-		
1.1)		CATE - SPATIAL COLOR	24650 (940			
Cations	1.074956		956	Within preferred range (0.9-		
1.1)						
Calculated TDS to	0.700		OK			
Measured TDS to EC ratio		0.600		OK		
Organic Mass Bala						
$DOC \ge Sum of Org$		2 000	/ T			
Dissolved Organic Carbon 2.800 mg/L Sum of Organics 0.000 mg/L			OK			
Sum of Organics		0.000	IIIg/L	OK		

			CSMI	RI-9	
Water Type	Ca-Cl				
Dissolved Solids	741.79 mg/kg	ſ	740 mg/L	Measured	
Density	0.99759 g/cm			Calculated	
Conductivity	1620 µmho/ci			Measured	
Hardness (as CaCO					
Total	581.33 mg/kg	r.	579.93 mg/L	Calculated	
Carbonate	476.85	, ,	475.7	Cultured	
Non-Carbonate	104.48		104.23		
Primary Tests					
Anion-Cation Bala	nce				
Anions		12.7			
Cations		13.6			
% Difference		3.422		OK	
Measured TDS = C	alculated TDS				
Measured		741.7	87		
Calculated		873.4	04		
Ratio		0.849		Not within range 1.0 to 1.2	
Measured EC = Ca	Iculated EC			tan aran in analasi tan "na kandaran 💳 ku sekarat setata kanalas	
Measured		1620.	000		
Calculated		1222.	562		
Ratio		1.325		Not within range 0.9 to 1.1	
Secondary Tests					
Measured EC and	lon Sums:				
Anions		0.784	273	Not within preferred range	
(0.9-1.1)					
Cations		0.839	845	Not within preferred range	
(0.9-1.1)					
Calculated TDS to EC ratio		0.539		Not within preferred range	
(0.55-0.7)					
Measured TDS to EC ratio		0.458		Not within preferred range	
(0.55-0.7)					
Organic Mass Bala					
$DOC \ge Sum of Org$					
Dissolved Organi	c Carbon	1.600			
Sum of Organics		0.000	mg/L	OK	

			SW	-1
Water Type	Ca-SO ₄			
Dissolved Solids	240.67 mg/kg	,	240 mg/L	Measured
Density	0.99721 g/cm		240 mg/L	Calculated
Conductivity	465 μmho/cm			Measured
Hardness (as CaCO				Wiedsdred
Total	5.		121.54 mc/I	Calculated
Carbonate	131.9 mg/kg 82.247		131.54 mg/L 82.017	Calculated
Non-Carbonate	49.657		49.519	
Non-Carbonate	49.037		49.519	
Primary Tests				
Anion-Cation Bala	nce			
Anions		3.56		
Cations		3.44		
% Difference		1.690		OK
Measured TDS = Calculated TDS			230	
Measured		240.67 241.27		
	Calculated		73	
Ratio		0.998		Not within range 1.0 to 1.2
Measured EC = Calculated EC				
Measured		465.00		
Calculated		381.747		
Ratio		1.218		Not within range 0.9 to 1.1
Secondary Tests				
Measured EC and	lon Sums:	0.764	co.o.	
Anions		0.7646	500	Not within preferred range
(0.9-1.1)		0 720	100	
Cations		0.7391	188	Not within preferred range
(0.9-1.1)		0 510		
Calculated TDS to EC ratio (0.55-0.7)		0.519		Not within preferred range
Measured TDS to EC ratio		0.518		Not within preferred range
(0.55-0.7)	10.1			
Organic Mass Bala				
$DOC \ge Sum of Org$	1 100	ma/I		
Dissolved Organi	1.100		ОК	
Sum of Organics		0.000	IIIg/L	UK

		SW-2					
	Water Type	Ca-SO ₄					
	Dissolved Solids	240.67 mg/kg	240 mg/L	Measured			
	Density	0.99721 g/cm ³		Calculated			
	Conductivity	483 µmho/cm		Measured			
	Hardness (as CaCC						
	Total	138.56 mg/kg	138.18 mg/L	Calculated			
	Carbonate	80.602	80.377				
	Non-Carbonate	57.962	57.8				
	Primary Tests						
	Anion-Cation Bala	nce					
	Anions		3.35				
	Cations		3.57				
	% Difference		3.187	Not within $\pm 2\%$			
	Measured TDS = C	alculated TDS					
	Measured		240.671				
	Calculated		250.699				
	Ratio		0.960	Not within range 1.0 to 1.2			
	Measured EC = Ca	lculated EC					
	Measured		483.000				
	Calculated		385.263				
	Ratio		1.254	Not within range 0.9 to 1.1			
	Secondary Tests						
	Measured EC and	Ion Sums:	16 - 6000 60 60 8000	57 58 20-20102-1 90 92			
	Anions		0.693569	Not within preferred range			
	(0.9-1.1)			terme devolution tour 10			
	Cations		0.739228	Not within preferred range			
		0.9-1.1)					
		Calculated TDS to EC ratio		Not within preferred range			
	(0.55-0.7)						
	Measured TDS to I	EC ratio	0.498	Not within preferred range			
	(0.55-0.7)						
	Organic Mass Bala						
	$DOC \ge Sum of Org$						
	DOC unavailable						

1				
		SW-3		
Water Type	Ca-SO ₄			
Dissolved Solids	ACCOUNT OF A DESCRIPTION OF A DESCRIPTIO	230 mg/L	Measured	
Density	0.9972 g/cm^3	200 mg 2	Calculated	
Conductivity	471 μmho/cm		Measured	
Hardness (as Ca			medulou	
Total	138.15 mg/kg	137.77 mg/L	Calculated	
Carbonate	78.957	78.737		
Non-Carbona	te 59.194	59.029		
Primary Tests				
Anion-Cation B	alance			
Anions		3.44		
Cations		3.56		
% Difference		1.683	OK	
Measured TDS	= Calculated TDS			
Measured		230.645		
Calculated		247.592		
Ratio		0.932	Not within range 1.0 to 1.2	
Measured EC =	Calculated EC		-	
Measured		471.000		
Calculated		386.833		
Ratio		1.218	Not within range 0.9 to 1.1	
Secondary Tests	5			
Measured EC a	nd Ion Sums:			
Anions		0.731216	Not within preferred range	
(0.9-1.1)				
Cations		0.756249	Not within preferred range	
(0.9-1.1)				
Calculated TDS	to EC ratio	0.526	Not within preferred range	
(0.55-0.7)				
Measured TDS	to EC ratio	0.490	Not within preferred range	
(0.55-0.7)				
Organic Mass B				
$DOC \ge Sum of O$	Organics			
Dissolved Org		1.100 mg/L		
Sum of Organ	ics	0.000 mg/L	OK	

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